

## SUPPLEMENTARY INFORMATION

### **Investigation of photophysical, structural aspects and nonlinear optical properties of Foron blue SR analogs using Density Functional Theory (DFT)**

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**Table S1.** Selected bond lengths, bond angles and torsional angles of **2a** and **2b**, using B3LYP 6-31G(d) in DMF (bond lengths are in Å, dihedral angles are in degree)

**Table S2.** Selected bond lengths, bond angles and torsional angles of **3a** and **3b**, using B3LYP 6-31G(d) in DMF (bond lengths are in Å, dihedral angles are in degree)

**Table S3.** Selected bond lengths, bond angles and torsional angles of **1a** and **1b**, using B3LYP 6-31G(d) in EA (bond lengths are in Å, dihedral angles are in degree)

**Table S4.** Selected bond lengths, bond angles and torsional angles of **2a** and **2b**, using B3LYP 6-31G(d) EA (bond lengths are in Å, dihedral angles are in degree)

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**Table S1.** Selected bond lengths, bond angles and torsional angles of **2a** and **2b**, using B3LYP 6-31G(d) in DMF (bond lengths are in Å, dihedral angles are in degree)

Atom number	<b>2a</b>			<b>2b</b>		
	Gas	DMF		Gas	DMF	
	GS	GS	ES	GS	GS	ES
Bond length						
N <sub>27</sub> -C <sub>19</sub>	1.37	1.36	1.36	1.38	1.37	1.36
C <sub>22</sub> -C <sub>17</sub>	1.46	1.45	1.50	1.46	1.45	1.50
C <sub>17</sub> -C <sub>7</sub>	1.37	1.38	1.41	1.37	1.38	1.41
bond angle						
C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub>	128.79	129.47	122.14	125.77	126.37	121.91
C <sub>17</sub> -C <sub>7</sub> -C <sub>8</sub>	124.30	124.57	127.43	125.61	125.85	127.51
C <sub>17</sub> -C <sub>7</sub> -S <sub>14</sub>	125.90	125.67	121.91	124.26	123.99	121.81
Dihedral angle						
C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub> -S <sub>14</sub>	9.67	11.57	0.10	12.13	15.94	4.91
C <sub>23</sub> -C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub>	25.99	24.30	90.68	43.00	39.50	90.92

**Table S2.** Selected bond lengths, bond angles and torsional angles of **3a** and **3b**, using B3LYP 6-31G(d) in DMF (bond lengths are in Å, dihedral angles are in degree)

Atom number	<b>3a</b>			<b>3b</b>		
	Gas	DMF		Gas	DMF	
	GS	GS	ES	GS	GS	ES
Bond length						
N <sub>26</sub> -C <sub>18</sub>	1.36895	1.35661	1.35504	1.37151	1.35781	1.35423
C <sub>21</sub> -C <sub>16</sub>	1.4341	1.4227	1.48405	1.42759	1.41296	1.4815
C <sub>16</sub> -C <sub>7</sub>	1.37231	1.38166	1.36244	1.37511	1.38765	1.36236
Bond angle						
C <sub>21</sub> -C <sub>16</sub> -C <sub>7</sub>	132.3531	132.4504	126.1881	131.6421	132.0738	125.769
C <sub>16</sub> -C <sub>7</sub> -C <sub>8</sub>	126.9239	126.4101	132.4256	126.55	126.0589	132.4425
C <sub>16</sub> -C <sub>7</sub> -S <sub>13</sub>	123.0723	123.3975	118.3908	123.3237	123.5224	118.3267
Dihedral angle						
C <sub>21</sub> -C <sub>7</sub> -C <sub>16</sub> -S <sub>13</sub>	12.38786	15.7182	1.4734	14.16729	17.67082	2.9818
C <sub>22</sub> -C <sub>21</sub> -C <sub>16</sub> -C <sub>7</sub>	17.67245	16.68476	91.50966	20.00739	17.56089	91.33579

**Table S3.** Selected bond lengths, bond angles and torsional angles of **1a** and **1b**, using B3LYP 6-31G(d) in EA (bond lengths are in Å, dihedral angles are in degree)

Atom number	<b>1a</b>		<b>1b</b>	
	Gas	EA	Gas	EA

	GS	GS	ES	GS	GS	ES
<b>Bond length</b>						
N <sub>28</sub> -C <sub>18</sub>	1.36112	1.36733	1.37404	1.36389	1.3712	1.37371
C <sub>21</sub> -C <sub>17</sub>	1.43763	1.43055	1.47923	1.433	1.42372	1.47611
C <sub>17</sub> -C <sub>7</sub>	1.37239	1.36352	1.35709	1.37779	1.36822	1.35849
<b>Bond angle</b>						
C <sub>21</sub> -C <sub>17</sub> -C <sub>7</sub>	133.6811	134.1668	125.7073	133.0528	133.7498	126.4356
C <sub>17</sub> -C <sub>7</sub> -C <sub>8</sub>	120.8806	121.12	126.5021	120.7833	120.9579	125.8129
C <sub>17</sub> -C <sub>7</sub> -S <sub>14</sub>	128.8208	128.7452	123.1707	128.933	128.8503	123.2158
<b>Dihedral angle</b>						
C <sub>21</sub> -C <sub>7</sub> -C <sub>17</sub> -S <sub>14</sub>	4.69675	4.95756	0.43354	5.99833	6.49452	3.58115
C <sub>22</sub> -C <sub>21</sub> -C <sub>16</sub> -C <sub>7</sub>	10.65527	9.30751	90.37907	13.30338	11.45619	90.72562

**Table S4.** Selected bond lengths, bond angles and torsional angles of **2a** and **2b**, using B3LYP 6-31G(d) in EA (bond lengths are in Å, dihedral angles are in degree)

Atom number	<b>2a</b>			<b>2b</b>		
	Gas	EA		Gas	EA	
	GS	GS	ES	GS	GS	ES
<b>Bond length</b>						
N <sub>27</sub> -C <sub>19</sub>	1.37472	1.38056	1.40637	1.36992	1.37646	1.40616
C <sub>22</sub> -C <sub>17</sub>	1.46067	1.45224	1.5003	1.4644	1.45498	1.4966
C <sub>17</sub> -C <sub>7</sub>	1.37006	1.36057	1.35851	1.37808	1.36868	1.35896
<b>bond angle</b>						
C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub>	128.7936	129.3646	121.7883	125.767	126.3136	121.4828
C <sub>17</sub> -C <sub>7</sub> -C <sub>8</sub>	124.2978	124.4839	127.4136	125.6106	125.7831	127.5687
C <sub>17</sub> -C <sub>7</sub> -S <sub>14</sub>	125.9027	125.77	121.8487	124.2608	124.0955	121.6937
<b>Dihedral angle</b>						
C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub> -S <sub>14</sub>	9.67094	11.02093	1.87261	12.1311	14.73794	5.01798
C <sub>23</sub> -C <sub>22</sub> -C <sub>17</sub> -C <sub>7</sub>	25.99396	24.52921	90.53349	43.00418	40.29802	90.79016

**Table S5.** Selected bond lengths, bond angles and torsional angles of **3a** and **3b**, using B3LYP 6-31G(d) in EA (bond lengths are in Å, dihedral angles are in degree)

Atom number	<b>3a</b>			<b>3b</b>		
	Gas	EA		Gas	EA	
	GS	GS	ES	GS	GS	ES
<b>Bond length</b>						
N <sub>26</sub> -C <sub>18</sub>	1.37231	1.35925	1.36057	1.37511	1.36094	1.3603
C <sub>21</sub> -C <sub>16</sub>	1.4341	1.42503	1.4858	1.42759	1.41614	1.48311

C <sub>16</sub> -C <sub>7</sub>	1.36895	1.37956	1.3572	1.37151	1.38468	1.35596
Bond angle						
C <sub>21</sub> -C <sub>16</sub> -C <sub>7</sub>	132.3531	132.4513	125.9703	131.6421	131.9982	125.5351
C <sub>16</sub> -C <sub>7</sub> -C <sub>8</sub>	126.9239	126.5275	132.5587	126.55	126.1485	132.6317
C <sub>16</sub> -C <sub>7</sub> -S <sub>13</sub>	123.0723	123.3217	118.2164	123.3237	123.5082	118.1283
Dihedral angle						
C <sub>21</sub> -C <sub>7</sub> -C <sub>16</sub> -S <sub>13</sub>	12.38786	14.97754	1.80818	14.16729	16.86787	3.09344
C <sub>22</sub> -C <sub>21</sub> -C <sub>16</sub> -C <sub>7</sub>	17.67245	16.81219	91.43491	20.00739	18.15935	91.37291

**Table S6.** Linear polarizability  $\alpha_0$ , first order hyperpolarizability  $\beta_0$  and second order hyperpolarizability  $\gamma$  in DMF solvent

	$\alpha_0$ ( $10^{-24}$ esu)			$\beta_0$ ( $10^{-30}$ esu)			$\gamma$ ( $10^{-36}$ esu)		
	B3LYP P	CAM- B3LYP	BHand HLYP	B3LYP	CAM- B3LYP	BHand HLYP	B3LYP	CAM- B3LYP	BHand HLYP
<b>1a</b>	65.05	58.98	58.98	296.94	222.06	222.06	660.33	475.28	475.28
<b>1b</b>	73.80	67.14	67.14	265.86	199.22	199.22	620.47	415.60	415.60
<b>2a</b>	72.22	65.48	65.48	382.37	337.45	337.45	606.59	680.80	680.80
<b>2b</b>	79.24	68.65	68.65	417.23	314.72	314.72	837.17	764.65	764.65
<b>3a</b>	85.54	77.51	77.51	486.32	416.81	416.81	776.73	851.88	851.88
<b>3b</b>	94.30	85.98	85.98	404.96	352.18	352.18	641.75	609.06	609.06

**Table S7.** Intrinsic hyperpolarizability of **1a-3b** in DMF solvent

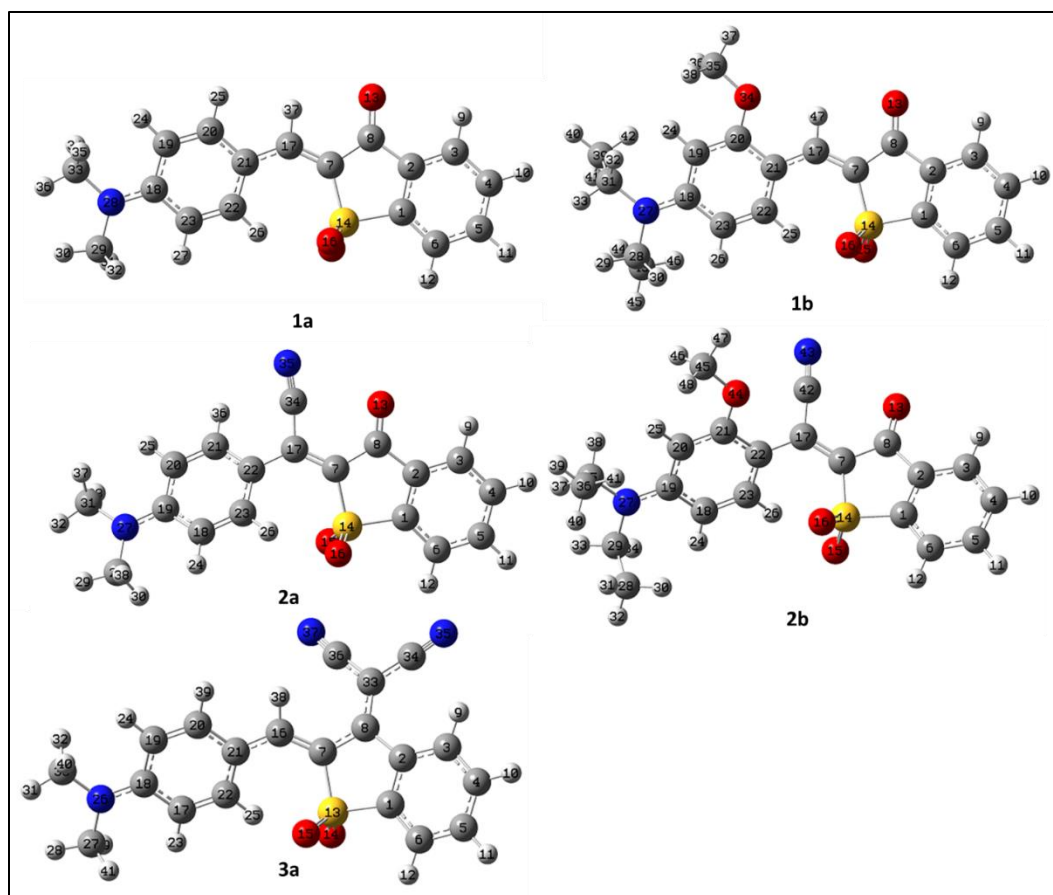
	$\beta_0$ ( $10^{-30}$ esu)			$\beta$ limit ( $10^{-30}$ esu)	$\beta^{int}$		
	B3LYP	CAM- B3LYP	BHandHLYP		B3LYP	CAM- B3LYP	BHandHLYP
<b>1a</b>	296.94	222.06	222.06	2530	0.1174	0.0878	0.0878
<b>1b</b>	265.86	199.22	199.22	2755	0.0965	0.0723	0.0723
<b>2a</b>	382.37	337.45	337.45	4096	0.0934	0.0824	0.0824
<b>2b</b>	417.23	314.72	314.72	4631	0.0901	0.0680	0.0680
<b>3a</b>	486.32	416.81	416.81	4976	0.0977	0.0838	0.0838
<b>3b</b>	404.96	352.18	352.18	5462	0.0741	0.0645	0.0645

**Table S8.** Intrinsic hyperpolarizability of **1a-3b** in EA solvent

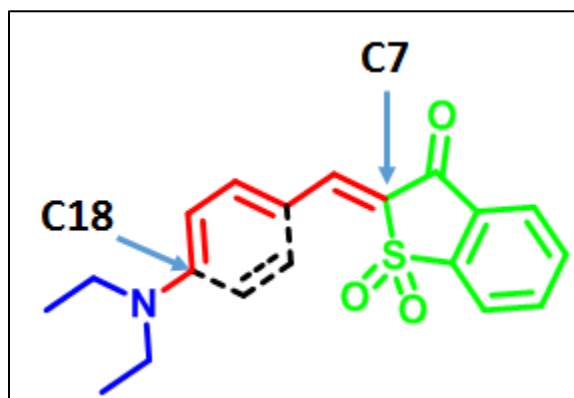
	$\beta_o$ ( $10^{-30}$ esu)			$\beta$ limit ( $10^{-30}$ esu)	$\beta^{int}$		
	B3LYP	CAM	BHandHLYP		B3LYP	CAM- B3LYP	BHandHLYP
<b>1a</b>	232.897	191.179	174.916	2005	0.1162	0.0954	0.0872
<b>1b</b>	211.325	172.061	156.862	2173	0.0973	0.0792	0.0722
<b>2a</b>	290.074	267.341	251.536	3188	0.0910	0.0839	0.0789
<b>2b</b>	319.072	250.598	237.762	3443	0.0927	0.0728	0.0691
<b>3a</b>	373.314	349.919	317.281	3853	0.0969	0.0908	0.0823
<b>3b</b>	322.589	295.244	269.641	4504	0.0716	0.0656	0.0599

**Table S9.** Amplitude of the sine shaped potential for **1a-3b**

Compound	Solvent	$V_o$
<b>1a</b>	DMF	0.107608
	EA	0.143918
<b>1b</b>	DMF	0.105557
	EA	0.140998
<b>2a</b>	DMF	-0.32594
	EA	-0.28399
<b>2b</b>	DMF	-0.47953
	EA	-0.44989
<b>3a</b>	DMF	-0.46877
	EA	-0.44106
<b>3b</b>	DMF	-0.45852
	EA	-0.43682



**Figure S1.** Optimized structures of 1a-3a



**Figure S2.** Polyene system considered for calculation of potential ( $V_0$ )