

## SUPPLEMENTARY INFORMATION

### **Rhenium-Containing Compound(PyHReO<sub>4</sub>): Synthesis, Characterization and Catalytic Application in Olefin Epoxidation and Baeyer-Villiger Oxidation**

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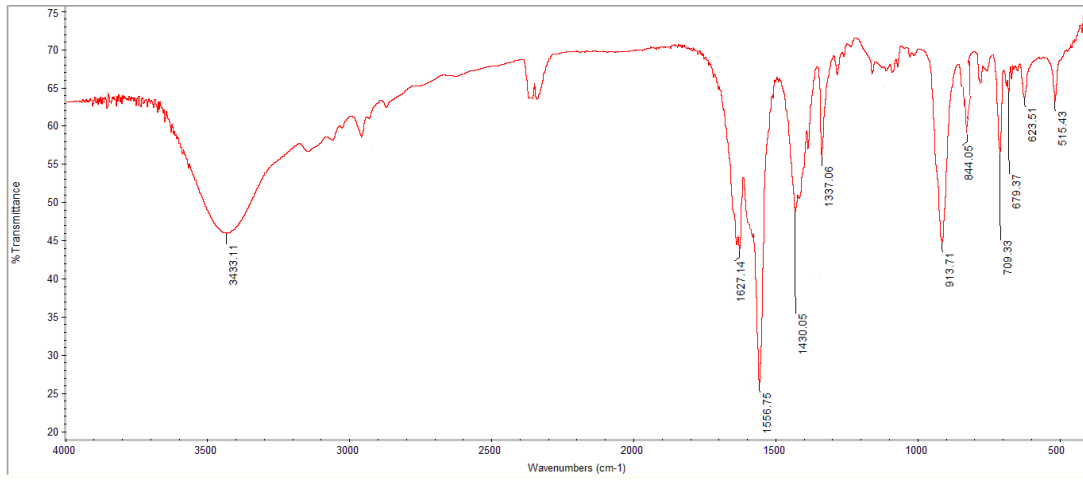


Fig. 1 FT-IR spectrum of PyHReO<sub>4</sub>

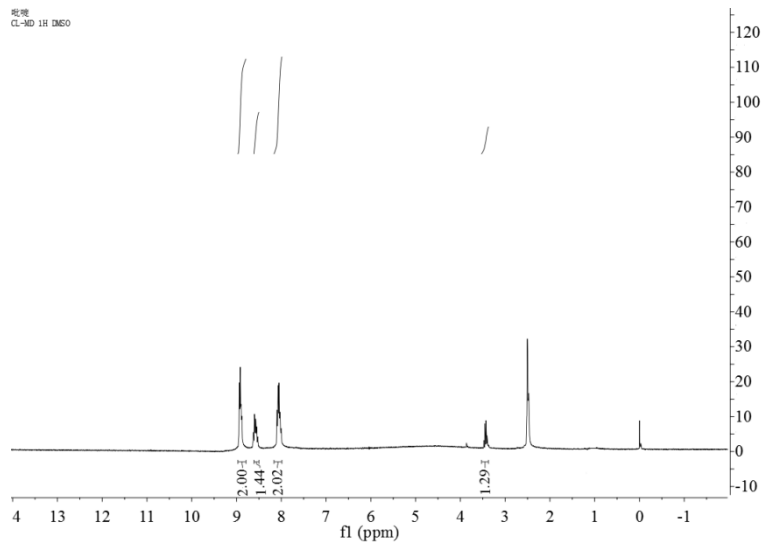


Fig. 2 <sup>1</sup>H NMR(300 MHz,DMSO) spectrum of PyHReO<sub>4</sub>

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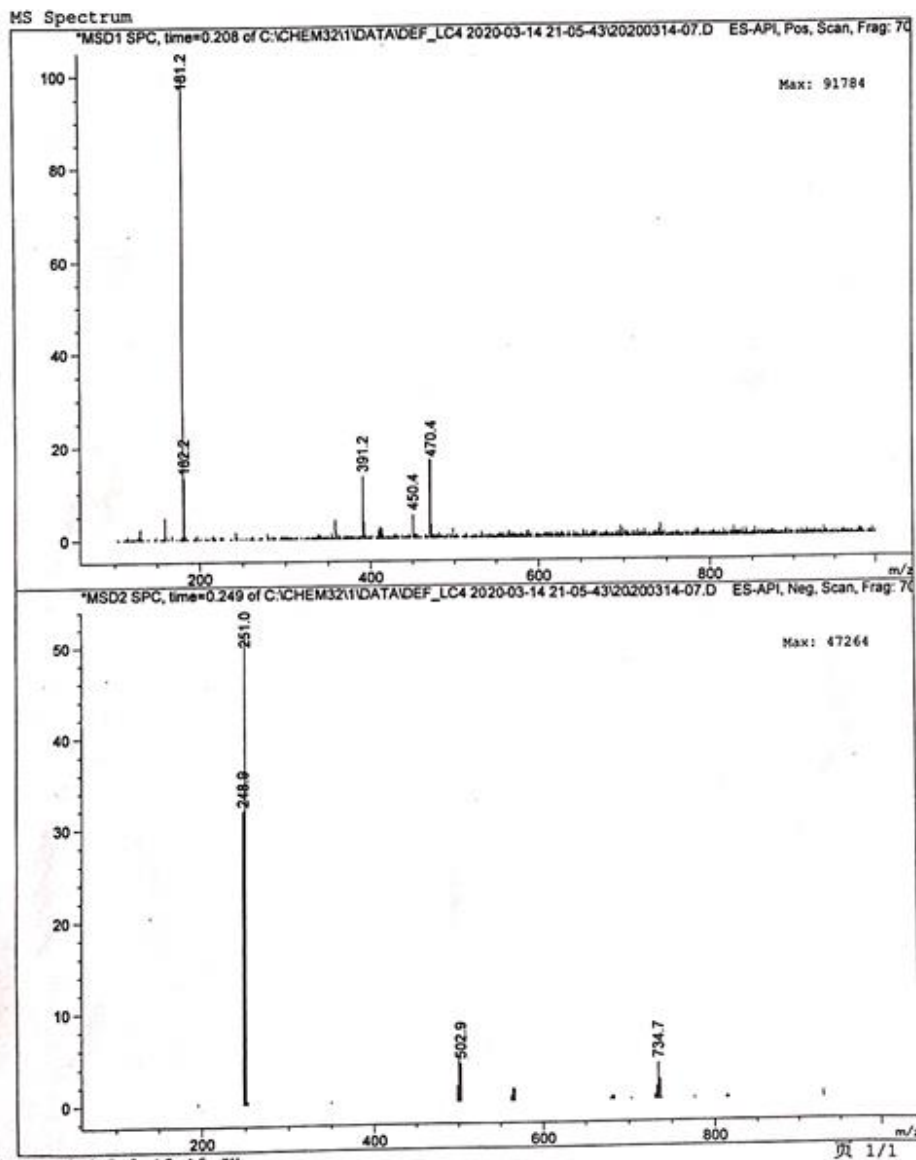


Fig. 3 ESI-MS(Acetonitrile) spectra of PyHReO<sub>4</sub>

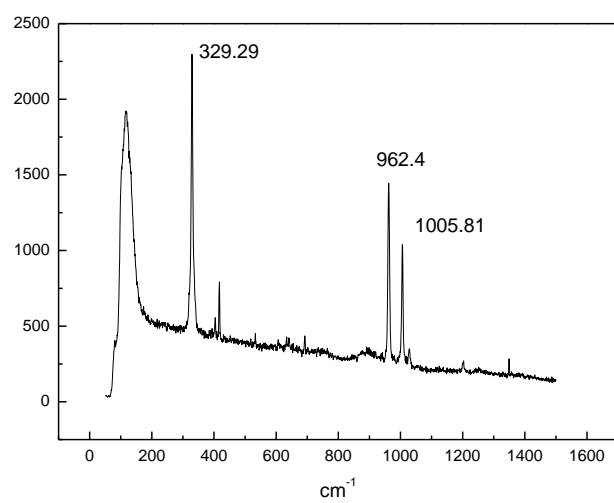


Fig. 4 The Raman spectra of PyHReO<sub>4</sub>

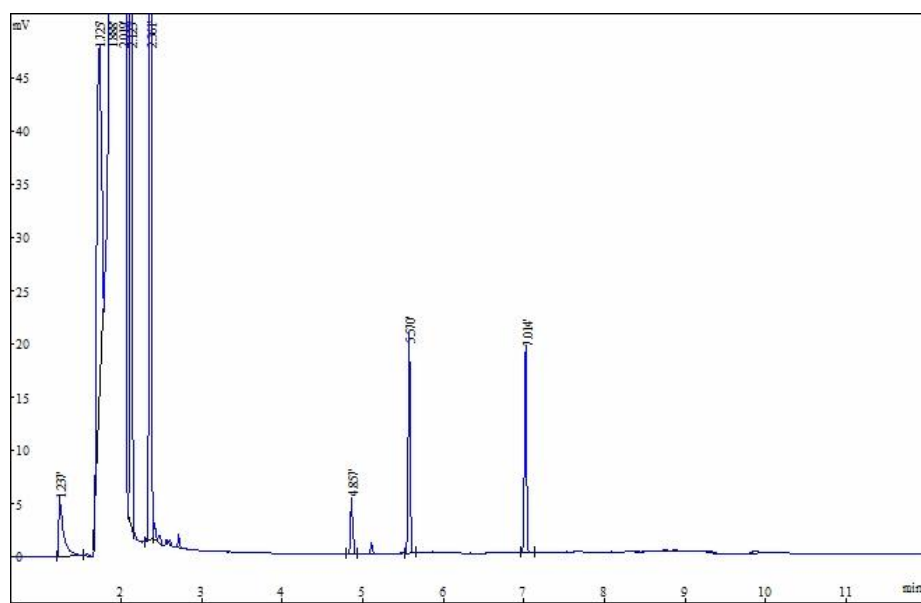


Fig. 5 GC data of optimal experiment on cyclooctene epoxidation

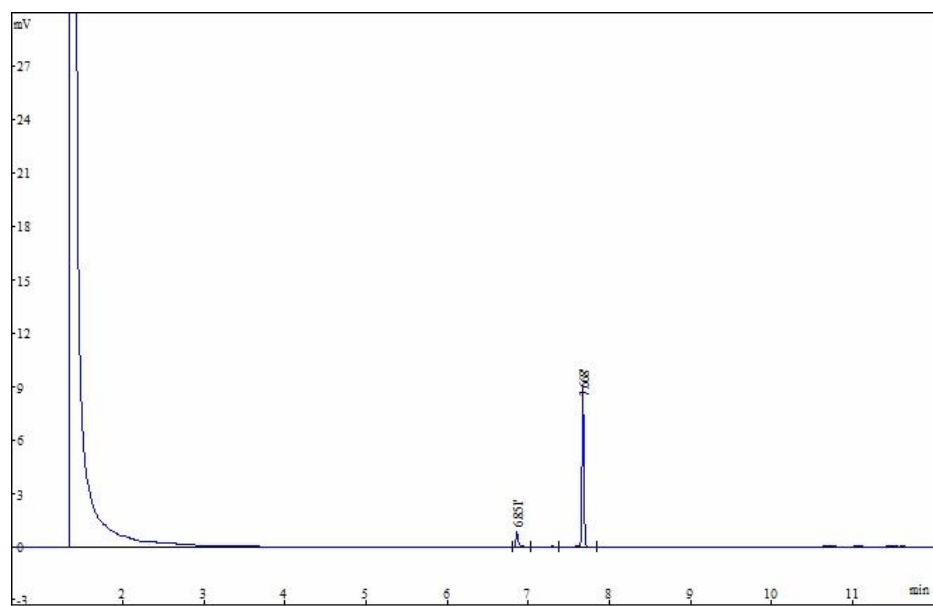


Fig. 6 GC data of optimal experiment of 2-adamantanone on Baeyer-Villiger oxidation



Fig. 7 Crystal photo of  $\text{PyHReO}_4$

**Abstract****Computing details**

Program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997).

**References**

NOT FOUND

**(shelxl)***Crystal data*

$C_5H_6NO_4Re$	$V = 1491.7 (3) \text{ \AA}^3$
$M_r = 330.31$	$Z = 8$
?, ?	$F(000) = 1200$
$a = 7.230 (1) \text{ \AA}$	$D_x = 2.941 \text{ Mg m}^{-3}$
$b = 12.1070 (15) \text{ \AA}$	Mo $K\alpha$ radiation, $\lambda = 0.71075 \text{ \AA}$
$c = 17.0420 (18) \text{ \AA}$	$\mu = 16.26 \text{ mm}^{-1}$
$\alpha = 90^\circ$	$T = 113 \text{ K}$
$\beta = 90^\circ$	$0.24 \times 0.22 \times 0.18 \text{ mm}$
$\gamma = 90^\circ$	

*Data collection*

Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.125$
graphite	$\theta_{\text{max}} = 30.1^\circ$ , $\theta_{\text{min}} = 2.4^\circ$
18604 measured reflections	$h = -10 \rightarrow 10$
2178 independent reflections	$k = -17 \rightarrow 16$
2160 reflections with $I > 2\sigma(I)$	$l = -23 \rightarrow 23$

*Refinement*

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.046$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.121$	$w = 1/[\sigma^2(F_o^2) + (0.0637P)^2 + 19.3911P]$
$S = 1.13$	where $P = (F_o^2 + 2F_c^2)/3$
2178 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
101 parameters	$\Delta\rho_{\text{max}} = 3.81 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -6.16 \text{ e \AA}^{-3}$

0 restraints

Extinction correction: *SHELXL*,  
 $F_c^* = kFc[1 + 0.001x\lambda^3/\sin(2\theta)]^{-1/4}$ 

Primary atom site location: structure-invariant direct methods

Extinction coefficient: 0.00086 (15)

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\text{sigma}(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Re1	0.11765 (3)	0.26308 (2)	0.363573 (14)	0.00525 (15)
O1	0.1589 (8)	0.4036 (5)	0.3585 (3)	0.0162 (10)
O2	0.0227 (8)	0.2185 (4)	0.2752 (3)	0.0162 (10)
O3	0.3252 (8)	0.1937 (5)	0.3810 (3)	0.0216 (11)
O4	-0.0328 (8)	0.2374 (4)	0.4392 (3)	0.0152 (10)
N1	0.6583 (8)	0.0831 (5)	0.3518 (3)	0.0099 (10)
H1	0.5895	0.1357	0.3313	0.012*
C1	0.7629 (8)	0.0209 (5)	0.3037 (4)	0.0096 (11)
H1A	0.7614	0.0339	0.2488	0.012*
C2	0.8732 (9)	-0.0626 (7)	0.3346 (4)	0.0133 (14)
H2	0.9475	-0.1070	0.3011	0.016*
C3	0.8729 (8)	-0.0801 (7)	0.4157 (4)	0.0128 (13)
H3	0.9482	-0.1362	0.4379	0.015*
C4	0.7614 (8)	-0.0146 (5)	0.4641 (4)	0.0108 (12)
H4	0.7587	-0.0264	0.5192	0.013*
C5	0.6549 (9)	0.0678 (6)	0.4300 (4)	0.0114 (11)
H5	0.5794	0.1136	0.4620	0.014*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Re1	0.00068 (18)	0.0073 (2)	0.0078 (2)	0.00115 (7)	0.00023 (6)	-0.00003 (7)
O1	0.017 (2)	0.009 (2)	0.022 (2)	-0.004 (2)	0.0004 (18)	0.0004 (17)
O2	0.019 (2)	0.019 (2)	0.011 (2)	-0.004 (2)	-0.0001 (18)	-0.0027 (18)
O3	0.015 (3)	0.029 (3)	0.021 (2)	0.014 (2)	0.000 (2)	0.003 (2)
O4	0.018 (3)	0.015 (2)	0.013 (2)	-0.0031 (18)	0.003 (2)	0.0023 (16)
N1	0.003 (2)	0.012 (3)	0.014 (2)	-0.002 (2)	0.0013 (19)	0.002 (2)
C1	0.006 (2)	0.011 (3)	0.012 (3)	0.000 (2)	0.0007 (18)	-0.002 (2)
C2	0.008 (3)	0.013 (3)	0.019 (3)	0.001 (2)	0.002 (2)	-0.006 (3)
C3	0.010 (3)	0.012 (3)	0.017 (3)	0.000 (2)	0.000 (2)	0.002 (3)

## structure report

C4	0.005 (2)	0.016 (3)	0.012 (3)	0.002 (2)	0.0007 (18)	-0.001 (2)
C5	0.003 (2)	0.017 (3)	0.015 (3)	-0.003 (2)	0.000 (2)	-0.001 (2)

### Geometric parameters (Å, °)

Re1—O4	1.715 (5)	C1—H1A	0.9500
Re1—O1	1.730 (6)	C2—C3	1.398 (10)
Re1—O2	1.740 (5)	C2—H2	0.9500
Re1—O3	1.745 (5)	C3—C4	1.399 (9)
N1—C1	1.345 (8)	C3—H3	0.9500
N1—C5	1.347 (9)	C4—C5	1.387 (9)
N1—H1	0.8800	C4—H4	0.9500
C1—C2	1.391 (9)	C5—H5	0.9500
O4—Re1—O1	109.0 (2)	C1—C2—C3	118.9 (6)
O4—Re1—O2	110.1 (3)	C1—C2—H2	120.6
O1—Re1—O2	109.2 (2)	C3—C2—H2	120.6
O4—Re1—O3	109.3 (3)	C2—C3—C4	119.9 (7)
O1—Re1—O3	109.5 (3)	C2—C3—H3	120.1
O2—Re1—O3	109.7 (3)	C4—C3—H3	120.1
C1—N1—C5	122.4 (6)	C5—C4—C3	118.7 (7)
C1—N1—H1	118.8	C5—C4—H4	120.6
C5—N1—H1	118.8	C3—C4—H4	120.6
N1—C1—C2	120.0 (6)	N1—C5—C4	120.2 (6)
N1—C1—H1A	120.0	N1—C5—H5	119.9
C2—C1—H1A	120.0	C4—C5—H5	119.9
C5—N1—C1—C2	-0.2 (10)	C2—C3—C4—C5	-0.9 (10)
N1—C1—C2—C3	0.0 (10)	C1—N1—C5—C4	-0.2 (10)
C1—C2—C3—C4	0.6 (10)	C3—C4—C5—N1	0.7 (9)

### Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1 $\cdots$ O3	0.88	2.20	2.800 (8)	125
C5—H5 $\cdots$ O4 <sup>i</sup>	0.95	2.60	3.517 (9)	163
N1—H1 $\cdots$ O2 <sup>ii</sup>	0.88	2.13	2.887 (8)	144
C5—H5 $\cdots$ O4 <sup>i</sup>	0.95	2.60	3.517 (9)	163

Symmetry codes: (i)  $x+1/2, -y+1/2, -z+1$ ; (ii)  $x+1/2, y, -z+1/2$ .