

Enhanced antibacterial performance of Fe₃O₄-Ag and MnFe₂O₄-Ag nanocomposites

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Abstract. In this work, we have described the antibacterial activities of Fe₃O₄ nanoparticles with different organic parts, including Humic acid (HA), Nicotinic acid (Nico) and Histidine (His), and the antibacterial activity of MnFe₂O₄ nanoparticles coated with PANI and SiO₂ against different bacteria and some standard antibacterial drugs. The present study revealed that the newly fabricated various Fe₃O₄ and MnFe₂O₄ nanocomposites, when combined with some different organic parts, are superior antibacterial agents. Also, the synthesized nanocomposites can be easily separated from aqueous solution by magnetic filtration without any contamination of the medium.

Keywords. Antibacterial property; magnetic nanoparticles; silver.

1. Introduction

For over two decades, developments of organic parts integrated with magnetic nanoparticles have been reported because of their promising properties that are difficult to obtain separately with the individual components [1,2]. Many researchers have been trying to fabricate new composite materials by modifying the magnetic surface with organic molecules or inorganic parts, such as SiO₂, PANI, APTES, caffeic acid, folic acid, etc., since such new hybrid nanostructures have significant properties due to their unique multifunctional applications in industrial and biomedical fields [3–9].

Antibacterial materials are of great importance and necessity in our daily life. Many researchers are trying to develop and design these kinds of materials, which can be used very efficiently because of low cost for better antibacterial performance. For instance, Wang *et al* [10] prepared and developed an antibacterial and high light transmittance bulk materials to incorporate the hydrophobic and hydrophilic antibiotics [10]. In another work, Majumdar *et al* [11] developed an antibacterial material using polyester-silver nanocomposite fibres. Furthermore, Yadav *et al* [12] synthesized nickel-doped TiO₂ nanoparticles and reported visible light photocatalytic antibacterial activity against Gram-positive and Gram-negative bacteria. The nanomaterials, including silver, copper and some other metal or metal-based materials, are also efficient bacteriostatic agents [13].

Silver nanoparticles are reported to bear antimicrobial property against an array of pathogenic microorganisms but in order to improve the antibacterial activity of Ag NPs, some supports are needed. The combination of surface properties of Ag NPs and magnetic properties of magnetite leads to the potential applications of these magnetic nanocomposites in biological targeting, biological separation, high-density magnetic recording, catalysis and targeting therapy [14]. In addition, the introduction of silver nanoparticles into Fe₃O₄ enhances the biological activity of Ag NPs, but many studies have shown that the recovery of silver-based antibacterial agents is a general problem since they have a potential toxicity for both human and aqua systems and thus removing them from the disinfected media is efficiently required. The striking feature of Fe₃O₄ nanocomposites combined with silver nanoparticles is that they can be easily separated using an external magnet, providing a simple separation of the nanocomposite [2,15]. Finally, using these materials for killing bacteria under a magnetic field, we can expect a short sample preparation time and an easy cleanup procedure.

Here, we combined the collective advantages of magnetite that is coated with some organic molecules like Humic acid (HA), Nicotinic acid (Nico) and Histidine (His) with Ag nanoparticles and MnFe₂O₄ coated with PANI and SiO₂ with Ag NPs to produce multifunctional materials showing novel, good antibacterial activity and magnetic separability. Five magnetic nanocomposites, i.e., Fe₃O₄@HA@Ag, Fe₃O₄@His@Ag, Fe₃O₄@Nico@Ag, MnFe₂O₄@PANI@Ag and MnFe₂O₄@SiO₂@Ag, have been reported for the first time for their antibacterial activity against Gram-positive and Gram-negative bacteria in this study.

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2. Experimental

2.1 Synthesis of magnetic nanocomposites

The detailed experimental procedure for all products is given in our previous reports [16–21] except for the $\text{Fe}_3\text{O}_4@HA@Ag$ magnetic nanocomposite, which is newly synthesized.

2.1a Preparation of $\text{Fe}_3\text{O}_4@HA$ magnetic nanocomposites: $\text{Fe}_3\text{O}_4@HA$ nanocomposite was prepared by a simple reflux method. For synthesis of $\text{Fe}_3\text{O}_4@HA$ nanocomposite, $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ and $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ salts were taken in 1:2 ratio and dissolved in 50 ml of distilled water in a three-neck round-bottom flask and their homogeneous solution was prepared using magnetic stirring. Then 0.2 g of HA was added to this mixture under constant stirring. After that concentrated NH_3 solution was added drop-wise to adjust the pH level of the solution up to 10, at which the precipitation of all ferrites takes place. Then the flask was transferred to the heating mantle, where it was refluxed under Argon at 80°C for

2 h. The synthesized $\text{Fe}_3\text{O}_4@HA$ nanocomposites were separated using a permanent magnet and washed with distilled water several times to remove impurities. Finally a black powder product was obtained, which was dried at 80°C for 4 h.

2.1b Preparation of $\text{Fe}_3\text{O}_4@HA@Ag$ magnetic nanocomposites: The prepared $\text{Fe}_3\text{O}_4@HA$ nanocomposite (150 mg) was dispersed in 50 ml of deionized water. Then it was sonicated for 1 h, followed by the addition of 30 ml of AgNO_3 solution (0.2 mmol l^{-1}). The solution was vigorously stirred for 1 h, then 0.6 g of NaBH_4 was quickly added and the mixture was allowed to react for 2 h under rapid stirring. The product was separated magnetically and washed several times with deionized water to eliminate impurities.

2.2 Antibacterial study

2.2a Bacterial strains: Several doses of Fe_3O_4 , $\text{Fe}_3\text{O}_4@HA$, $\text{Fe}_3\text{O}_4@HA@Ag$, $\text{Fe}_3\text{O}_4@His$, $\text{Fe}_3\text{O}_4@His@Ag$, $\text{Fe}_3\text{O}_4@Nico$,

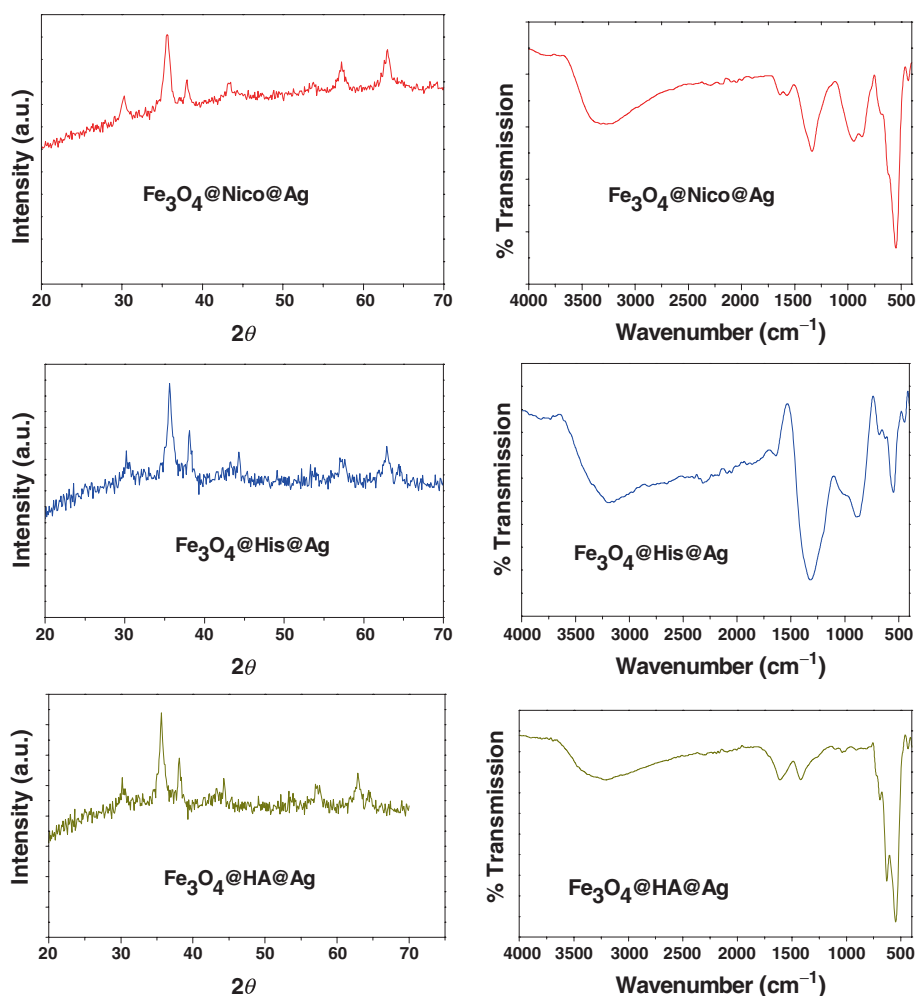


Figure 1. XRD and FT-IR results of $\text{Fe}_3\text{O}_4@Nico@Ag$, $\text{Fe}_3\text{O}_4@His@Ag$ and $\text{Fe}_3\text{O}_4@HA@Ag$ magnetic nanocomposites.

Fe₃O₄@Nico@Ag, MnFe₂O₄, MnFe₂O₄@PANI, MnFe₂O₄@PANI@Ag, MnFe₂O₄@SiO₂ and MnFe₂O₄@SiO₂@Ag in sterile distilled water were prepared. All these agents were examined against both Gram-positive bacteria such as *Staphylococcus aureus* (ATCC®25923™), *Bacillus subtilis* (ATCC®31785™) and *Enterococcus faecalis* (ATCC®29212™) and Gram-negative bacteria such as *Escherichia coli* (ATCC®25922™), *Pseudomonas aeruginosa* (ATCC®27853™) and *Burkholderia cepacia* (ATCC®25416™). All strains were stored at -80°C in stocks containing glycerol (2.5 M).

2.2b Agar well diffusion assay: In this study, four concentrations including 5, 10, 20 and 40 mg ml⁻¹ of Fe₃O₄, Fe₃O₄@HA, Fe₃O₄@HA@Ag, Fe₃O₄@His, Fe₃O₄@His@Ag, Fe₃O₄@Nico, Fe₃O₄@Nico@Ag, MnFe₂O₄, MnFe₂O₄@PANI, MnFe₂O₄@PANI@Ag, MnFe₂O₄@SiO₂ and MnFe₂O₄@SiO₂@Ag were prepared in sterile water and dispersed by sonication. All the magnetic materials have been screened for antimicrobial activity using the agar well diffusion method by measuring the inhibition zone in mm.

The assay was performed in triplicate based on the previous studies work of Sangnoi *et al* [22] with modifications.

Bacteria were first grown in Mueller Hinton agar (MHA; Merck, Germany), and incubated at 37°C for 24 h. Each bacterial strain in sterile saline was adjusted to 0.5 McFarland scale, which corresponds to 1.5 × 10⁸ CFU ml⁻¹. The agar well diffusion assay was applied to determine the difference in the antibacterial activity of sample MnFe₂O₄, MnFe₂O₄@PANI, MnFe₂O₄@PANI@Ag, MnFe₂O₄@SiO₂ and MnFe₂O₄@SiO₂@Ag [23,24]. Briefly, bacterial suspensions were adjusted to 10⁷-10⁸ CFU ml⁻¹ (McFarland turbidity standard no. 0.5) and spread on the MHA plates using a sterile triangular loop. This suspension was spread on MHA medium plates (thickness of approximately 4 mm). Following inoculation, a sterile glass borer was used to punch 5-mm wells into the surface of the agar. These wells were filled with 50 µl aqueous solution of the magnetic materials. Known standard antibiotic was filled into the well as a control at the centre of the plate. All plates were incubated overnight at 37°C. After incubation, microbial growth was determined by measuring the diameter of zone of inhibition. For each bacterial strain, controls were the selected standard antibiotics

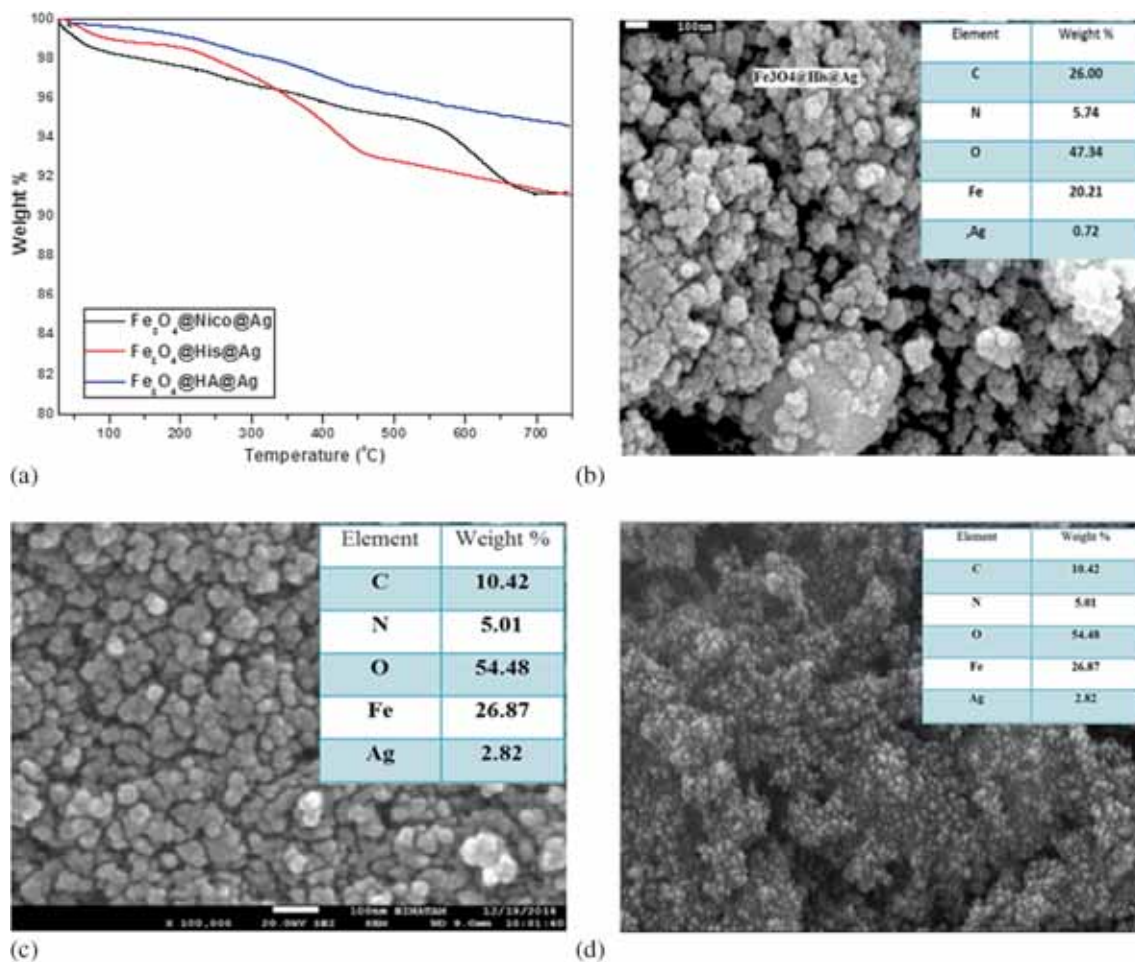


Figure 2. (a) TGA results of each magnetic nanocomposites and SEM result of (b) Fe₃O₄@Nico@Ag, (c) Fe₃O₄@His@Ag and (d) Fe₃O₄@HA@Ag magnetic nanocomposites.

such as streptomycin and amoxicillin. In this study, the selection of the standard antibiotics and their concentrations and the evaluation of the antimicrobial activity of the tested materials were based on the CLSI (Clinical and Laboratory Standards Institute) guidelines. The experiment was done three times and the clinical mean values are presented [25].

3. Results and discussion

3.1 Characterization of nanocomposites

3.1a Characterization of $Fe_3O_4@Nico@Ag$, $Fe_3O_4@His@Ag$ and $Fe_3O_4@HA@Ag$ nanocomposites: Complete XRD, FT-IR, TGA and SEM characterizations of $Fe_3O_4@Nico@Ag$, $Fe_3O_4@His@Ag$ and $Fe_3O_4@HA@Ag$ magnetic nanocomposite were given in our previous studies [10–13]. Briefly, in figure 1, the XRD result of each magnetic nanocomposites confirmed the presence of both Fe_3O_4 ((2 2 0), (3 1 1), (4 0 0), (4 2 2), (5 1 1), (4 4 0)) and Ag ((1 1 1) and (2 0 0)) NPs [10–13], and FT-IR results confirmed the successful modification of the Fe_3O_4 nanoparticles with the organic parts [16–19]. In figure 2a, TGA analysis of each nanocomposites shows the complete degradation of organic part [16–19], SEM analysis of each nanocomposites confirmed the homogeneity of the product (the presence of very tiny nanoparticles) and EDAX analysis indicated the

presence of Fe, Ag, C, O and N as the major elements in the product and absence of any impurity (figure 2b–d) [16–19].

3.1b Characterization of $MnFe_2O_4@PANI@Ag$ and $MnFe_2O_4@SiO_2@Ag$ nanocomposites: Similarly full XRD, FT-IR and SEM characterization of magnetic nanocomposite $MnFe_2O_4@PANI@Ag$ and $MnFe_2O_4@SiO_2@Ag$ were given in our reported study [20,21]. In short, the XRD result of each magnetic nanocomposites confirmed the presence of both $MnFe_2O_4$ ((2 2 0), (3 1 1), (4 0 0), (5 1 1) and (4 4 0)) and Ag NPs ((1 1 0), (2 0 0), (2 2 0)) [20,21]. FT-IR results confirmed the successful modification of the PANI and SiO_2 with $MnFe_2O_4$ in figure 3 [20,21]. SEM characterization of $MnFe_2O_4@PANI@Ag$ and $MnFe_2O_4@SiO_2@Ag$ nanocomposites confirmed the homogeneity of the product and EDAX analysis confirmed the presence of Mn, Ag, C, O and N and Mn, Fe, Ag, Si, and O for $MnFe_2O_4@PANI@Ag$ and $MnFe_2O_4@SiO_2@Ag$, respectively, as the major elements in the product and the absence of any impurity in figure 4 [20,21].

3.2 Results and discussion

3.2a Antibacterial performance of antimicrobial activities of various Fe_3O_4 nanocomposites with different coated organic parts including HA, His and Nico: Due to the

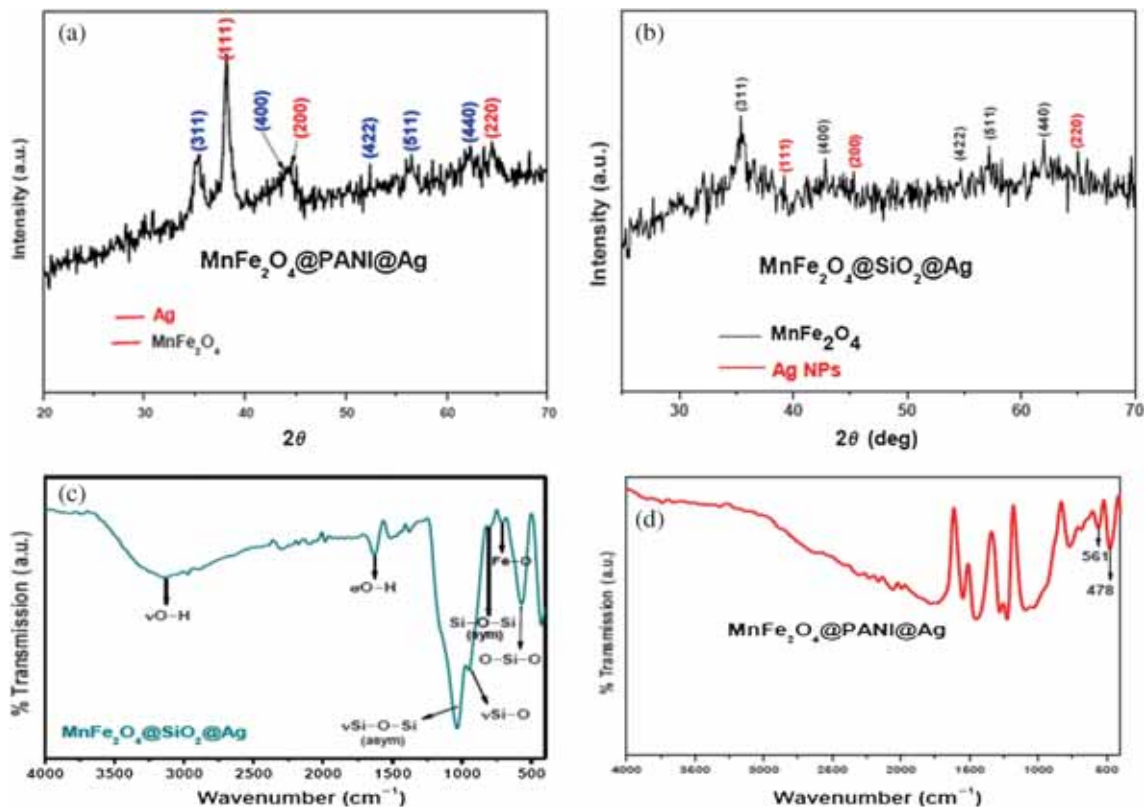


Figure 3. XRD and FT-IR results of $MnFe_2O_4@PANI@Ag$ and $MnFe_2O_4@SiO_2@Ag$ magnetic nanocomposites.

catalytic activity, optical and electronic properties, antimicrobial activity and magnetic properties, metal nanoparticles have been studied extensively [26]. Despite the fact that there have been many reports in novel studies allocated to studying nano-silver antibacterial activity, the study of silver-coated magnetic nanocomposite still requires investigation. Thus, the antibacterial effect of these nanocatalysts with different concentrations was investigated against both

Gram-positive strains such as *S. aureus*, *B. subtilis*, and *E. faecalis* and Gram-negative strains such as *P. aeruginosa*, *E. coli*, and *B. cepacia* using the agar diffusion method. The diameter of inhibition zones (in millimetres) around the wells are shown in table 1.

The antibacterial activities of amoxicillin and streptomycin showed bactericidal effect clearly against the test strains as positive controls. Fe₃O₄ alone had no effect

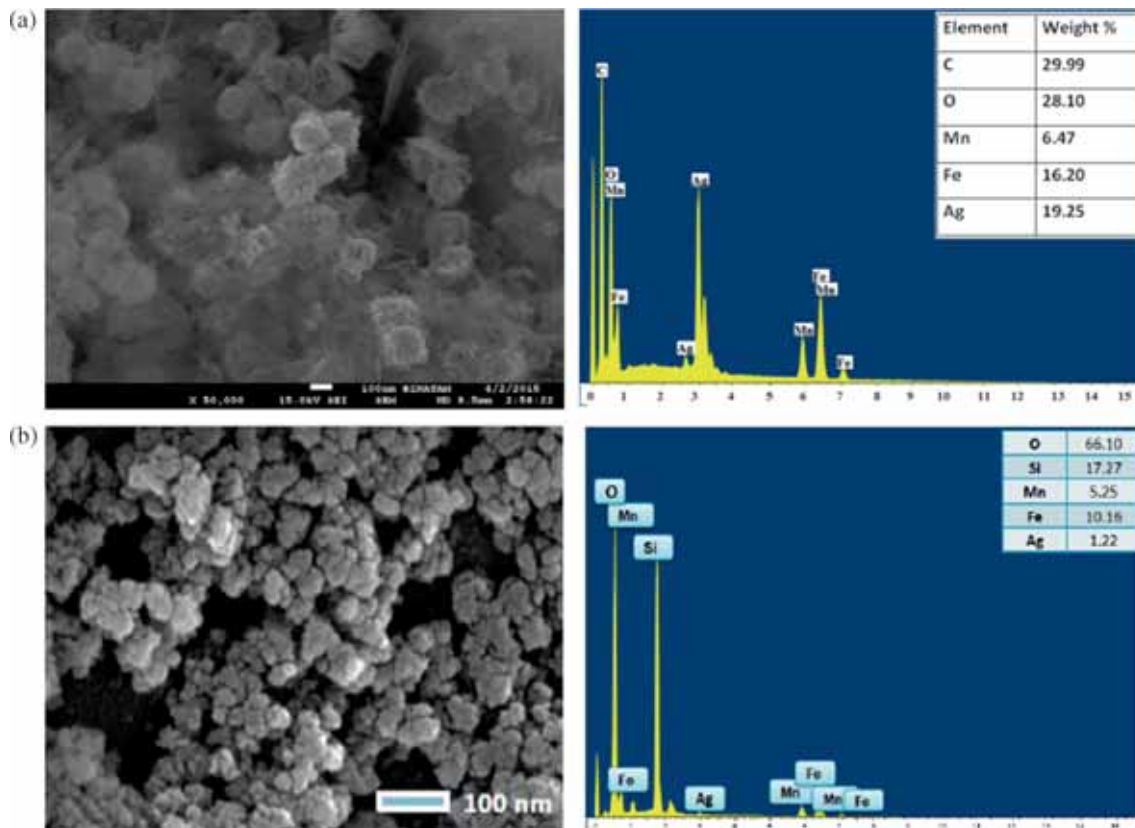


Figure 4. SEM and EDAX results of (a) MnFe₂O₄@PANI@Ag and (b) MnFe₂O₄@SiO₂@Ag magnetic nanocomposites.

Table 1. Antimicrobial activities of various Fe₃O₄ nanocomposites with different organic parts including HA, His and Nico coated.

Fe ₃ O ₄ nanocomposites C (mg ml ⁻¹)	Microorganisms																										
	Gram-positive												Gram-negative														
	<i>S. aureus</i>				<i>B. subtilis</i>				<i>E. faecalis</i>				<i>P. aeruginosa</i>		<i>E. coli</i>		<i>B. cepacia</i>										
	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40			
	Zone of inhibition (mm)																										
Fe ₃ O ₄	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—		
Fe ₃ O ₄ @HA	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—		
Fe ₃ O ₄ @HA@Ag	8	9	9	10	8	9	9	11	12	14	15	17	—	9	9	12	—	—	—	—	—	—	10	8	9	9	12
Fe ₃ O ₄ @His	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Fe ₃ O ₄ @His@Ag	10	10	11	12	9	10	12	13	16	17	19	20	11	12	13	15	10	10	14	14	10	13	14	16	16	16	16
Fe ₃ O ₄ @Nico	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Fe ₃ O ₄ @Nico@Ag	10	10	11	12	11	10	10	14	12	10	14	15	12	12	15	16	10	10	14	14	13	12	14	16	16	16	16
Amoxicillin (25 mg)		32				27				40				0			30				35						
Streptomycin (10 mg)		22				18				20				20			25				30						

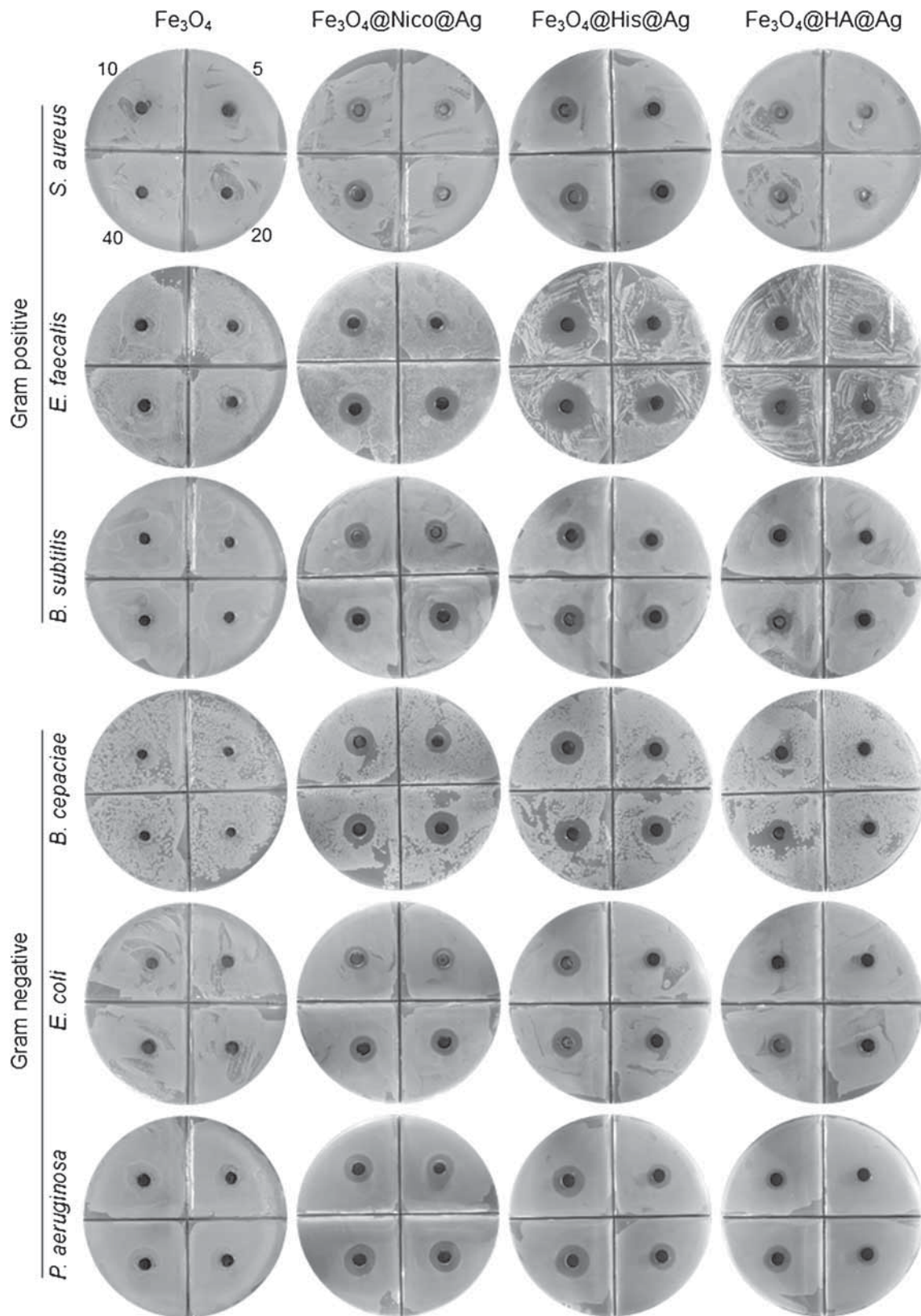


Figure 5. Antimicrobial effects of Fe_3O_4 @Nico@Ag, Fe_3O_4 @His@Ag and Fe_3O_4 @HA@Ag against Fe_3O_4 with different concentrations (5, 10, 20 and 40 mg ml^{-1}) on both Gram-positive and -negative bacteria.

Table 2. Antimicrobial activities of MnFe₂O₄@PANI, MnFe₂O₄@PANI@Ag and MnFe₂O₄@SiO₂@Ag.

Nanocatalyst C (mg ml ⁻¹)	Microorganisms																							
	Gram-positive												Gram-negative											
	<i>S. aureus</i>				<i>B. subtilis</i>				<i>E. faecalis</i>				<i>P. aeruginosa</i>				<i>E. coli</i>				<i>B. cepaciae</i>			
	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40	5	10	20	40
	Zone of inhibition (mm)																							
MnFe ₂ O ₄	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MnFe ₂ O ₄ @SiO ₂	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MnFe ₂ O ₄ @PANI	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MnFe ₂ O ₄ @SiO ₂ @Ag	10	11	16	15	10	11	13	13	11	12	13	13	11	13	17	15	8	10	12	11	12	13	15	13
MnFe ₂ O ₄ @PANI@Ag	10	11	13	14	13	11	14	12	12	13	12	14	7	10	13	11	11	12	12	13	12	14	12	15
Amoxicillin (25 mg)	32				27				40				0				30				35			
Streptomycin (10 mg)	22				18				20				20				25				30			

on tested bacteria and also Fe₃O₄@HA, Fe₃O₄@His and Fe₃O₄@Nico showed no antibacterial effects. On the other hand, the effects of Fe₃O₄@HA@Ag, Fe₃O₄@His@Ag and Fe₃O₄@Nico@Ag on the antibacterial activity were observed and the inhibition zones in diameter are presented in figure 5 and table 1. Generally, when the concentration of nanocatalysts including Ag NPs was increased, the antibacterial effects were increased. According to the comparison of the highest concentration of the nanocatalysts (40 mg ml⁻¹), the obtained biggest diameter of inhibition zone (table 1) for Fe₃O₄@HA@Ag was 17 mm against *E. faecalis*, for Fe₃O₄@His@Ag it was 20 mm against *E. faecalis* and for Fe₃O₄@Nico@Ag it was 16 mm against both *P. aeruginosa* and *B. cepacia*.

The lowest concentration of Fe₃O₄@HA@Ag had no antibacterial effect against *P. aeruginosa*; likewise, the concentration of 5, 10 and 20 mg ml⁻¹ had no effects against *E. coli*. However, good effect was seen at 40 mg ml⁻¹ as compared with amoxicillin antibiotic against *P. aeruginosa*. Fe₃O₄@His@Ag with all its concentrations showed the highest antibacterial effects as strong as that of streptomycin against *E. faecalis* strain among the others. The zone of inhibition was 20 mm, the same for streptomycin and half of the amoxicillin. Fe₃O₄@His@Ag against *E. faecalis* had a good antibacterial activity, the same as that of streptomycin, at the highest concentration of 40 mg ml⁻¹. Fe₃O₄@Nico@Ag with all its concentrations showed the highest inhibition zones against *P. aeruginosa* as 12, 12, 15 and 16 mm, correspondingly, as compared with amoxicillin. This antibiotic had no effect on this strain; however, streptomycin had.

According to obtained results, it is necessary to emphasize that the tested Fe₃O₄ nanocomposites including Ag NPs have shown generally bacteriostatic effects but sometimes bactericidal because Ag NPs have been known to have inhibitory and bactericidal effects [26,27]. The different susceptibilities of Gram-negative and Gram-positive strains towards these examined antibacterial agents possibly depend on their cell wall structure. The mechanism of the bactericidal effect of Ag may be due to its attaching to the surface of the cell membrane and disrupting the permeability and respiration. Various concentrations required for growth inhibition or

killing bacteria were determined by the biological properties of individual bacterial species.

3.2b Antibacterial performance of antimicrobial activities of various MnFe₂O₄@PANI@Ag and MnFe₂O₄@SiO₂@Ag nanocomposites: Additionally, MnFe₂O₄, which is also a superparamagnetic nanoparticle, with some modified nanocomposites such as MnFe₂O₄@PANI, MnFe₂O₄@PANI@Ag, MnFe₂O₄@SiO₂ and MnFe₂O₄@SiO₂@Ag, was also investigated in this study in terms of bactericidal and bacteriostatic effects against both Gram-positive and Gram-negative bacteria determined by the same agar well diffusion method and the zone of inhibition evaluated as presented in table 2. Amoxicillin and streptomycin are the broad-spectrum antibiotics that are used as positive control. MnFe₂O₄ nanoparticle alone had no effect on tested bacterial strains and the coated MnFe₂O₄@PANI and MnFe₂O₄@SiO₂ nanocomposites also showed no antibacterial effects. However when Ag nanoparticles were added to these composites, antibacterial activity was observed positively and the zone of inhibition around the wells on agar was obtained (figure 6).

As observed, MnFe₂O₄@SiO₂@Ag nanocomposite had no bactericidal effects on Gram-positive bacteria whereas it possessed bacteriostatic effects as effective as half of that of the antibiotics. There have been four different concentrations of MnFe₂O₄@SiO₂@Ag used: 5, 10, 20 and 40 mg ml⁻¹. Generally the highest antibacterial effects were obtained at a concentration of 20 mg ml⁻¹ on Gram-positive strains. On the other hand, MnFe₂O₄@SiO₂@Ag showed bactericidal effects only on *P. aeruginosa* as Gram-negative bacterium. This bacterium is resistant to amoxicillin but sensitive to streptomycin. The fact remains that MnFe₂O₄@SiO₂@Ag exhibited the highest zone of inhibition on *P. aeruginosa* of 17 mm, then *B. cepaciae* of 15 mm and *E. coli* of 12 mm at a concentration of 20 mg ml⁻¹. We could say that this superparamagnetic nanocomposite was not sensitive against *S. aureus*, *B. subtilis*, *E. faecalis*, *E. coli* and *B. cepaciae* but showed intermediate susceptibility.

For MnFe₂O₄@PANI@Ag, nearly the same antibacterial effects on both Gram-positive and Gram-negative bacteria are demonstrated. For this nanocomposite, it was observed

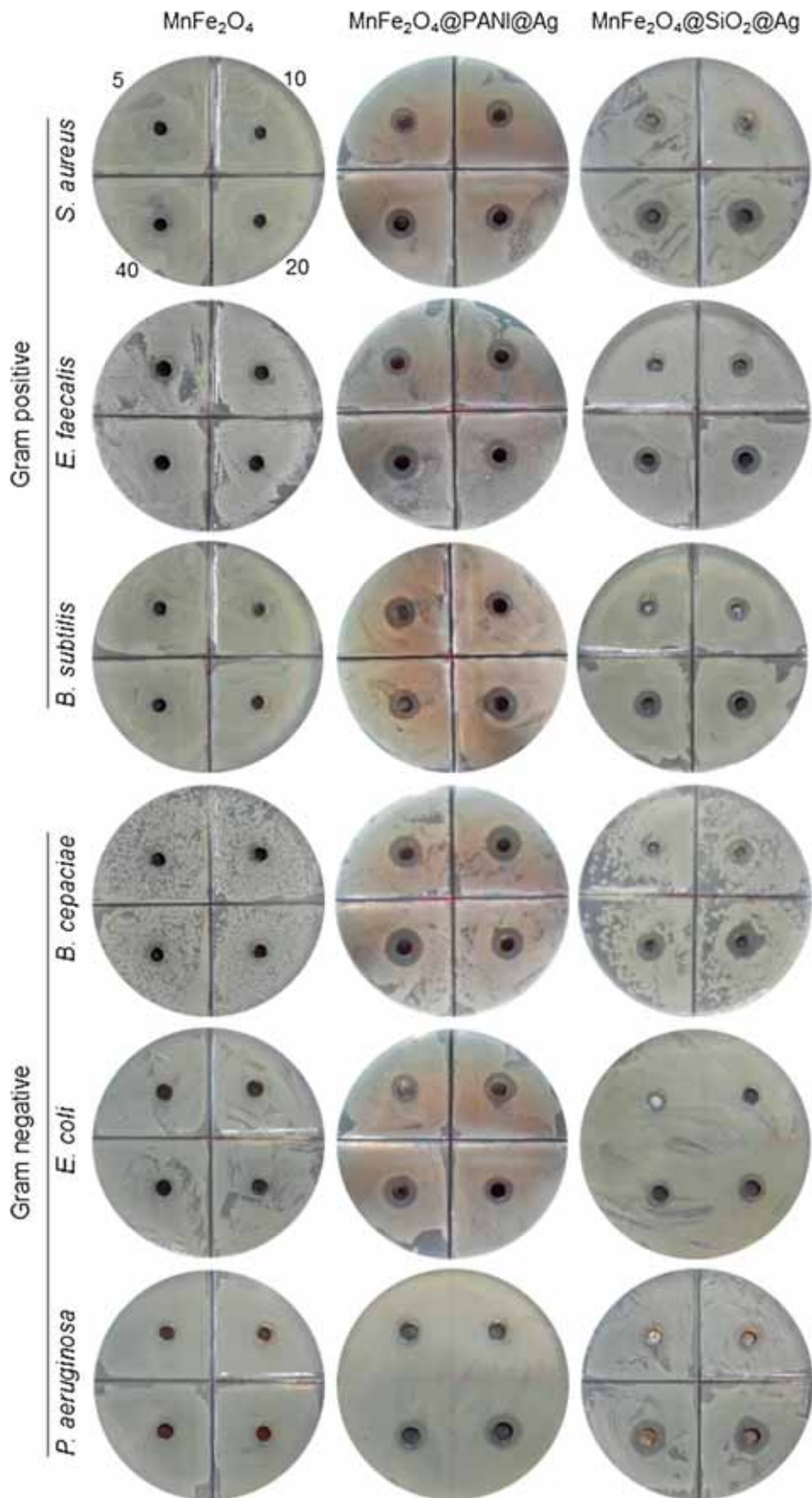


Figure 6. Antimicrobial effects of $\text{MnFe}_2\text{O}_4@\text{PANI}@Ag$ and $\text{MnFe}_2\text{O}_4@\text{SiO}_2@Ag$ against MnFe_2O_4 with different concentrations (5, 10, 20 and 40 mg ml^{-1}) on both Gram-positive and -negative bacteria ($\text{MnFe}_2\text{O}_4@\text{PANI}$ and $\text{MnFe}_2\text{O}_4@\text{SiO}_2$ are not shown here).

that when the concentration was increased, the antibacterial activity was also increased. For *P. aeruginosa*, the same situation was notable. Although MnFe₂O₄@PANI@Ag had a bacteriostatic effects on both Gram-positive and Gram-negative bacteria, it showed bactericidal effect only on *P. aeruginosa*.

According to the antimicrobial properties of the composites with obtained inhibition zones, the results demonstrated that the magnetic MnFe₂O₄, MnFe₂O₄@PANI and MnFe₂O₄@SiO₂ were not active, while the composite binding with Ag nanoparticle was active in antimicrobial property with inhibition zone diameters. The bacteriostatic persistence evaluation showed that this kind of silver-loaded magnetic composite particles had high stability and persistent bacteriostatic activities.

4. Conclusion

Newly synthesized various Ag-containing Fe₃O₄ and MnFe₂O₄ nanocomposites can be considered as antibacterial materials with a broad spectrum antibacterial ability to both Gram-positive and Gram-negative bacteria. The obtained results revealed that these nanocomposites have higher antibacterial performance than that of Fe₃O₄, MnFe₂O₄, Fe₃O₄@HA, Fe₃O₄@Nico, Fe₃O₄@His, MnFe₂O₄@PANI and MnFe₂O₄@SiO₂ nanocomposites and some of the standard antibacterial drugs. Also these magnetic nanocomposites can be reused because of their magnetic property. These results suggest that the unique nanocomposites we have prepared here may serve as an effective, sustainable and recyclable antibacterial agent. This method allows promise for future production technologies and also provides an interesting tool for materials science and a number of novel applications in biological sciences.

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