



REGULAR ARTICLE

# Design, synthesis and characterization of functionalized pyrazole derivatives bearing amide and sulfonamide moieties from aza-aurones

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**Abstract.** In this work, we report the synthesis of new pyrazole derivatives bearing amide and sulfonamide frameworks from aza-aurones. Firstly, the intermediate spiropyrazolines were obtained through a highly regioselective 1,3-dipolar cycloaddition of nitrilimines with aza-aurones. Subsequently, the obtained cycloadducts were subjected to hydrochloric acid in hot ethanol which conducts to 5-(2-aminobenzoyl)-3,4-diaryl-1-phenylpyrazoles. Finally, the target compounds were obtained separately by the action of acetic anhydride, benzoyl chloride and tosyl chloride on the intermediates 2-aminobenzoylpyrazoles, respectively. Structures of all the synthesized compounds were established using IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR and mass spectroscopy.

**Keywords.** Aza-aurone; nitrilimine; spiropyrazoline; pyrazole; amide; sulfonamide.

## 1. Introduction

Heterocyclic compounds are of great interest, owing to a wide variety of applications they possess in many fields.<sup>1–6</sup> Among the heterocyclic compounds, pyrazole derivatives constitute the most studied compounds due to its widespread applications.<sup>7–15</sup> In addition, pyrazole moiety is an active pharmacophore which enters into the design of a broad spectrum of natural and synthetic bioactive compounds.<sup>16,17</sup> Further, the pyrazole bearing sulfonamide and amide moieties are an important scaffold for therapeutic agents, namely celecoxib,<sup>18</sup> deracoxib,<sup>19</sup> rimonabant,<sup>20</sup> sulfaphenazole,<sup>21</sup> pyrazofurin,<sup>22</sup> difenamizole<sup>23</sup> ..., etc. (Figure 1).

In general, the pyrazole compounds are prepared using two conventional approaches: cyclocondensation of hydrazine derivatives with 1,3-dielectrophilic

compounds and 1,3-dipolar cycloaddition.<sup>24,25</sup> There is a general consensus that 1,3-dipolar cycloaddition reaction is employed as the most practical method to provide highly substituted pyrazoles.<sup>26</sup>

Aza-aurones, namely 2-arylideneindolin-3(2*H*)-ones are known to possess various biological activities,<sup>27–31</sup> including anticancer,<sup>27,30</sup> antimalarial,<sup>28,29,31</sup> and antibacterial.<sup>32</sup> It has also been found that these interesting compounds are used as key building blocks to develop many heterocyclic compounds.<sup>33–38</sup> In addition, indolin-3-one moiety is frequently found in natural products and biologically active molecules.<sup>39–43</sup> Hence, the introduction of this framework in the core of the designed compounds may lead to new heterocycles with promising pharmacological activity.

As part of our ongoing research, we aim to synthesize heterocyclic compounds for therapeutic

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purposes.<sup>44–49</sup> Here, we report the synthesis of poly-substituted pyrazole-bearing amide and sulfonamide frameworks from aza-aurones. The synthesis proceeds through two successive reactions: 1,3-dipolar cycloaddition and *N*-alkylation.

## 2. Experimental

### 2.1 General information

All reagents and solvents were of AR grade. These chemicals were purchased from Sigma-Aldrich and other commercial suppliers, and were used without further purification. The progress of the reactions was monitored by TLC, performed on pre-coated Merck silica gel 60 F254 plates. Column chromatography was carried out using Merck silica gel (70–230 mesh), eluting with *n*-hexane and ether solution. Melting points were determined using a Kofler Bench apparatus. IR spectra were recorded using a Fourier Transform Bruker Vertex 70 Spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded at room temperature on a Bruker Avance II 300 Ultra-Shield (300 MHz for <sup>1</sup>H and 75 MHz for <sup>13</sup>C) spectrometer using CDCl<sub>3</sub> and DMSO-*d*<sub>6</sub> solvents. <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts ( $\delta$ ) are expressed in parts per million (ppm) with reference to the solvent used, and proton coupling constants (*J*) are reported in hertz (Hz). The spin multiplicities are reported as singlet (s), doublet (d), triplet (t), multiplet (m), doublet of doublets (dd), doublet of triplets (dt) and broad (br). Mass spectra were performed by the CESAMO (Bordeaux, France) on a Q Exactive Mass Spectrometer (Thermo). The instrument is equipped with an ESI source and the spectra were recorded in a positive mode. The spray voltage was maintained at 3200 V and capillary temperature was set at 320°C. Samples were introduced by injecting through a 20  $\mu$ L sample loop into a 300  $\mu$ L/min flow of methanol from the LC pump. Aza-aurones **1** was used as starting material in this work and were prepared according to the procedure described in the literature.<sup>35,50,51</sup>

### 2.2 General procedures

**2.2a Synthesis of aza-aurones (1a–b):** In a 100 mL three-necked flask containing 50 mL of NaOH (2N), equipped with a condenser, dropping funnel and a nitrogen bubbler. The aqueous solution of NaOH was purged with a nitrogen stream under magnetic stirring for 10 min. Diacetyloxyl (0.01 mol) was added to the solution and the mixture was heated under stirring

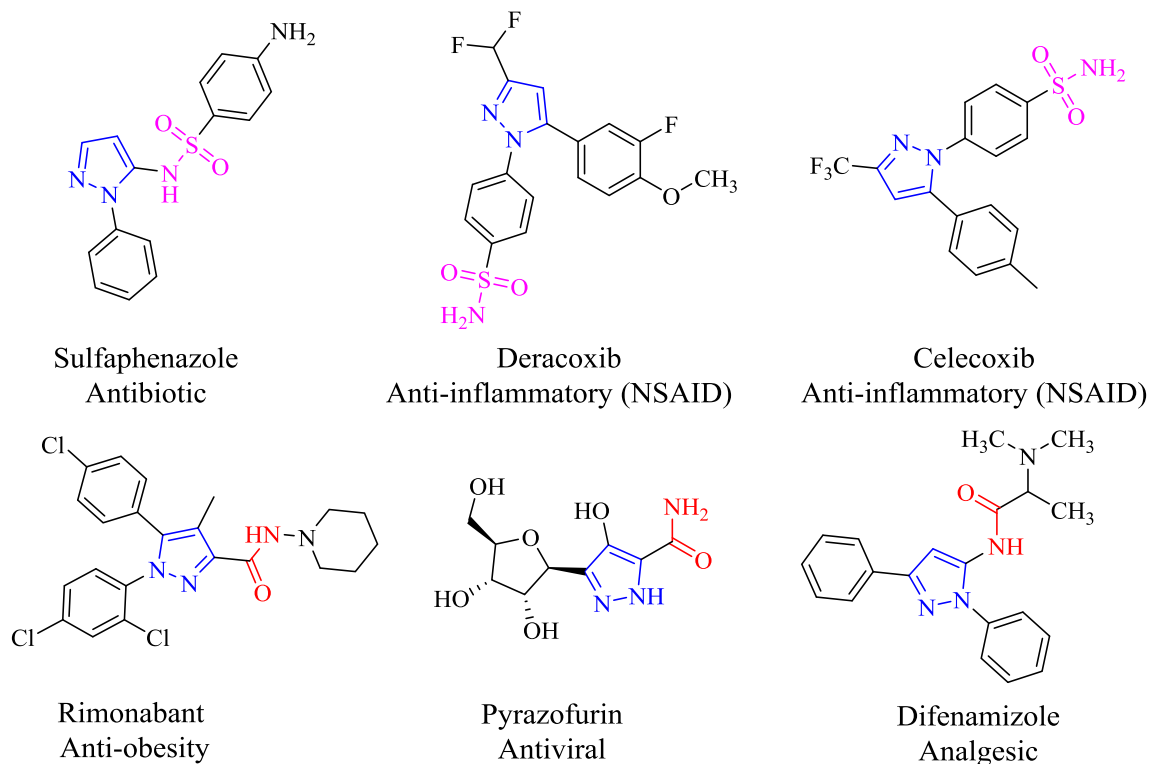
for 20 min. Then, the proper aldehyde (0.015 mol) dissolved in 4 mL of ethanol was added dropwise through the dropping funnel. The mixture was refluxed until the formation of an orange solid which was filtered and washed thoroughly with water. The obtained solid was recrystallized in ethanol to afford aza-aurones in moderate to good yield, which was used without further purification.

**2.2b Synthesis of spiropyrazolines (2aa–2db):** To a solution of appropriate aza-aurone **1** (1 mmol) in 20 mL of chloroform, the proper  $\alpha$ -chloroarylidene phenylhydrazone (1.2 mmol) and trimethylamine (1.2 mmol) was added. The mixture was held under stirring at room temperature, and the progress of the reaction was monitored by TLC. After the reaction completed, the solvent was removed to provide a crude product. The obtained residue was crystallized in ethanol and the formed precipitate was filtered and washed with cold ethanol.

**2.2c Synthesis of 5-(2-aminobenzoyl)-3,4-diaryl-1-phenylpyrazoles (3aa–3db):** In a 100 mL flask equipped with a condenser, cycloadduct **2** (1 mmol) was dissolved in ethanol (40 mL), and three drops of hydrochloric acid was added to the solution. The mixture was refluxed until the completion of the reaction, as monitored by TLC. After the 3/4 of the solvent was removed by vacuum, 60 mL of cold water was added. The formed precipitate was filtered and washed thoroughly with water. The obtained solid was purified by recrystallization in ethanol.

**2.2d Synthesis of pyrazoles-bearing benzamide moiety (4aa–4db):** To a solution of appropriate compound **3** (1 mol) in 20 mL of absolute DCM, triethylamine (2 mL) was added and the solution was stirred at 0°C in an ice bath. Benzoyl chloride (1 mol) was added in small portions while maintaining the temperature up to 0°C. Then, the reaction mixture was kept under magnetic stirring at room temperature. The progress of the reaction was monitored by TLC. After completion of the reaction, the mixture was transferred into a separatory funnel and washed several times with water. The organic layer was dried over anhydrous sodium sulfate (Na<sub>2</sub>SO<sub>4</sub>) and the solvent was removed under reduced pressure. The resulting residue was purified by column chromatography using hexane–ether mixture (4:1).

**2.2e Synthesis of pyrazole bearing acetamide moiety (5aa–5db):** In a 50 mL flask equipped with a



**Figure 1.** Pyrazoles bearing amide and sulfonamide moieties available as drugs.

condenser, 5-(2-aminobenzoyl)pyrazole (5 mmol) and acetic anhydride (20 mL) were introduced. The mixture was stirred at reflux until the completion of the reaction, as evidenced by TLC. The reaction mixture was poured into ice-cold water (200 mL) and extracted with DCM (3 x 20 mL). The DCM layer was washed thrice with a saturated solution of  $\text{NaHCO}_3$  and water. The organic layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$  and the solvent was removed under vacuum. The crude product was crystallized in ethanol and the solid obtained was filtered and used.

**2.2f Synthesis of pyrazole-bearing sulfonamide moiety (6aa–6da):** In a 100 mL Erlenmeyer flask, pyridine (6 mL) was added to a solution of 5-(2-aminobenzoyl)pyrazole **3** (0.019 mol) in 2 mL dichloromethane. The mixture was stirred under ice-cold condition. Tosyl chloride (0.019 mol) was added by fractions, maintaining the temperature of the reaction mixture up to  $0^\circ\text{C}$ . When the addition was completed, the stirring was continued at room temperature until TLC revealed the completion of the reaction. The mixture was quenched in ice-cold water and neutralized dilute HCl (pH = 2) and extracted with DCM (3 x 20 mL). The organic layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated under reduced pressure. The resulting

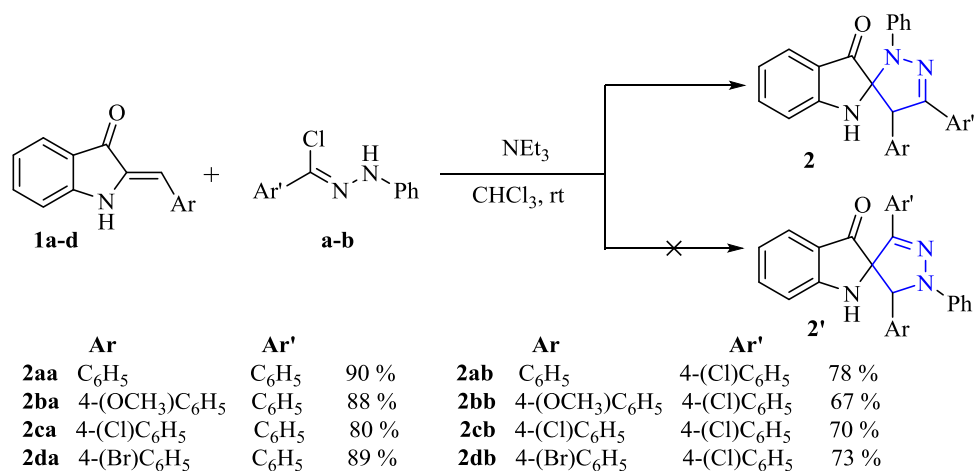
residue was crystallized in a mixture of EtOH–DCM (5:3).

### 3. Results and Discussion

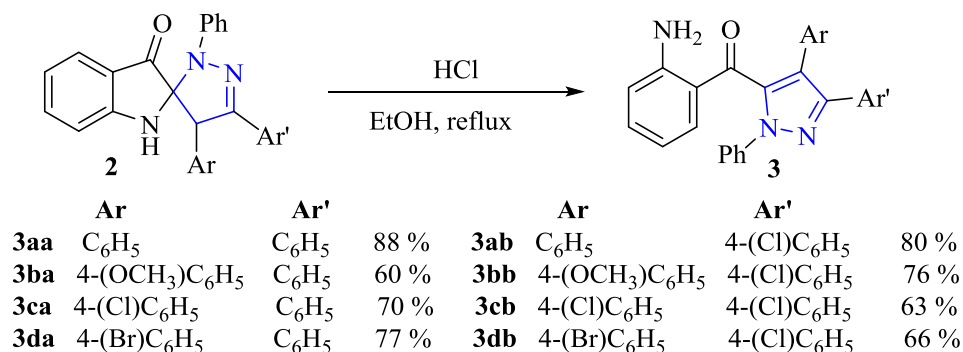
#### 3.1 Synthesis of spiropyrazolines (2)

Spiropyrazolines **2** were prepared via 1,3-dipolar cycloaddition reaction of *N*-phenyl-*C*-arylnitrilimines with aza-aurones **1** in chloroform at room temperature. Nitrilimine derivatives were generated *in situ* from the  $\alpha$ -chloroarylidene-phenylhydrazones under the action of triethylamine. The reaction led to the formation of a spiranic cycloadduct **2** as a unique regioisomer with a good yield (Scheme 1).

The regiochemistry proposed in this reaction is comparable to that observed with similar dipolarophiles, namely 2-arylidene-indan-1-ones,<sup>52</sup> 2-arylidene-tetral-1(2*H*)-ones<sup>53,54</sup> and 3-arylidene-chromen-4(3*H*)-ones.<sup>53</sup> This regiochemistry was also checked with  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data. Indeed,  $^1\text{H}$  NMR spectra of cycloadducts **2** showed a broad singlet at 4.5 ppm exchangeable with  $\text{D}_2\text{O}$  corresponding to NH proton and a singlet around 5 ppm due to pyrazoline protons. Whereas, with a regioisomer **2'**, the chemical shift of the pyrazoline protons would normally be deshielded under the effect of the



**Scheme 1.** 1,3-Dipolar cycloaddition reaction affords spiro-pyrazolines **2**.



**Scheme 2.** Synthetic pathway of 5-(2-aminobenzoyl)-3,4-diaryl-1-phenylpyrazoles **3**.

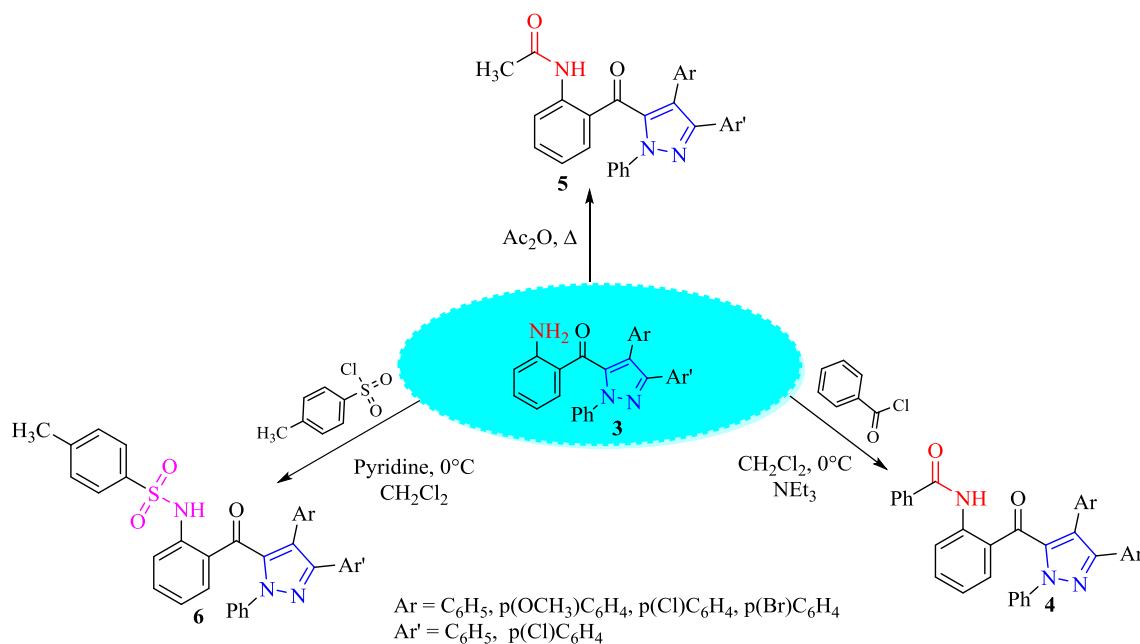
neighboring nitrogen atom. Further, <sup>13</sup>C NMR spectra showed chemical shift around 87 ppm assigned to spiranic carbon C<sub>52</sub>; for the regioisomer **2'** lower chemical shift values should be expected for the spiranic carbon C<sub>42</sub>. These outcomes corroborate further the proposed regiochemistry. In addition, FT-IR spectra of the cycloadducts **2** revealed, in particular, the presence of absorption bands attributed around 3300 cm<sup>-1</sup> and 1700 cm<sup>-1</sup>, respectively, to N-H and C=O stretching bands while the absorption band at 1600 cm<sup>-1</sup> was assigned to C=N stretching bands because of ring closure.

### 3.2 Synthesis of 5-(2-aminobenzoyl)-3,4-diaryl-1-phenylpyrazoles (**3**)

The prepared spiro-pyrazolines **2** were treated with concentrated hydrochloric acid in ethanol at reflux. The reaction undergoes ring-opening of the spiranic cycloadducts at the nitrogen atom of the indolin-

3(2*H*)-one frameworks and leads to the formation of 5-(2-aminobenzoyl)-3,4-diaryl-1-phenylpyrazoles **3** with good yield (Scheme 2).<sup>55</sup>

The chemical structures of pyrazoles **3** were established by spectroscopic data (FT-IR, <sup>1</sup>H and <sup>13</sup>C NMR). Indeed, FT-IR spectra of compounds **3** exhibited two characteristic bands ranging between 3200 and 3420 cm<sup>-1</sup>, indicating the presence of a primary amine NH<sub>2</sub> group. Absorption band around 1600 cm<sup>-1</sup> was assigned to C=O stretching band, which appear less than that observed on cycloadducts **2** spectra. The <sup>1</sup>H NMR spectra of compounds **3** showed in addition to signals corresponding to aromatic protons, a broad signal at 6.3 ppm assigned to NH<sub>2</sub> protons. Their <sup>13</sup>C NMR spectra displayed a signal at 189 ppm assigned to C=O carbon 10 ppm shielded, compared with the one observed on the spectra of the corresponding cycloadduct **2**. In addition, the disappeared signals at 5 ppm (pyrazoline proton) and 87 ppm (spiranic carbon), respectively, in <sup>1</sup>H and <sup>13</sup>C NMR spectra of compounds **3** provided the evidence for the ring-opening of spiranic cycloadducts **2**.



**Scheme 3.** Synthetic route of pyrazole derivatives.

**Table 1.** Physical and MS data of the synthesized pyrazoles.

Entry	Ar	Ar'	Time (h)	m.p. (°C)	Yield (%)	HRMS (m/z*)
<b>4aa</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	10	162–164	84	542.1832
<b>4ab</b>	C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	12	186–188	67	576.1439
<b>4ba</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	20	112–114	59	572.1930
<b>4bb</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	15	140–142	60	606.1538
<b>4ca</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	8	198–200	72	576.1437
<b>4cb</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	11	214–216	61	610.1058
<b>4da</b>	4-(Br)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	24	168–170	55	620.0931
<b>4db</b>	4-(Br)C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	18	224–226	70	654.0544
<b>5aa</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	6	170–172	61	480.1680
<b>5ab</b>	C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	3	196–198	53	514.1284
<b>5ba</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	4	184–186	66	510.1774
<b>5bb</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	7	216–218	70	544.1383
<b>5ca</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	8	180–182	58	544.1279
<b>5cb</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	6	212–214	77	548.0891
<b>5da</b>	4-(Br)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	2	190–192	72	558.0777
<b>5db</b>	4-(Br)C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	4	202–204	81	592.0387
<b>6aa</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	5	180–182	60	592.16635
<b>6ab</b>	C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	9	172–174	54	626.1272
<b>6ba</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	4	188–190	76	622.1756
<b>6bb</b>	4-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	8	182–184	66	656.1364
<b>6ca</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	10	196–198	80	626.1262
<b>6cb</b>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	4-(Cl)C <sub>6</sub> H <sub>5</sub>	7	208–210	71	660.0869
<b>6da</b>	4-(Br)C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	5	200–202	74	670.0757

\*Mass obtained for [M+Na]<sup>+</sup> ion.

### 3.3 Synthesis of pyrazoles bearing amide and sulfonamide

The synthetic pathway followed for the preparation of novel pyrazole derivatives is illustrated in Scheme 3. The 5-(2-aminobenzoyl)-3,4-diaryl-1-

phenylpyrazoles **3** on treating with benzoyl chloride, acetic anhydride and para-toluenesulfonyl chloride formed the required pyrazole derivatives containing amide and sulfonamide scaffold, which are known for their biological activities (Scheme 3).<sup>18,19,22</sup>

The structures of compounds **4**, **5** and **6** were confirmed by FT-IR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR and mass spectrometry analysis. The FT-IR spectra of compounds **4**, **5** and **6**, in particular, showed the presence of absorption bands around  $3300\text{ cm}^{-1}$  correspond to N-H bond stretching. Two absorption bands around  $1700\text{ cm}^{-1}$  and  $1600\text{ cm}^{-1}$  indicate the presence of C=O stretchings of carboxamide and carbonyl groups, respectively. Their  $^1\text{H}$  NMR spectra revealed a singlet at a range of 9–12 ppm exchangeable with  $\text{D}_2\text{O}$  corresponding to NH proton. Two singlets at 2.25 ppm and 2.44 ppm related to  $\text{CH}_3$  protons, respectively, of acetamide for compounds **5** and tosyl for compounds **6**. The  $^{13}\text{C}$  NMR spectra of the same compounds exhibited a signal at 166 ppm attributed to C=O carbon of carboxamide for compounds **4** and **5** as well as a signal at 192 ppm corresponding to C=O carbon of ketone. In addition, the signals between 21 and 25 ppm were assigned to methyl carbon  $\text{CH}_3$  for compounds **5** and **6**. About high-resolution mass spectrometry, spectra in positive mode have been recorded with an electrospray ionization chamber which led to  $[\text{M}+\text{Na}]^+$  adducts. All first molecular mass peaks were close to the theoretical value ( $\Delta_{\text{mass peak}} < 3\text{ ppm}$ ). Finally, spectroscopic data were found to be in good agreement with the proposed structures (Table 1).

#### 4. Conclusion

In this study, we prepared novel pyrazole derivatives-bearing amide and sulfonamide moieties through respective 1,3-dipolar cycloaddition and *N*-alkylation reactions. We have shown that the 1,3-dipolar cycloaddition reactions occur in a regioselective manner and conduct to one regioisomer **2** in each studied case. The spiropyrazoline **2** are developed into pyrazole-containing 2-aminobenzoyl frameworks, which are engaged in *N*-alkylation reactions through the amino group. The structures and regiochemistry of the cycloadducts and other obtained products were justified by spectroscopic techniques, *viz.*, IR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR and HR-mass. The structures of all synthesized products are in perfect agreement with spectroscopic data.

#### Supplementary Information (SI)

Spectroscopic data and copies of spectra (IR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, and HR-mass) of all obtained compounds are given in the supplementary information, which are available at [www.ias.ac.in/chemsci](http://www.ias.ac.in/chemsci).

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