

Electromagnetic and microwave absorbing properties of hollow carbon nanospheres

TIANCHUN ZOU^{1,*}, HAIPENG LI², NAIQIN ZHAO³ and CHUNSHENG SHI³

¹Airworthiness Certification Technology Research and Management Centre, Civil Aviation University of China, Tianjin 300 300, China

²School of Materials Science and Engineering, Hebei University of Technology, Tianjin 300 130, China

³School of Materials Science and Engineering, Tianjin University, Tianjin 300 072, China

MS received 2 March 2012; revised 30 April 2012

Abstract. A mass of hollow carbon nanospheres (HCNSs) was fabricated by chemical vapour deposition of methane over Ni/Al₂O₃ catalyst at 600 °C. The products were characterized with high-resolution transmission electron microscope images, and the results showed that the external diameter of the HCNSs was 5–90 nm and the thickness of wall was about 10 nm. Microwave absorption of HCNSs/paraffin composites was mainly attributed to dielectric loss. The microwave-absorbing peaks of composites containing HCNSs shifts to low frequencies, and the bandwidth below –10 dB and minimum RL decrease with increasing thickness of HCNSs/paraffin composites.

Keywords. Nanomaterials; nanospheres; CVD; electric; magnetic; microwave absorption properties.

1. Introduction

In recent years, microwave absorbing materials have attracted considerable attention because it is an essential part of a stealthy defense system for all military platforms, either as aircraft, sea or land vehicles. They are also used to eliminate electromagnetic interference (EMI), which is a specific kind of environmental pollution due to the explosive growth in the utilization of electrical and electronic devices in industrial and commercial applications (Bayrakdar 2011; Nam *et al* 2011; Zhang and Sun 2012).

Carbon materials, which are considered as the most promising absorbents since world war II, have been widely employed in microwave-absorbing materials in reducing backscattering from objects or radar targets, EMI suppressors and paints (Ting *et al* 2012; Zhai *et al* 2012). Recently, hollow-carbon nanospheres (HCNSs), which represent a special class of materials, have been attracting intense interest because of their diverse potential applications in drug delivery, heterogeneous catalysis, dye encapsulation, contaminated waste removal and the protection of enzymes and proteins, etc. (Li *et al* 2009, 2010; Zhu *et al* 2012). However, to the best of our knowledge, little has been reported about HCNSs' performance of applications such as microwave absorbent in the open literature. The present work attempts to investigate electromagnetic and microwave absorbing properties of HCNSs synthesized by the catalytic method at low temperature.

2. Experimental

The process for synthesizing HCNSs involved the preparation of catalyst precursor, Ni(OH)₂/Al(OH)₃, and the catalytic decomposition of methane over Ni/Al₂O₃ catalyst. Previously (He *et al* 2006), we fabricated the Ni/Al catalyst using a precipitation route. Here also we used this process but excess NaOH (till the pH value of the solution arrived at 10) to produce binary colloid Ni(OH)₂/Al(OH)₃ (weight ratio, Ni/Al = 2/3). After that the colloid was calcined in air to yield clusters made of Ni and Al oxides and then reduced in a H₂ flow (100 ml/min) at 600 °C for 120 min to form Ni nanoclusters supported on Al₂O₃. To synthesize HCNSs, 50 mg NiO/Al₂O₃ was distributed uniformly in a quartz boat and was placed into a tube furnace. When the furnace reached 600 °C under nitrogen and hydrogen (100 ml/min, 99.99% purity) was introduced to reduce the catalyst for 2 h. Then the hydrogen flow was stopped, and a mixture of CH₄/N₂ (60/420 ml/min, v/v) was introduced into the tube and maintained for 1 h. After the growth, the furnace was cooled to room temperature under N₂ and black powder was obtained. The black mixture was added in a round-bottomed flask containing distilled water and dispersed sufficiently by ultrasonication for 1 h. The resultant HCNSs were collected by centrifugation, then they were dried at 80 °C in an oven for 2 days. After cooling it to ambient temperature naturally, the final products were obtained.

The morphology of as-grown HCNSs was characterized using transmission electron microscope (TEM) (Philips Tecnai G² F20, 200 kV). Co-axial line method was used to determine the electromagnetic parameters of HCNSs/paraffin composite with a HP8722ES vector network

*Author for correspondence (zoutianchun@yahoo.com.cn)

analyser in the frequency range of 2–18 GHz. 5 wt% HCNSs were mixed with paraffin and prepared as the toroidal shape with an outer diameter of 7.0 mm, an inner diameter of

3.04 mm and a thickness of 2.0 mm (see figure 1). The reflection loss (RL) with different thickness was calculated from equations shown below (Feng *et al* 2007):

$$RL(\text{dB}) = 20 \log \left| \frac{Z_{\text{in}} - 1}{Z_{\text{in}} + 1} \right|, \quad (1)$$

$$Z_{\text{in}} = \left(\frac{\mu_r}{\epsilon_r} \right)^{1/2} \tan h \left[j \left(\frac{2\pi f d}{c} \right) (\mu_r \epsilon_r)^{1/2} \right], \quad (2)$$

where Z_{in} is the normalized input impedance at free space and material interface, $\epsilon_r = \epsilon' - j\epsilon''$ and $\mu_r = \mu' - j\mu''$ are the complex relative permittivity and permeability of the material, d the thickness of the microwave absorbing materials and c and f are the velocity of light and the frequency of microwave in free space, respectively.

In the frequency range 2–18 GHz, the complex relative permittivity and permeability of paraffin are small and almost constant ($\epsilon_r \approx 2.0 - 0j$, $\mu_r \approx 1.0 - 0j$). So the electromagnetic properties of HCNSs/paraffin composites mostly represent the electromagnetic performances of HCNSs.

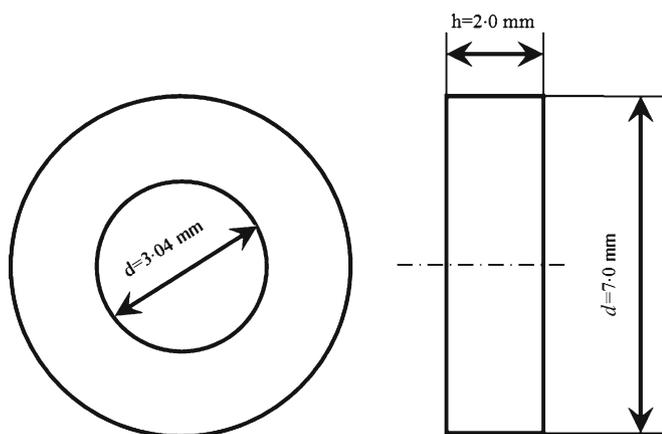


Figure 1. Toroidal shaped HCNSs/paraffin composites for electromagnetic parameter measurements.

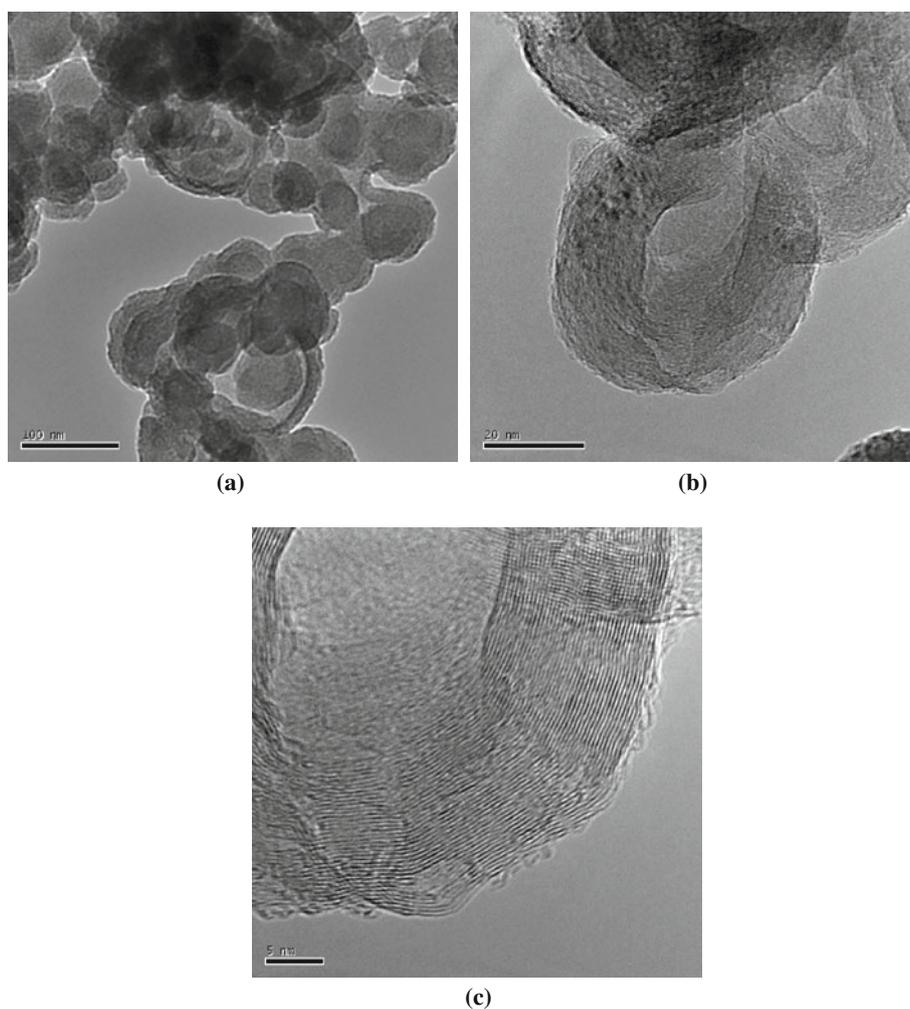


Figure 2. Typical TEM images of as-prepared HCNSs fabricated by catalytic method over Ni/Al₂O₃ catalyst.

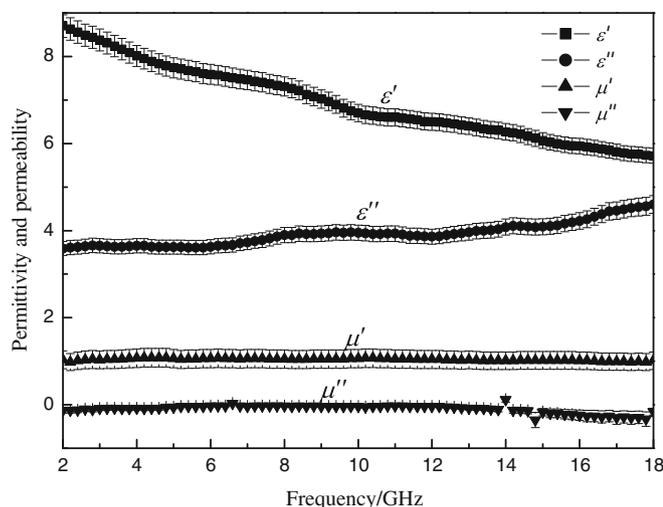


Figure 3. Permittivity and permeability spectra of HCNSs composites.

3. Results and discussion

A representative low-magnification TEM micrograph of the products is shown in figure 2(a). From this we can see that the external diameter of the HCNSs is 5–90 nm and the thickness of the wall ranges from 2 to 50 nm. Typical HRTEM images of the HCNSs are presented in figure 2(b) and (c). Broken sphere and strong contrast between the dark edge and pale centre are revealed in figure 2(b), providing further proof of the hollow nature. Figure 2(c) shows one typical HCNS in high magnification. It indicates that the hollow spheres are composed of carbon sheets and the average wall thickness of HCNSs is about 10 nm. The HCNSs shell consists of about 32 graphitic layers. Figure 2(c) also reveals that the sheets are well-graphitized and the interlayer spacing is about 0.34 nm, which is consistent with the d value of (0 0 2) plane of hexagonal graphite.

Figure 3 shows real and imaginary parts of the complex permittivity, ϵ_r (ϵ' and ϵ'') and complex permeability, μ_r (μ' and μ'') of HCNSs/paraffin composites dependent on the frequency. As shown in figure 3, the values of ϵ' are bigger than those of ϵ'' , and the variation of the ϵ' spectrum is different from the spectrum of ϵ'' . The ϵ' of the HCNSs composite exhibits a fall from 8.70 to 5.72 with an increase in frequency from 2 to 18 GHz, and the ϵ'' of the HCNSs composite gradually rises from 3.59 to 4.6 with an increasing frequency. The μ' and μ'' of the HCNSs composite are almost constant ($\mu' \approx 1.0$, $\mu'' \approx 0$) over 2–18 GHz. This is most likely to be caused by the intrinsically small magnetic loss tangent ($\tan \delta_M$) of HCNSs. So microwave absorption of HCNSs results mainly from dielectric loss rather than magnetic loss.

According to (1) and (2), the RL of microwave-absorbing materials is a function of six characteristic parameters: ϵ' , ϵ'' , μ' , μ'' , f and d . Thus, if the parameters of the materials are known, the absorbing properties of the material for single layer coating can be calculated. RL values calculated by using (1) and (2) for measured values of

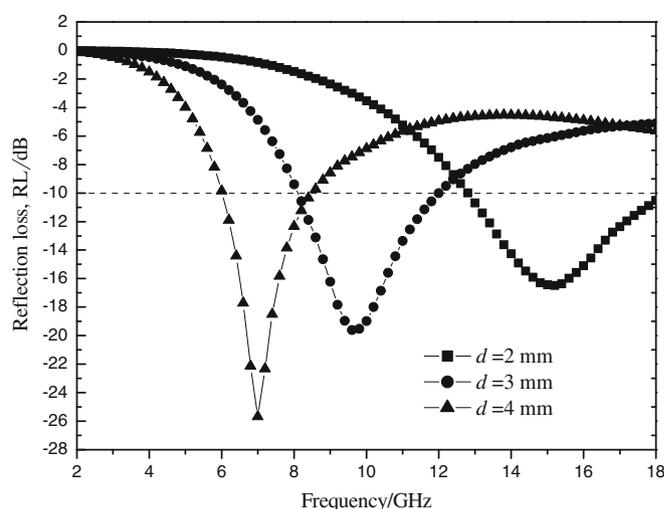


Figure 4. Microwave absorbing characteristics of HCNSs composites.

ϵ' , ϵ'' , μ' and μ'' are shown in figure 4 for HCNSs composites with different matching thickness (d). It can be seen that the values of matching thickness influence the RL of HCNSs composites greatly. Microwave-absorbing peaks move towards lower frequencies with increasing d values, correspondingly, the frequencies corresponding to the minimum RL values of absorbing peaks are 15.2 GHz ($d = 2$ mm), 9.6 GHz ($d = 3$ mm) and 7.0 GHz ($d = 4$ mm). In addition, it can be observed that the bandwidth below -10 dB (90% absorption) becomes narrower and the minimum RL decreases with the increase in the matching thickness, d . When the values of d are 2, 3 and 4 mm, the corresponding bandwidth below -10 dB and minimum RL are 5.2 GHz and -16.5 dB, 4.2 GHz and -19.6 dB, 2.6 GHz and -25.7 dB, respectively.

4. Conclusions

In conclusion, hollow carbon nanospheres (HCNSs) were fabricated by chemical vapour deposition of methane over Ni/Al₂O₃ catalyst. TEM images show that the external diameter of HCNSs was 5–90 nm and the thickness of the wall was about 10 nm. Microwave absorptions of HCNSs/paraffin composites result mainly from dielectric loss rather than magnetic loss. With increasing thickness, microwave-absorbing peaks of composites containing HCNSs move towards low frequencies, and the bandwidth below -10 dB and minimum RL decrease. Our studies indicate that HCNSs may have potential applications in the field of light weight microwave-absorbing materials.

References

- Bayrakdar H 2011 *J. Magn. Magn. Mater.* **323** 1882
- Feng Y B, Qiu T and Shen C Y 2007 *J. Magn. Magn. Mater.* **318** 8

- He C N, Du X W, Ding J, Zhao N Q and Shi C S 2006 *Carbon* **44** 2330
- Li M, Wu Q S, Wen M and Shi J L 2009 *Nano. Res. Lett.* **171** 125
- Li J J, Yuan R, Chai Y Q, Zhang T T and Che X 2010 *Microchim. Acta* **171** 125
- Nam I W, Lee H K and Jang J H 2011 *Compos. Part A: Appl. Sci.* **42** 1110
- Ting T H, Jau Y N and Yu R P 2012 *Appl. Surf. Sci.* **258** 3184
- Zhai Y H, Zhang Y and Ren W 2012 *Mater. Chem. Phys.* **133** 176
- Zhang X Z and Sun W 2012 *Procedia Eng.* **27** 348
- Zhu H L, Bai Y J, Qi Y X, Lun N and Zhu Y 2012 *Carbon* **50** 1871