

SUPPLEMENTARY INFORMATION

**NiFe<sub>2</sub>O<sub>4</sub> nanoparticles decorated activated carbon nanocomposite based electrochemical sensor for selective detection of dopamine in presence of uric acid and ascorbic acid**

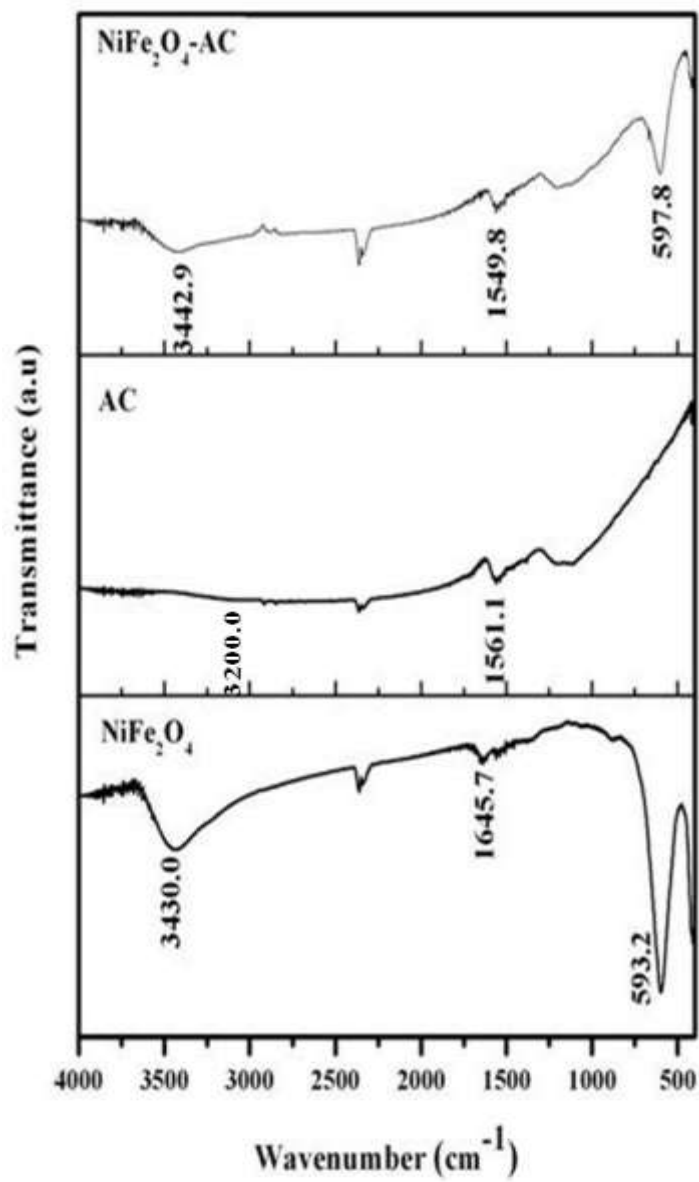
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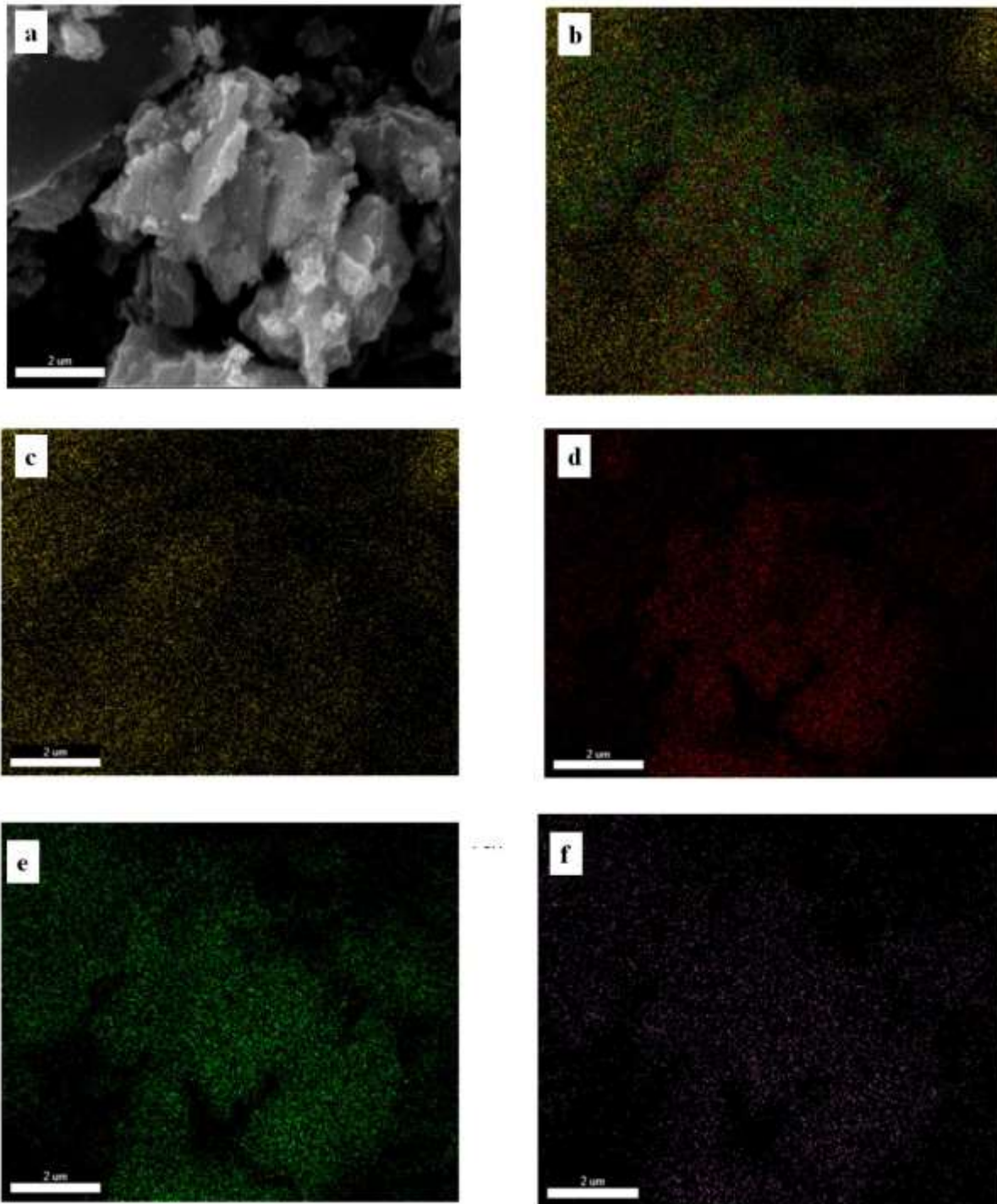
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**Table of Contents**

Figure S1. FTIR image of the nanocomposite.....	2
Figure S2. SEM image of the nanocomposite.....	3
Table S1. Impedance parameters obtained from the Nyquist plot of the modified electrode....	4
Figure S3. Cyclic voltammetric response of the nanocomposite.....	4
Estimation of active surface area using Randles-Sevcik equation.....	4
Table S2. Estimation of active surface area of the nanocomposite .....	5
Calculation of limit of detection.....	5
Figure S4. Standard deviation plot and DPV response of blank voltammogram on NiFe <sub>2</sub> O <sub>4</sub> -AC/GCE.....	5



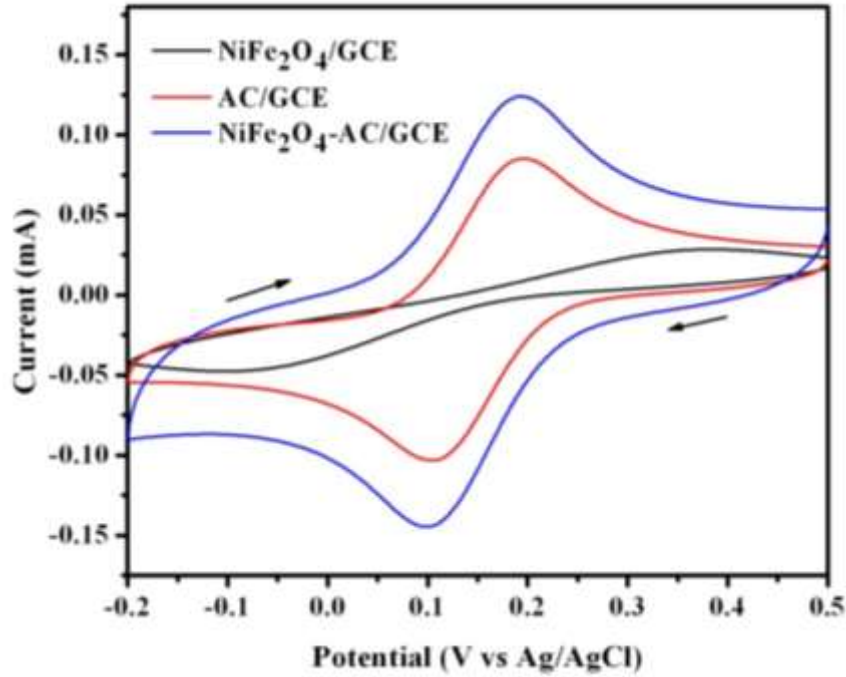
**Figure S1.** FTIR spectrum of NiFe<sub>2</sub>O<sub>4</sub>, AC and NiFe<sub>2</sub>O<sub>4</sub>-AC nanocomposite.



**Figure S2.** (a) SEM image of  $\text{NiFe}_2\text{O}_4\text{-AC}$ ; (b) their overall elemental mapping and (c to f) depicts the elemental mapping of individual constituents viz (c) C, (d) O, (e) Fe and (f) Ni.

**Table S1.** Impedance parameters obtained from the Nyquist plot of the modified electrode.

Modified electrode	$R_s$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )	$C_{dl}$ (F)	W
NiFe <sub>2</sub> O <sub>4</sub> /GCE	17.17	3517.00	$0.86 \times 10^{-7}$	0.000184
AC/GCE	19.64	43.05	$6.25 \times 10^{-7}$	0.000237
NiFe <sub>2</sub> O <sub>4</sub> -AC/GCE	21.99	14.46	$1.39 \times 10^{-6}$	0.000439



**Figure S3.** CV response on NiFe<sub>2</sub>O<sub>4</sub>/GCE, AC/GCE and NiFe<sub>2</sub>O<sub>4</sub>-AC/GCE from a solution of 1 mM K<sub>4</sub>[Fe(CN)<sub>6</sub>] and 0.1 M PBS at a scan rate of 100 mV s<sup>-1</sup>.

#### Estimation of Active surface area using Randles-Sevcik Equation

Given as

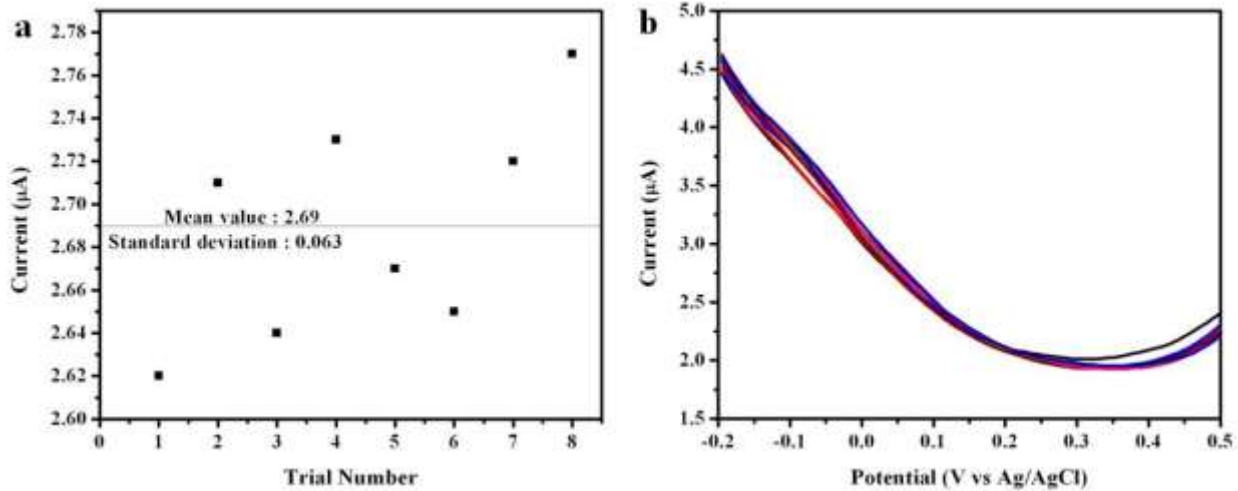
$$I_p = 2.65 \times 10^5 n^{3/2} A C D^{1/2} v^{1/2} \quad (1)$$

Where  $i_p$  is the peak current (mA),  $n$  is the number of electrons,  $A$  is the active surface area of the electrode (cm<sup>2</sup>),  $C$  is the concentration of K<sub>4</sub>[Fe(CN)<sub>6</sub>],  $D$  is the diffusion coefficient ( $7.6 \times 10^{-6}$  cm<sup>2</sup> s<sup>-1</sup>) and  $v$  is the scan rate (V s<sup>-1</sup>)

**Table S2.** Peak current and surface area parameters estimated from Randles-Sevcik equation.

Modified electrode	Anodic Peak current (mA)	Active Surface Area (cm <sup>2</sup> )
NiFe <sub>2</sub> O <sub>4</sub> /GCE	0.028	0.120
AC/GCE	0.084	0.362
NiFe <sub>2</sub> O <sub>4</sub> -AC/GCE	0.123	0.530

**Calculation of limit of detection (LOD):**



**Figure S4** (a) depicts the plot between current obtained at 0.06 V vs trial number constructed from the DPV response of blank voltammogram (without addition of DA) on NiFe<sub>2</sub>O<sub>4</sub>-AC/GCE in 0.1 M PBS. (b) shows the corresponding blank DPV response recorded in the range of -0.2 V to 0.5 V.

The limit of detection (LOD) is estimated by

$$\text{LOD} = (S_m - S_{bl})/m \quad (2)$$

where  $S_m$  is the minimum distinguishable analytical signal ( $\mu\text{A}$ ),  $S_{bl}$  is the mean value of the blank signal ( $\mu\text{A}$ ) and  $m$  is the slope of the calibration plot ( $\mu\text{A } \mu\text{M}^{-1}$ ).  $m = 0.47 \mu\text{A } \mu\text{M}^{-1}$

$S_m$  is defined as

$$S_m = S_{bl} + k\sigma_{bl} \quad (3)$$

Where  $k$  is a number illustrating the confidence level of the detection, usually 3 and  $\sigma_{bl}$  is the standard deviation of the blank ( $0.063 \mu\text{A}$ ).

hence equation (3) can be modified as

$$S_m - S_{bl} = 3 \sigma_{bl} \quad (4)$$

Substituting equation (4) in equation (2) we get

$$\text{LOD} = 3 \sigma_{bl} / m \quad (5)$$

By substituting the values of  $\sigma_{bl}$  and  $m$  in equation (5) the LOD is determined as

$$\begin{aligned} \text{LOD} &= (3 \times 0.063 \mu\text{A}) / 0.47 \mu\text{A} \mu\text{M}^{-1} \\ &= 0.4 \mu\text{M} \end{aligned}$$