

A single molecule switch based on two Pd nanocrystals linked by a conjugated dithiol

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Abstract. Tunneling spectroscopy measurements have been carried out on a single molecule device formed by two Pd nanocrystals (dia. ~ 5 nm) electronically coupled by a conducting molecule, dimercaptodiphenylacetylene. The I - V data, obtained by positioning the tip over a nanocrystal electrode, exhibit negative differential resistance (NDR) on a background M-I-M characteristics. The NDR feature occurs at ~ 0.67 V at 300 K and shifts to a higher bias of 1.93 V at 90 K. When the tip is held in the middle region of the device, a Coulomb blockade region is observed ($\pm \sim 0.3$ V).

Keywords. Conducting molecule; nanocrystals; scanning tunneling microscopy; negative differential resistance.

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1. Introduction

Molecules were proposed as active electronic devices as early as 1974, by Aviram and Ratner who put forth the concept of unimolecular rectification [1]. Candidates for molecular wires and switches include porphyrins, phenylenes and thiophenes as well as their polymeric derivatives with extended π -conjugation [2,3]. Organic electronics using conducting polymers has seen a speedy growth in the last three decades but for molecular electronics to come of age, fabrication and measurement techniques would have to reach the nanoscale. With the advent of atomic imaging tools such as scanning tunneling microscopy (STM) and atomic force microscopy (AFM), along with nanolithography and fabrication techniques, it has become possible to carry out reliable measurements on individual molecules in a circuit. There have been several studies of electrical conduction through conjugated oligomers [4,5], typical examples being phenylene ethynylene molecules [6]. An early study based on DC conduction measurements on an Au cluster array cross-linked by 1,4-di(4-isocyanophenylethynyl)-2-ethylbenzene showed enhanced conductivity in the nanocrystal film revealing the conducting nature of the molecule [7]. Using STM, Bumm *et al* [8,9] reported that 4,4'-di(phenylene-ethynylene)benzothioacetate exhibits a relatively high conductivity compared to neighboring alkanethiols in a

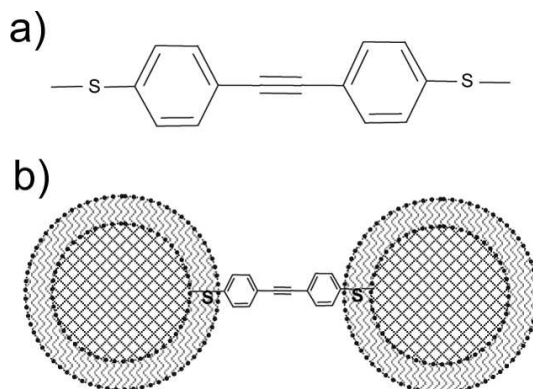


Figure 1. (a) Dimercaptodiphenylacetylene (dmdpa). (b) A schematic showing a dimer of Pd nanocrystals linked by a dmdpa molecule.

mixed self-assembled monolayer. A variant of this method has been tried out by anchoring Au nanocrystals on bifunctional molecules and performing I - V measurements by either STM or conducting AFM [10,11]. Fan *et al* [12] used a tuning fork tip method and observed features due to negative differential resistance (NDR) in a series of phenylenes. Chen *et al* [13,14] using the nanopore method, carried out a systematic study of the NDR effect in molecules containing a nitroamine redox center. In another study on molecules of bisthioterthiophene adsorbed on gold electrodes of a break junction, non-linear I - V curves with step-like features have been observed presumably involving conduction across discrete molecular levels [15]. More recently, Reichert *et al* [16] performed a study to compare I - V characteristics of symmetric and asymmetric phenylenes using the break junction method.

In spite of the studies cited above, there are certain unsettled issues related to electrical conduction through a metal-molecule-metal device. While issues such as the nature of molecule-electrode coupling [17,18] and the influence of a non-conducting molecular background in self-assembled monolayer [19] have caused intense debate, ensuring experimentally that only a single molecule actively present between the electrodes remains a problem. As part of continuing interest in electrical transport in molecular nanosystems [20,21], we have carried out measurements on dimercaptodiphenylacetylene linked to Pd nanocrystals of ~ 5 nm diameter on either side (figure 1), by employing low-temperature STM. We have found a clear signature of NDR which shifts to a higher energy upon lowering the temperature in this single molecule system.

2. Experimental

Dimercaptodiphenylacetylene (dmdpa) was prepared by a literature procedure [22] starting with a Sonogashira coupling reaction between 4-iodothioacetate and trimethylsilylacetylene. The resulting derivative was deprotected with tetrabutyl ammonium fluoride and coupled with 4-iodothioacetate followed by treatment with NaOH. Nanocrystals of Pd dispersed in ethanol were prepared following the

procedure of Teranishi and Miyake [23]. Briefly, 15 ml of 2.0 mM aqueous solution of H_2PdCl_4 was refluxed in a mixture of 10 ml of absolute ethanol and 18 ml of water in the presence of 33.3 mg PVP (Mw, 40,000 g mol^{-1}). In order to prepare dimeric nanocrystal species linked by the conducting molecule (see figure 1), 0.1 μl of 2 mM solution of dimercaptodiphenylacetylene in toluene was mixed with 1 ml of the Pd sol resulting in a Pd:dmdpa of 5000:1.

Transmission electron microscopy (TEM) was carried out using a JEOL 3010 operating at 300 kV. A drop of the sol treated with the conducting molecule was placed on a copper grid coated with thin carbon film and left to dry overnight. Scanning tunneling microscopy was performed using a SPM 32 (RHK technology, USA) attached to a low temperature stage in an ultra-high vacuum chamber. In order to prevent tip-induced damage and capture of the nanocrystal by the tip, typically a high impedance of 900 $\text{M}\Omega$ (bias, 900 mV, set-point current, 1 nA) was used for imaging. The microscope was initially calibrated against the 0001 surface of highly oriented pyrolytic graphite HOPG using an Au tip prepared by electrochemical means [24]. The same tip was used throughout the course of the study. After obtaining a stable non-drifting image, I - V data were collected in the spectroscopy mode by placing the tip atop a feature of interest with the feedback loop turned off. A typical voltage sweep was ± 2 V, with a sweep rate of 10 V/s and data interval of 5 mV. Imaging of the area was repeated after the I - V measurement to ensure that the nanocrystals had not drifted away during the I - V measurement. Lower temperatures were achieved by passing liquid nitrogen via a cold stage feedthrough. Temperature was read out using a thermocouple right below the sample mount.

3. Results and discussion

Figure 2a shows a TEM image revealing isolated near-spherical Pd nanocrystals from the as-prepared sol. The diameters of the nanocrystals are in the range of 3–7 nm as shown in the histogram in figure 2b, with a mean particle diameter of ~ 5 nm. The TEM image of Pd nanocrystals treated with dmdpa shown in figure 2c contains some primitive assemblies containing two or three nanocrystals. The observed assemblies, dimers and trimers, can only result from the linking of the Pd nanocrystals by the dithiol (dmdpa) molecules. In contrast, the as-prepared sol is completely devoid of such assemblies (figure 2a). Similar observations have been made previously [25] using bifunctional molecules. The inset in figure 2c shows a dimer species with well-resolved $\{111\}$ planes ($d \sim 2.2$ Å), although the particle boundaries are not discernible due to line of sight. The populations of the assemblies decrease with the nuclearity (figure 2d). This is rather an expected trend following a simple estimate based on the known Pd:dmdpa ratio and the assumption that one molecule links a pair of nanocrystals to form a dimer and three for a trimer. It would appear that the probability that a dimer is connected by two dmdpa molecules is negligibly small.

STM imaging was carried out over large areas ($\sim 500 \times 500$ nm^2) initially in order to isolate the monomeric and dimeric species of the nanocrystals. Figure 3 shows a typical STM micrograph of a region containing individual nanocrystals along

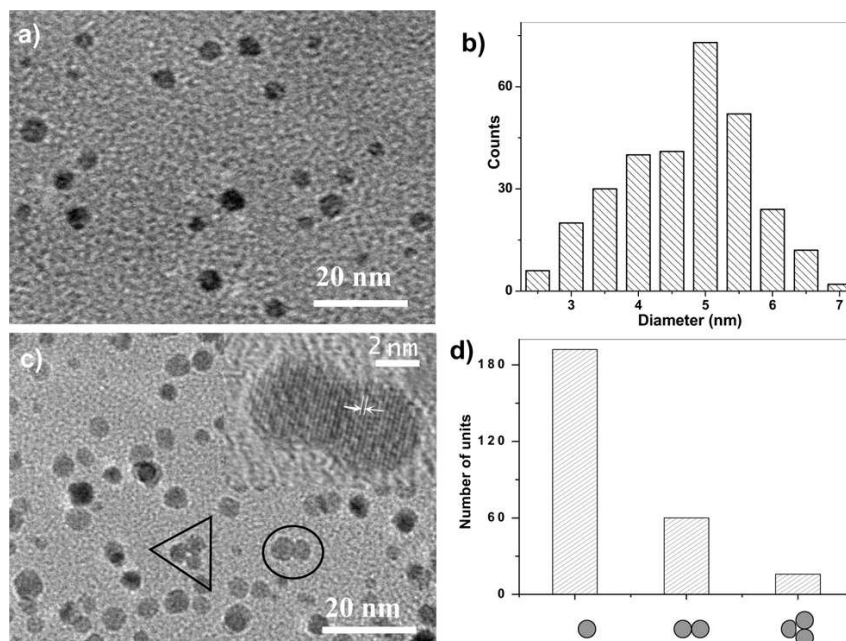


Figure 2. (a) TEM micrograph of Pd nanocrystals from the as-prepared sol. (b) Histogram showing size distribution of the nanocrystal in (a). (c) Micrograph taken after adding dmdpa molecules to the Pd sol shows several isolated nanocrystals along with assemblies containing two or three nanocrystals, as marked. Inset shows a high resolution micrograph of a dimeric species. The distance between Pd{111} lattice planes is ~ 2.2 Å. (d) Histogram showing the populations of monomers, dimers and trimers.

with a dimer. The nanocrystal diameters are somewhat larger (4–8 nm) than the diameters observed in TEM, the difference arising due to the PVP ligand shell around the nanocrystals. Tunneling spectroscopy measurements were carried out on monomers as well as on dimers of nanocrystals. Typical I - V data from an individual nanocrystal at 300 K shows Coulomb staircase behavior arising from incremental charging of the nanocrystal with the applied bias (figure 4). This observation is in accordance with earlier findings on polymer-coated metal nanocrystals [20]. I - V measurements on the dimeric species at 300 K obtained with the tip atop either of the nanocrystals reveal M-I-M characteristics but for the presence of a sharp feature (see figure 5a). Around 0.6 V, the current increases rapidly with increasing bias to reach its maximum at ~ 0.67 V and falls sharply thereafter. The presence of such a sharp feature in the I - V data is indicative of a negative differential resistance (NDR) region in the conductance behavior. The on-off ratio (peak-to-valley ratio) of this molecular device is 2.35. At 90 K, the NDR feature appears at a much higher bias of 1.93 V consistent with the temperature-dependent behavior of NDR in such molecules [13]. The background current is much less at the low temperature giving rise to a higher on-off ratio of 2.64.

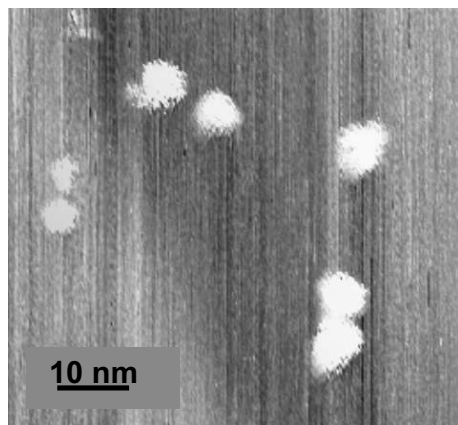


Figure 3. STM micrograph of Pd nanocrystals on HOPG substrate.

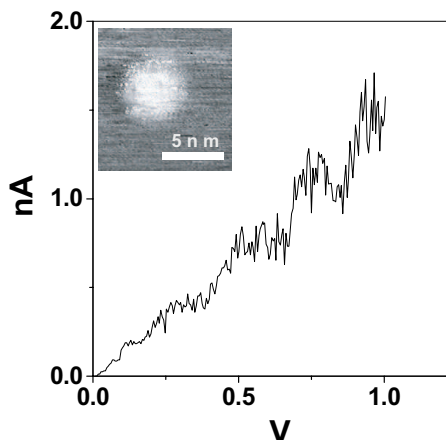


Figure 4. I - V data from an isolated Pd nanoparticle showing a Coulomb staircase behavior. Inset shows the STM micrograph.

When the tip is brought in the middle region of the nanocrystal dimer possibly right above the conducting molecule, we see an altogether different behavior (figure 5b). In this case, the I - V data are devoid of an NDR feature but exhibit a distinct Coulomb blockade region in the range, -0.3 and 0.25 V. We also observe an overall decrease in the current by at least an order of magnitude compared to the case in figure 5a. These observations can be understood on the basis of a notional circuit consisting of two RC segments representing the PVP covered nanocrystals, interlinked by a molecular resistance (see right side of figure 5). When the bias is applied on one of the nanocrystals using the STM tip, a voltage drop is expected along the molecular axis which now controls the overlap between the conduction states. An increase in the bias could result in an increased overlap between the conduction states giving rise to a sudden rise in the current till the states flip away at a critical bias (which in this case is, 0.67 V at room temperature) following say, charging or conformational change in the molecule [26]. When the bias is applied to the center of the device (figure 5b), the two molecular ends carry similar potential and no current is expected to flow in the circuit till the intrinsic gap of the molecule is overcome. This gives rise to the observed Coulomb blockade behavior as shown in figure 5b.

Having demonstrated clearly the switching action of this single molecular device, the merits of the present system deserves some mention. To our knowledge, this is the first study on electrical transport in dimercaptodiphenylacetylene, which is the simplest among the series of phenylene ethynylene oligomers. The molecule is electrically coupled to the Pd metal electrodes (nanocrystals) through Pd-S bonds. It has been suggested that the Pd-S linkage is the best option among many metal-molecule interfaces, the Au-S bond being the worst [17,18]! The method of making dimeric species of nanocrystals employed by us ensures that the electrical transport

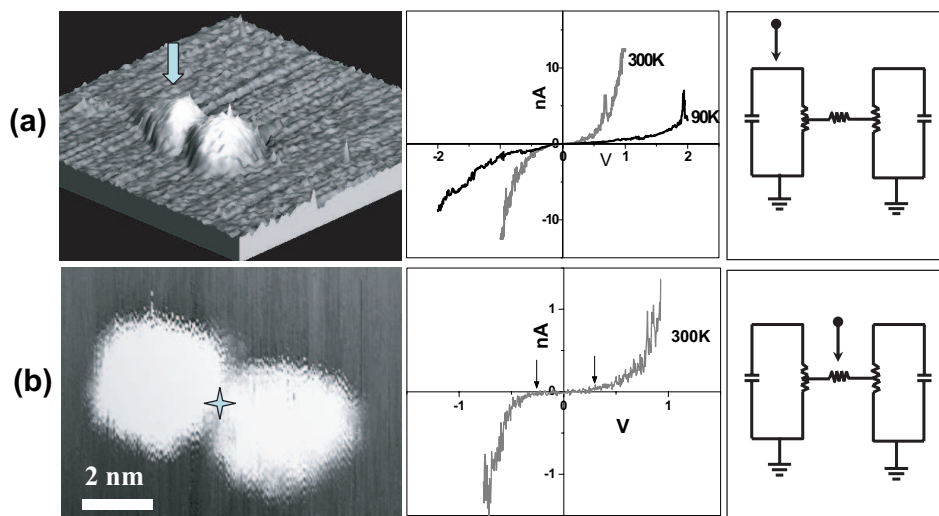


Figure 5. Tunneling microscopy and spectroscopy on a dimer of nanocrystals at 300 and 90 K. STM micrographs are shown on the left, I - V data in the middle and schematic circuits on the right side. (a) An isometric view, the tip bias being applied to one of the nanocrystals is shown with an arrow, (b) a top view with tip bias applied to the central region, as marked.

measurements indeed pertain to single molecules with a high degree of confidence, thereby avoiding interactions prevalent in a self-assembled monolayer of molecules. Furthermore, we are able to study the electrical behavior along the molecular axis. Thus, a Coulomb blockade ($\sim \pm 0.3$ V) occurs when the tip bias is applied to the central region of the molecule. The switching action of the device is evident in the form of the NDR feature when the bias is applied to one of the nanocrystal electrodes.

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References

- [1] A Aviram and M A Ratner, *Chem. Phys. Lett.* **29**, 277 (1974)
- [2] J Jortner and M Ratner, *Molecular electronics* (Blackwell Science, Oxford, United Kingdom, 1997)
- [3] D L Pearson, J M Tour, J S Schumm and L Jones II, *Synth. Metals* **84**, 303 (1997)
- [4] R M Metzger, *Chem. Rev.* **103**, 3803 (2003)
- [5] A Salomon, D Cahen, S Lindsay, J Tomfohr, V B Engelkes and C D Frisbie, *Adv. Mater.* **15**, 1881 (2003)

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- [6] J M Tour, L Jones II, D L Pearson, J J S Lamba, T P Burgin, G M Whitesides, D L Allara, A N Parikh and S V Atre, *J. Am. Chem. Soc.* **117**, 9529 (1995)
- [7] D B Janes, V R Kolagunta, R G Osifchin, J D Bielefeld, R P Andres, J I Henderson and C P Kubiak, *Superlattices and Microstructures* **18**, 275 (1995)
- [8] L A Bumm, J J Arnold, M T Cygan, T D Dunbar, T P Burgin, L Jones II, D L Allara, J M Tour and P S Weiss, *Science* **271**, 1705 (1996)
- [9] L A Bumm, J J Arnold, L F Charles, T D Dunbar, D L Allara and P S Weiss, *J. Am. Chem. Soc.* **121**, 8017 (1999)
- [10] G Leatherman, E N Durantini, D Gust, T A Moore, A L Moore, S Stone, Z Zhou, P Rez, Y Z Liu and S M Lindsay, *J. Phys. Chem.* **B103**, 4006 (1999)
- [11] A M Rawlett, T J Hopson, L A Nagahara, R K Tsui, G K Ramachandran and S M Lindsay, *Appl. Phys. Lett.* **81**, 3043 (2002)
- [12] F-R F Fan, J Yang, S M Dirk, D W Price, D Kosynkin, J M Tour and A J Bard, *J. Am. Chem. Soc.* **123**, 2454 (2001)
- [13] J Chen, M A Reed, A M Rawlett and J M Tour, *Science* **286**, 1550 (1999)
- [14] J Chen, W Wang, M A Reed, A M Rawlett, D W Price and J M Tour, *Appl. Phys. Lett.* **77**, 1224 (2000)
- [15] C Kergueris, J -P Bourgoin, S Palacin, D Esteve, C Urbina, M Magoga and C Joachim, *Phys. Rev.* **B59**, 12505 (1999)
- [16] J Reichert, R Ochs, D Beckmann, H B Weber, M Mayor and H v Lhneysen, *Phys. Rev. Lett.* **88**, 176804 (2002)
- [17] J M Semanario, C E D L Cruz and P A Derosa, *J. Am. Chem. Soc.* **123**, 5616 (2001)
- [18] J G Kushmerick, D B Holt, J C Yang, J Naciri, M H Moore and R Shashidhar, *Phys. Rev. Lett.* **89**, 086802 (2002)
- [19] G K Ramachandran, A M Rawlett, T J Hopson, L A Nagahara, R K Tsui and S M Lindsay, *Mat. Res. Soc. Symp. Proc.* **728**, S10.1.1 (2002)
- [20] P J Thomas, G U Kulkarni and C N R Rao, *Chem. Phys. Lett.* **321**, 163 (2000)
- [21] V V Agrawal, G U Kulkarni and C N R Rao, *J. Phys. Chem.* **B109**, 7300 (2005)
- [22] D L Pearson and J M Tour, *J. Org. Chem.* **62**, 1376 (1997)
- [23] T Teranishi and M Miyake, *Chem. Mater.* **10**, 594 (1998)
- [24] A J Nam, A Teren, J A Lusby and A J Melmed, *J. Vac. Sci. Technol.* **B115**, 2046 (1996)
- [25] L C Brousseau III, J P Novak, S M Marinakos and D L Feldheim, *Adv. Mater.* **11**, 447 (1999)
- [26] S Lakshmi and S K Pati, *J. Chem. Phys.* **121**, 11998 (2004)