

Anomalous absorption in H₂CN and CH₂CN molecules

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Abstract. Structures of H₂CN and CH₂CN molecules are similar to that of H₂CO molecule. The H₂CO has shown anomalous absorption for its transition $1_{11} - 1_{10}$ at 4.8 GHz in a number of cool molecular clouds. Though the molecules H₂CN and CH₂CN have been identified in TMC-1 and Sgr B2 through some transitions in ortho as well as in para species, here we have investigated the condition under which transitions $1_{11} - 1_{10}$ and $2_{12} - 2_{11}$ of these molecules may show anomalous absorption.

For the present investigation, we have calculated energy levels and radiative transition probabilities. However, we have used scaled values for collisional rate coefficients. We found that relative values of collisional rate coefficients can produce the required anomalous absorption in $1_{11} - 1_{10}$ and $2_{12} - 2_{11}$ transitions in the molecules.

Keywords. Interstellar molecules; H₂CN; CH₂CN.

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1. Introduction

Many radicals, molecular ions and other molecules which have short lifetimes in laboratories are found in many molecular clouds, as well as in the envelopes around evolved stars. Radio astronomers have discovered many molecular lines from the interstellar clouds. Interstellar clouds are neither uniform nor dynamically quiescent on long time-scales. The detailed investigation of the interstellar molecular clouds will give insight into the physical and chemical conditions of these objects. The physical conditions in the interstellar medium are rather extreme as compared to those in the laboratories. The densities are very low, typical values range from 1 to 10^5 particles per cm^{-3} and the temperature ranges from 5 to 60 K in cool molecular clouds such as taurus molecular cloud-1 (TMC-1) and up to several hundred Kelvins in the hot cores of Sagittarius B2 (Sgr B2).

H₂CN (methylene amidogen) is an *a*-type asymmetric top planar molecule having an electric dipole moment of 2.52 D [1]. Ohishi *et al* [2] reported detection of H₂CN molecule through its transition $1_{01} - 0_{00}$ in TMC-1 and transition $2_{02} - 1_{01}$ in Sgr B2. These transitions belong to the para species of the molecule. H₂CN has

a fractional abundance relative to H₂ of $\sim 1.5 \times 10^{-11}$ [2]. CH₂CN (cyanomethyl radical) is also an *a*-type asymmetric top planar molecule having an electric dipole moment of 1.621 D [3]. Irvine *et al* [4] reported detection of CH₂CN in TMC-1 and Sgr B2 through its transitions 1₀₁ – 0₀₀, 2₀₂ – 1₀₁, 2₁₂ – 1₁₁, 4₀₄ – 3₀₃, 5₀₅ – 4₀₄ and 5₁₄ – 4₁₃. Here, the transitions belong to both ortho as well as para species. CH₂CN has a fractional abundance relative to H₂ of $\sim 5 \times 10^{-9}$ [4].

The anomalous absorption was first detected for the transition 1₁₁ – 1₁₀ in H₂CO molecule [5]. Here, we are interested to find out the condition under which the transition 1₁₁ – 1₁₀ of H₂CN as well as CH₂CN molecules can be found in anomalous absorption. With the calculated radiative transition probabilities and scaled values of collisional rates, we have solved a set of statistical equilibrium equations coupled with the equations of radiative transfer. Following the information about collisional rates for H₂CO molecule, we modified the collisional rates for some transitions and found that the transitions 1₁₁ – 1₁₀ and 2₁₂ – 2₁₁ of the present molecules can also be found in anomalous absorption.

We propose identification of H₂CN molecule through its transitions 1₁₁ – 1₁₀ and 2₁₂ – 2₁₁ at 4.92 and 14.72 GHz, respectively and of CH₂CN molecule through the same transitions at 2.19 and 1.11 GHz, respectively, and through absorption against the cosmic microwave background (CMB). In the investigation, relative values of collisional rate coefficients are found to play an important role in the mechanism of anomalous absorption.

2. Modeling

We have calculated rotational energy levels (table 1) by using the molecular and distortional constants for H₂CN [6] and for CH₂CN [7]. The lines of our interest, 1₁₁ – 1₁₀ and 2₁₂ – 2₁₁ belong to the ortho species of both molecules. In the investigation, we have solved a set of 30 simultaneous equations coupled with 66 equations of radiative transfer for H₂CN molecule. For CH₂CN molecule, a set of 20 simultaneous equations coupled with 28 equations of radiative transfer are solved. The statistical equilibrium equations coupled with equations of radiative transfer can be expressed as follows:

$$n_i \sum_{\substack{j=1 \\ j \neq i}}^z P_{ij} = \sum_{\substack{j=1 \\ j \neq i}}^z n_j P_{ji}, \quad i = 1, 2, \dots, z,$$

where $z = 30$ for H₂CN and $z = 20$ for CH₂CN. Here, n 's are the population densities of the rotational levels and the parameter P is:

(i) For optically allowed transitions,

$$P_{ij} = \begin{cases} (A_{ij} + B_{ij}I_{\nu,bg})\beta_{ij} + n_{\text{H}_2}C_{ij}, & i > j \\ B_{ij}I_{\nu,bg}\beta_{ij} + n_{\text{H}_2}C_{ij}, & i < j \end{cases}.$$

(ii) For optically forbidden transitions,

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Table 1. Energies (in cm^{-1}) of rotational levels of ortho species of H_2CN and CH_2CN .

H ₂ CN					
$J_{k_a k_c}$	Energy	$J_{k_a k_c}$	Energy	$J_{k_a k_c}$	Energy
1 ₁₁	10.627	6 ₁₆	57.918	5 ₃₃	111.111
1 ₁₀	10.791	6 ₁₅	61.351	5 ₃₂	111.112
2 ₁₂	15.360	7 ₁₇	74.450	9 ₁₉	114.559
2 ₁₁	15.851	7 ₁₆	79.025	9 ₁₈	121.895
3 ₁₃	22.458	3 ₃₁	89.077	6 ₃₄	125.804
3 ₁₂	23.440	3 ₃₀	89.077	6 ₃₃	125.805
4 ₁₄	31.919	8 ₁₈	93.332	10 _{1,10}	138.127
4 ₁₃	33.555	4 ₃₂	98.869	7 ₃₅	142.948
5 ₁₅	43.740	4 ₃₁	98.870	7 ₃₄	142.952
5 ₁₄	46.193	8 ₁₇	99.208	10 ₁₉	147.078
CH ₂ CN					
$J_{k_a k_c}$	Energy	$J_{k_a k_c}$	Energy	$J_{k_a k_c}$	Energy
1 ₁₁	9.860	4 ₁₃	15.969	8 ₁₈	33.135
1 ₁₀	9.873	5 ₁₅	19.171	8 ₁₇	33.579
2 ₁₂	11.190	5 ₁₄	19.356	9 ₁₉	39.127
2 ₁₁	11.227	6 ₁₆	23.161	9 ₁₈	39.675
3 ₁₃	13.185	6 ₁₅	23.419	10 _{1,10}	45.769
3 ₁₂	13.259	7 ₁₇	27.815	10 ₁₉	46.449
4 ₁₄	15.846	7 ₁₆	28.161		

$$P_{ij} = n_{H_2} C_{ij},$$

where A 's and B 's are Einstein coefficients,

$$A_{ul} = \frac{2h\nu^3}{c^2} B_{ul}, \quad B_{ul} = \frac{g_l}{g_u} B_{lu},$$

where C 's are the collisional rate coefficients, n_{H_2} is the density of hydrogen molecules, and the escape probability β for the transition is

$$\beta_{lu} = \beta_{ul} = \frac{1 - \exp(-\tau_\nu)}{\tau_\nu},$$

where optical thickness τ_ν is

$$\tau_\nu = \frac{hc}{4\pi(dv_r/dr)} [B_{lu}n_l - B_{ul}n_u].$$

Here, (dv_r/dr) is the velocity gradient in the region. By solving these coupled equations, population densities of the levels are calculated. This set of equations is solved through iterative procedure for the given values of n_{H_2} and $\gamma \equiv n_{\text{mol}}/(dv_r/dr)$,

Table 2. Einstein *A*-coefficients of H₂CN and CH₂CN molecules.

H ₂ CN					
Transition	<i>A</i> -coeff. (s ⁻¹)	Transition	<i>A</i> -coeff. (s ⁻¹)	Transition	<i>A</i> -coeff. (s ⁻¹)
1 ₁₀ → 1 ₁₁	4.363 × 10 ⁻⁹	2 ₁₂ → 1 ₁₁	6.334 × 10 ⁻⁵	2 ₁₁ → 1 ₁₀	7.741 × 10 ⁻⁵
2 ₁₁ → 2 ₁₂	3.926 × 10 ⁻⁸	3 ₁₃ → 2 ₁₂	2.713 × 10 ⁻⁴	3 ₁₂ → 2 ₁₁	3.315 × 10 ⁻⁴
3 ₁₂ → 3 ₁₃	1.570 × 10 ⁻⁷	4 ₁₄ → 3 ₁₃	7.026 × 10 ⁻⁴	4 ₁₃ → 3 ₁₂	8.587 × 10 ⁻⁴
4 ₁₃ → 4 ₁₄	4.361 × 10 ⁻⁷	5 ₁₅ → 4 ₁₄	1.435 × 10 ⁻³	5 ₁₄ → 4 ₁₃	1.754 × 10 ⁻³
5 ₁₄ → 5 ₁₅	9.811 × 10 ⁻⁷	6 ₁₆ → 5 ₁₅	2.547 × 10 ⁻³	6 ₁₅ → 5 ₁₄	3.112 × 10 ⁻³
6 ₁₅ → 6 ₁₆	1.923 × 10 ⁻⁶	7 ₁₇ → 6 ₁₆	4.113 × 10 ⁻³	7 ₁₆ → 6 ₁₅	5.025 × 10 ⁻³
7 ₁₆ → 7 ₁₇	3.418 × 10 ⁻⁶	3 ₃₁ → 2 ₁₂	6.906 × 10 ⁻⁶	3 ₃₁ → 3 ₁₂	4.459 × 10 ⁻⁶
3 ₃₁ → 4 ₁₄	6.993 × 10 ⁻⁷	3 ₃₀ → 2 ₁₁	6.973 × 10 ⁻⁶	3 ₃₀ → 3 ₁₃	4.393 × 10 ⁻⁶
3 ₃₀ → 4 ₁₃	7.078 × 10 ⁻⁷	8 ₁₈ → 7 ₁₇	6.209 × 10 ⁻³	4 ₃₂ → 3 ₁₃	1.547 × 10 ⁻⁵
4 ₃₂ → 4 ₁₃	1.129 × 10 ⁻⁵	4 ₃₂ → 5 ₁₅	1.776 × 10 ⁻⁶	4 ₃₂ → 3 ₃₁	3.636 × 10 ⁻⁴
4 ₃₁ → 3 ₁₂	1.579 × 10 ⁻⁵	4 ₃₁ → 4 ₁₄	1.101 × 10 ⁻⁵	4 ₃₁ → 5 ₁₄	1.798 × 10 ⁻⁶
4 ₃₁ → 3 ₃₀	3.636 × 10 ⁻⁴	8 ₁₇ → 7 ₁₆	7.583 × 10 ⁻³	8 ₁₇ → 8 ₁₈	5.652 × 10 ⁻⁶
5 ₃₃ → 4 ₁₄	2.651 × 10 ⁻⁵	5 ₃₃ → 5 ₁₄	2.018 × 10 ⁻⁵	5 ₃₃ → 6 ₁₆	3.013 × 10 ⁻⁶
5 ₃₃ → 4 ₃₂	1.063 × 10 ⁻³	5 ₃₂ → 4 ₁₃	2.749 × 10 ⁻⁵	5 ₃₂ → 5 ₁₅	1.944 × 10 ⁻⁵
5 ₃₂ → 6 ₁₅	3.037 × 10 ⁻⁶	5 ₃₂ → 4 ₃₁	1.063 × 10 ⁻³	9 ₁₉ → 8 ₁₈	8.909 × 10 ⁻³
9 ₁₈ → 8 ₁₇	1.088 × 10 ⁻²	9 ₁₈ → 9 ₁₉	8.836 × 10 ⁻⁶	6 ₃₄ → 5 ₁₅	4.055 × 10 ⁻⁵
6 ₃₄ → 6 ₁₅	3.109 × 10 ⁻⁵	6 ₃₄ → 7 ₁₇	4.267 × 10 ⁻⁶	6 ₃₄ → 5 ₃₃	2.186 × 10 ⁻³
6 ₃₃ → 5 ₁₄	4.296 × 10 ⁻⁵	6 ₃₃ → 6 ₁₆	2.950 × 10 ⁻⁵	6 ₃₃ → 7 ₁₆	4.255 × 10 ⁻⁶
6 ₃₃ → 5 ₃₂	2.187 × 10 ⁻³	10 _{1,10} → 9 ₁₉	1.229 × 10 ⁻²	7 ₃₅ → 6 ₁₆	5.802 × 10 ⁻⁵
7 ₃₅ → 7 ₁₆	4.405 × 10 ⁻⁵	7 ₃₅ → 8 ₁₈	5.440 × 10 ⁻⁶	7 ₃₅ → 6 ₃₄	3.823 × 10 ⁻³
7 ₃₄ → 6 ₁₅	6.312 × 10 ⁻⁵	7 ₃₄ → 7 ₁₇	4.106 × 10 ⁻⁵	7 ₃₄ → 8 ₁₇	5.320 × 10 ⁻⁶
7 ₃₄ → 6 ₃₃	3.825 × 10 ⁻³	10 ₁₉ → 9 ₁₈	1.499 × 10 ⁻²	10 ₁₉ → 10 _{1,10}	1.321 × 10 ⁻⁵

CH ₂ CN					
Transition	<i>A</i> -coeff. (s ⁻¹)	Transition	<i>A</i> -coeff. (s ⁻¹)	Transition	<i>A</i> -coeff. (s ⁻¹)
1 ₁₀ → 1 ₁₁	7.7269 × 10 ⁻¹³	2 ₁₂ → 1 ₁₁	5.8179 × 10 ⁻⁷	2 ₁₁ → 1 ₁₀	6.147 × 10 ⁻⁷
2 ₁₁ → 2 ₁₂	6.9530 × 10 ⁻¹²	3 ₁₃ → 2 ₁₂	2.4930 × 10 ⁻⁶	3 ₁₂ → 2 ₁₁	2.634 × 10 ⁻⁶
3 ₁₂ → 3 ₁₃	2.7810 × 10 ⁻¹¹	4 ₁₄ → 3 ₁₃	6.4630 × 10 ⁻⁶	4 ₁₃ → 3 ₁₂	6.829 × 10 ⁻⁶
4 ₁₃ → 4 ₁₄	7.7250 × 10 ⁻¹¹	5 ₁₅ → 4 ₁₄	1.3220 × 10 ⁻⁵	5 ₁₄ → 4 ₁₃	1.397 × 10 ⁻⁵
5 ₁₄ → 5 ₁₅	1.7380 × 10 ⁻¹⁰	6 ₁₆ → 5 ₁₅	2.3490 × 10 ⁻⁵	6 ₁₅ → 5 ₁₄	2.482 × 10 ⁻⁵
6 ₁₅ → 6 ₁₆	3.4060 × 10 ⁻¹⁰	7 ₁₇ → 6 ₁₆	3.8000 × 10 ⁻⁵	7 ₁₆ → 6 ₁₅	4.015 × 10 ⁻⁵
7 ₁₆ → 7 ₁₇	6.0550 × 10 ⁻¹⁰	8 ₁₈ → 7 ₁₇	5.7470 × 10 ⁻⁵	8 ₁₇ → 7 ₁₆	6.072 × 10 ⁻⁵
8 ₁₇ → 8 ₁₈	1.0010 × 10 ⁻⁹	9 ₁₉ → 8 ₁₈	8.2910 × 10 ⁻⁵	9 ₁₈ → 8 ₁₇	8.731 × 10 ⁻⁵
9 ₁₈ → 9 ₁₉	1.5050 × 10 ⁻⁹	10 _{1,10} → 9 ₁₉	1.1390 × 10 ⁻⁴	10 ₁₉ → 9 ₁₈	1.208 × 10 ⁻⁴
10 ₁₉ → 10 _{1,10}	2.3510 × 10 ⁻⁹				

where n_{mol} is the density of the molecule and (dv_r/dr) is the velocity gradient in the region. In our investigation, non-local thermal equilibrium (NLTE) occupation numbers of the energy levels are calculated in an on-the-spot approximation by using the escape probability method [8], where the external radiation field, impinging on the volume element generating lines, is the CMB only.

The input data required in the investigation are the radiative transition probabilities (Einstein *A*-coefficients) and the collisional rate coefficients.

3. Einstein A-coefficients

Both H₂CN and CH₂CN are *a*-type asymmetric top molecules and the rotational radiative transitions are governed by the selection rules:

$$\begin{aligned} J & : \Delta J = 0, \pm 1 \\ k_a, k_c & : \text{even, odd} \longleftrightarrow \text{even, even} \\ & \quad \text{odd, even} \longleftrightarrow \text{odd, odd,} \end{aligned}$$

where J is the rotational quantum number and k_a and k_c are projections of J on the axis of symmetry in case of prolate and oblate symmetric tops, respectively. In the representation, where the axis of quantization is along the *a*-axis of inertia, Einstein *A*-coefficient for the transition $J'_{\tau'} \rightarrow J_{\tau}$ is given by [9,10]

$$A(J'_{\tau'} \rightarrow J_{\tau}) = \frac{64\pi^4\nu^3\mu^2(2J+1)}{3hc^3(2J'+1)} \left[\sum_{K=-J}^J g_{\tau K}^J g_{\tau' K}^{J'} C_{JK10}^{J'K} \right]^2, \quad (1)$$

where μ is the electric dipole moment of the molecule, $C_{JK10}^{J'K}$ the Clebsch Gordon coefficient and $\tau = k_a - k_c$ is a pseudoquantum number. Einstein *A*-coefficients for 66 radiative transitions between 30 energy levels of H₂CN and for 28 radiative transitions between 20 energy levels of CH₂CN are calculated. The values of *A*-coefficients are given in table 2.

4. Collisional rate coefficients

The collisional rate coefficients for H₂CN and CH₂CN are not available in literature and therefore, we have used scaled values for them. The rate coefficients for downward transitions $J'_{k'_a k'_c} \rightarrow J_{k_a k_c}$ at kinetic temperature T_k are taken as [11]

$$C(J'_{k'_a k'_c} \rightarrow J_{k_a k_c}) = 1 \times 10^{-11} (2J+1) \sqrt{\frac{T_k}{30}}. \quad (2)$$

This relation for collisional rate coefficients can be interpreted as cross-section times the thermal velocity. For the upward collisional rate coefficients, we accounted for the fact that downward and upward collisional rate coefficients are related through the detailed equilibrium [12]

$$C(J_{k_a k_c} \rightarrow J'_{k'_a k'_c}) = C(J'_{k'_a k'_c} \rightarrow J_{k_a k_c}) \frac{2J'+1}{2J+1} \exp\left(-\frac{\Delta E}{kT_k}\right), \quad (3)$$

where ΔE is the energy difference between the corresponding levels. In the absence of accurate collisional rates, our results can be treated as qualitative in nature.

5. Anomalous absorption

Temperature of the CMB (2.73 K) is supposed to be the minimum in the Universe. Thus, no radiation from any cosmic object are supposed to show absorption against

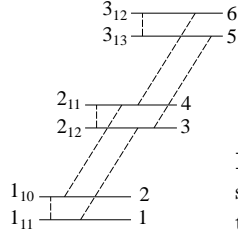


Figure 1. A set of three consecutive doublets corresponding to $J = 1, 2$ and 3 levels along with the radiative transitions between them.

the cosmic microwave background. The observation of a spectral line in absorption against the CMB is an unusual phenomenon, called as the anomalous absorption. The intensity, I_ν , of a line generated with homogeneous excitation conditions, is given by

$$I_\nu - I_{\nu,\text{bg}} = (S_\nu - I_{\nu,\text{bg}})(1 - e^{-\tau_\nu}), \quad (4)$$

where $I_{\nu,\text{bg}}$ is the intensity of the continuum against which the line is observed, τ_ν is the optical depth and S_ν is the source function. In the Rayleigh–Jeans limit, we have [13]

$$T_B = T_{\text{ex}} + (T_{\text{bg}} - T_{\text{ex}})e^{-\tau_\nu}. \quad (5)$$

For anomalous absorption, the excitation temperature T_{ex} , brightness temperature T_B and the background temperature T_{bg} satisfy the condition $0 < T_{\text{ex}} < T_B < T_{\text{bg}}$.

6. Results and discussion

In our model, free parameters are the molecular hydrogen density n_{H_2} and γ . Ohishi *et al* [2] and Irvine *et al* [4] found column density 10^{14} cm^{-2} for H_2CN and CH_2CN molecules, in Sgr B2 which corresponds to $\gamma = 10^{-5}$ for $\Delta v = 0.33 \text{ km s}^{-1}$. Therefore, we have taken $\gamma = 10^{-5}$ and $10^{-6} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$. In order to include large number of molecular clouds where these molecules may be found, n_{H_2} is varied over a wide range from 10^2 to 10^6 cm^{-3} , and calculations are performed for kinetic temperatures of 10, 20 and 30 K.

All energy levels of these molecules are in the form of doublets. The observed lines in absorption are due to the transitions between the levels of a doublet. In the terminology of Chandra *et al* [13], the levels 1, 2, 3, 4, 5 and 6 are 1_{11} , 1_{10} , 2_{12} , 2_{11} , 3_{13} and 3_{12} respectively, as shown in figure 1.

6.1 Anomalous absorption in the $1_{11} - 1_{10}$ transition

For anomalous absorption of the transition $1_{11} - 1_{10}$, in the optically thin limit a rough estimate made by Chandra *et al* [13] shows that rate coefficient C_{14} must be smaller than C_{23} . This criterion is similar to that derived by Townes and Cheung [14] semi-classically and by Green *et al* [15] on the basis of quantum mechanical calculations for H_2CO . Equation (2) here gives $C_{14} \approx C_{23}$ and therefore, we reduced

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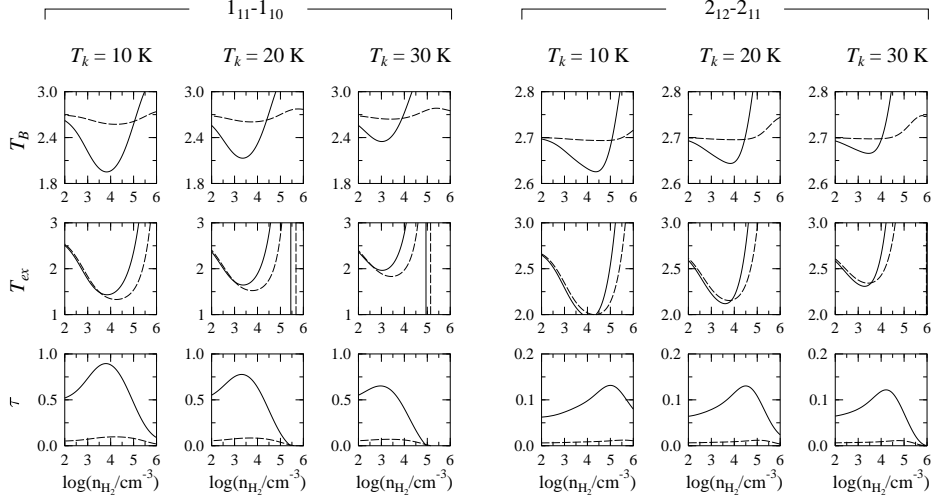


Figure 2. Variation of brightness temperature T_B (K), excitation temperature T_{ex} (K) and optical depth τ vs. hydrogen density n_{H_2} of the molecule H_2CN for kinetic temperatures 10, 20 and 30 K. Solid line is for $\gamma = 10^{-5} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$, and the dotted line for $\gamma = 10^{-6} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$. Collision rates of the transition $2_{11} - 1_{11}$, and its reverse are reduced by a factor of 2 with respect to that of the calculations from eq. (2).

the rate C_{41} and its reverse by a factor of 2 with respect to that of the calculations from eq. (2). This factor 2 is not very large. By doing so, the transition $1_{11} - 1_{10}$ is found to show the brightness temperature less than 2.73 K for kinetic temperatures 10–30 K (figure 2 for H_2CN and figure 3 for CH_2CN). However, transition $2_{12} - 2_{11}$ shows a weaker anomalous absorption than that of transition $1_{11} - 1_{10}$.

6.2 Anomalous absorption in the $2_{12} - 2_{11}$ transition

Following the rough estimate made by Chandra *et al* [13], to get anomalous absorption for transition $2_{12} - 2_{11}$, we have reduced the rates $C_{41}, C_{42}, C_{61}, C_{62}, C_{63}$ and their reverse by a factor of 2 with respect to that of the calculations from eq. (2). After reducing the rate coefficients, anomalous absorption in both the transitions $1_{11} - 1_{10}$ and $2_{12} - 2_{11}$ is found to increase (figure 4 for H_2CN and figure 5 for CH_2CN). In both situations anomalous absorption in transition $1_{11} - 1_{10}$ is stronger than that of transition $2_{12} - 2_{11}$. The regions with kinetic temperature around 10 K and molecular hydrogen density around 10^4 cm^{-3} are the best sites for the detection of these molecules through anomalous absorption of $1_{11} - 1_{10}$ and $2_{12} - 2_{11}$ transitions. With the increase of kinetic temperature (T_k), position of the minimum value of brightness temperature (T_B) and maximum value of optical depth (τ) are found to shift toward the low density region and the absorption is found to decrease.

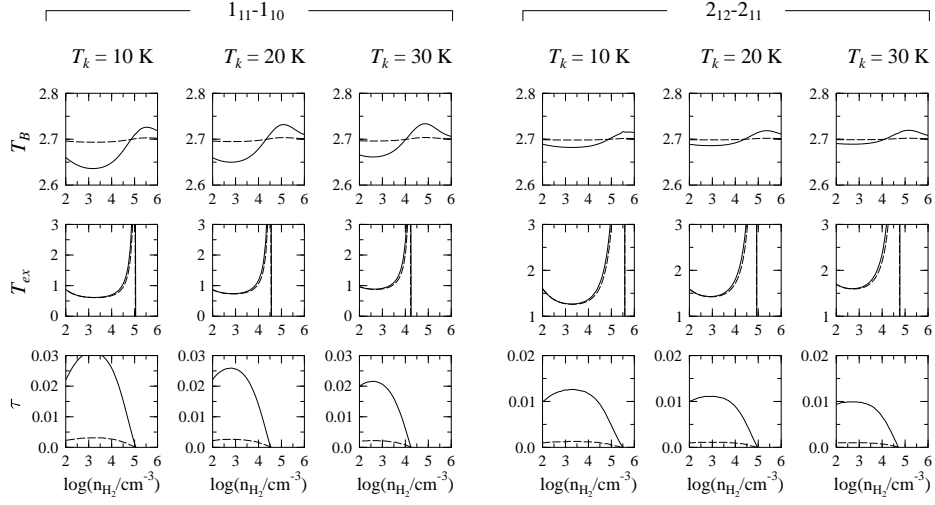


Figure 3. Variation of brightness temperature T_B (K), excitation temperature T_{ex} (K) and optical depth τ vs. hydrogen density n_{H_2} for kinetic temperatures 10, 20 and 30 K. Solid line is for $\gamma = 10^{-5} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$, and the dotted line for $\gamma = 10^{-6} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$. Collision rates of the transition $2_{11} - 1_{11}$, and its reverse are reduced by a factor of 2 with respect to that of the calculations from eq. (2).

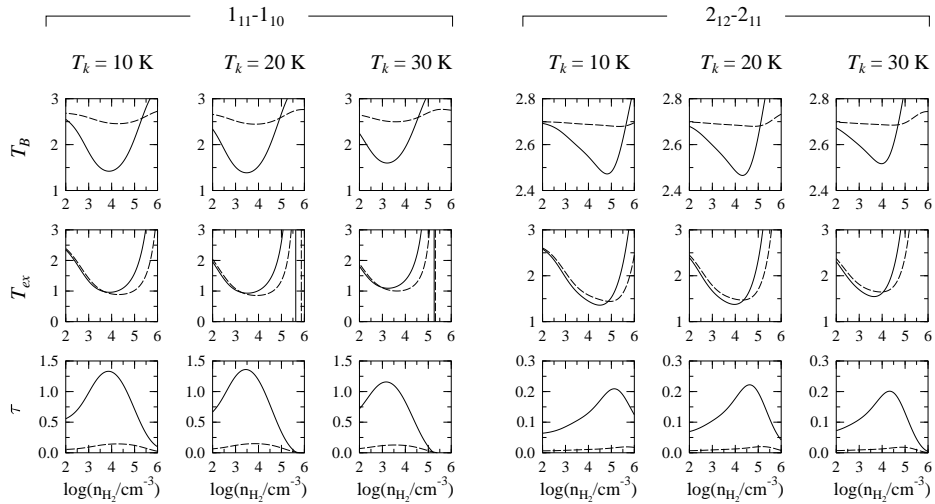


Figure 4. Same as figure 2 but the collision rates of the transitions $2_{11} - 1_{11}$, $3_{12} - 1_{11}$, $3_{12} - 1_{10}$, $3_{12} - 2_{12}$, $2_{11} - 1_{10}$ and their reverse are reduced by a factor of 2 with respect to that of the calculations from eq. (2).

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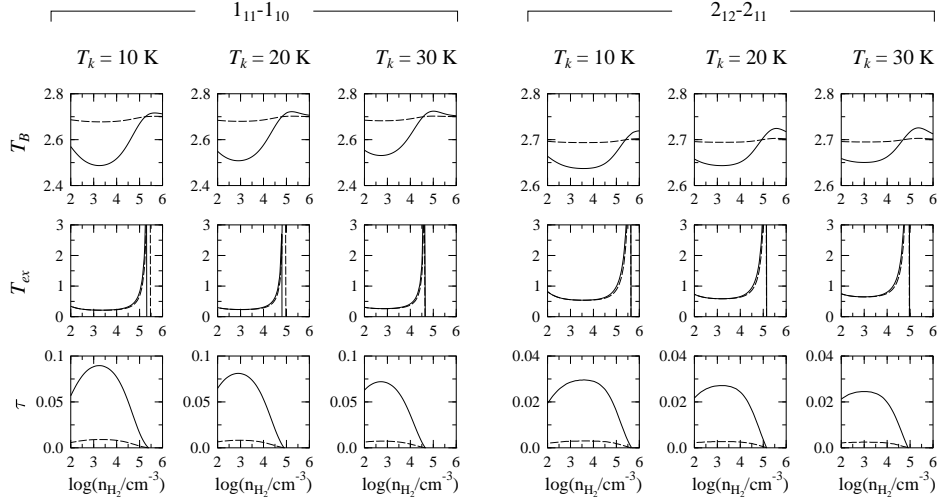


Figure 5. Same as figure 3 but the collision rates of the transitions $2_{11} - 1_{11}$, $3_{12} - 1_{11}$, $3_{12} - 1_{10}$, $3_{12} - 2_{12}$, $2_{11} - 1_{10}$ and their reverse are reduced by a factor of 2 with respect to that of the calculations from eq. (2).

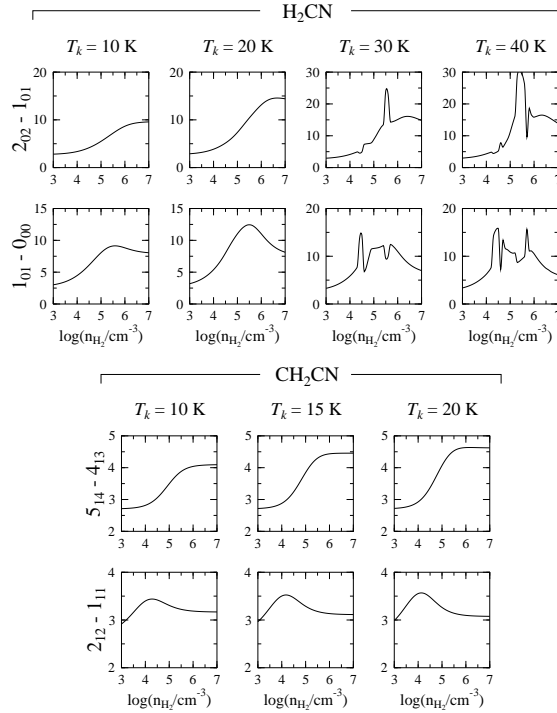


Figure 6. Variation of brightness temperature T_B vs. hydrogen density n_{H_2} at different kinetic temperatures and $\gamma = 10^{-5} \text{ cm}^{-3} (\text{km/s})^{-1} \text{ pc}$ for observed lines.

We have also calculated the brightness temperature T_B , excitation temperature T_{ex} and optical depth τ for the observed lines $1_{01} - 0_{00}$ and $2_{02} - 1_{01}$ of H_2CN and lines $2_{12} - 1_{11}$ and $5_{14} - 4_{13}$ of CH_2CN . The results confirm the emission feature of these lines. The fluctuation in the brightness temperature for the transitions $1_{01} - 0_{00}$ and $2_{02} - 1_{01}$ for temperatures 30 and 40 K (figure 6) are due to numerical instabilities. In principle, the variation should be smooth.

7. Conclusion

Our investigation shows that H_2CN and CH_2CN may be identified through anomalous absorption of $1_{11} - 1_{10}$ and $2_{12} - 2_{11}$ transitions in the cool cosmic objects. Though these molecules have been identified through emission lines, the anomalous absorption of the said lines can be used for confirmation purposes.

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