

## Nanocrystalline spinel ferrites by solid state reaction route

T K KUNDU\* and S MISHRA

Department of Physics, Visva-Bharati, Santiniketan 731 235, India

**Abstract.** Nanostructured  $\text{NiFe}_2\text{O}_4$ ,  $\text{MnFe}_2\text{O}_4$  and  $(\text{NiZn})\text{Fe}_2\text{O}_4$  were synthesized by aliovalent ion doping using conventional solid-state reaction route. With the doping of  $\text{Nb}_2\text{O}_5$ , the size of  $\text{NiFe}_2\text{O}_4$  is reduced down to 33 nm. Similarly, nanostructured manganese ferrites ( $\text{MnFe}_2\text{O}_4$ ) with diameters in the range of 45–30 nm were synthesized by  $\text{Ti}^{4+}$  ion doping. Particle diameters in all the specimens are found to decrease with increasing dopant content. The substitution of  $\text{Nb}^{5+}$  or  $\text{Ti}^{3+}$  ions essentially breaks up the ferrimagnetically active oxygen polyhedra. This created nanoscale regions of ferrites. Saturation magnetization and coercive field show a strong dependence on the size of the ferrite grains. Superparamagnetic behaviour is observed from the Mössbauer spectra of nanostructured  $\text{NiFe}_2\text{O}_4$ , if the particle size is reduced to 30 nm. Zero field cooled and field cooled curves from 30 nm sized  $\text{MnFe}_2\text{O}_4$  particles showed a peak at  $T_B$  (~125 K), typical of superparamagnetic blocking temperature. These results are explained in terms of core/shell structure of the materials. The d.c. resistivity of the doped specimens decreases by at least five orders of magnitude compared to pure sample. This is ascribed to the presence of an interfacial amorphous phase between the sites.

**Keywords.** Nanoparticles; nanoferrites; electrical conductivity.

### 1. Introduction

During the last few years nanoscale spinel ferrites have drawn a major attention because of their technological importance in magnetic recording, magnetic fluids and catalyst. Nanoferrites are usually prepared using various physical and chemical methods like ball milling, microwave plasma, coprecipitation (Ferele and Baberschke 1987; Tang *et al* 1991; Chatterjee *et al* 1993). The synthesis of ferrite nanoparticles is of great interest for studying and tailoring of specific magnetic properties. It was found that the doping of lanthanum impurity ions could suppress the long-range ferroelectric order in the lead zirconium titanate (PZT) system (Li *et al* 1995; Xu *et al* 1995). The nanostructured phase was found to be induced with increasing La impurity content in the ferroelectric ceramic system. A similar approach is made in the present case and a set of nanocrystalline spinel ferrites are prepared using the conventional ceramic route with suitable doping by aliovalent ions (Kundu and Chakravorty 1999). Nanostructured  $\text{NiFe}_2\text{O}_4$ ,  $\text{MnFe}_2\text{O}_4$  and  $(\text{NiZn})\text{Fe}_2\text{O}_4$  were synthesized by using  $\text{Nb}^{5+}$  or  $\text{Ti}^{3+}$  ion doping. The experimental results are reported in this paper.

### 2. Nanostructured $\text{NiFe}_2\text{O}_4$

The samples were prepared with the starting materials,  $\text{Ni}(\text{NO}_3)_2$ ,  $\text{Fe}_2\text{O}_3$  and  $\text{Nb}_2\text{O}_5$ . The composition of the mate-

rials is represented by  $z\text{Nb}_2\text{O}_5$ ,  $(0.5 - z)\text{NiO}$ ,  $0.5\text{Fe}_2\text{O}_3$  with  $z$  having values 0.0, 0.10 and 0.20, respectively. Weighed amounts of starting materials were taken in a mortar and mixed for a duration of 2–3 h under acetone. The powders were calcined in air for 2 h at 800°C. The calcined powders were pressed into pellet and sintered at 1000°C for 2 h.

The crystalline phases precipitated in the sintered specimens were investigated using X-ray diffractograms. The patterns of the samples show that all the expected lines of the  $\text{NiFe}_2\text{O}_4$  spinel structure are present and few lines of  $\text{Nb}_2\text{O}_5$  only appears in specimen 'c' which is doped with  $x = 0.20$ . Thus the present set of spinel ferrites are formed with  $\text{Nb}^{5+}$  getting into the structure.

The grain size of the crystallites in the sintered specimens was estimated from the broadening of the most intense X-ray line for  $\text{NiFe}_2\text{O}_4$  by using Scherrer formula. With the doping of  $\text{Nb}_2\text{O}_5$ , the size of the  $\text{NiFe}_2\text{O}_4$  grain is reduced. The lowest size of  $\text{NiFe}_2\text{O}_4$  grain is 30 nm. Figure 1 is a typical transmission electron micrograph for the specimen *c* of doped  $\text{NiFe}_2\text{O}_4$ . The average values of the particle size are calculated from the micrographs. The results are given in table 1. It is seen that the grain size of the crystallites in undoped  $\text{NiFe}_2\text{O}_4$  specimen is 2  $\mu\text{m}$ . The grain sizes in specimen *b* doped with 10 mole% and specimen *c* doped with 20 mole% are 67 nm and 30 nm, respectively.  $\text{Nb}^{5+}$  ions go into the structure of the doped specimens as it is evident from the broadening of the diffraction lines. Since the ionic radius of  $\text{Nb}^{5+}$  (0.07 nm) is quite close to that of  $\text{Ni}^{2+}$ ,  $\text{Nb}^{5+}$  will substitute the  $\text{Ni}^{2+}$  sites with the creation of vacancy. The presence of  $\text{Nb}^{5+}$  ions and the resultant vacancies can break the coupling of

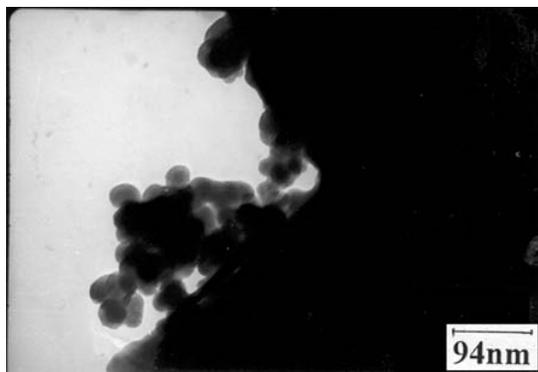
\*Author for correspondence (tkkundu1968@yahoo.com)

ferrimagnetically active oxygen polyhedra and the growth can be inhibited as it occurred in ferroelectric system by suppressing the long range ferroelectric ordering (Li *et al* 1995; Xu *et al* 1995).

Magnetization versus field for the sintered samples was measured using a PAR Vibrating Sample Magnetometer. The values of the room temperature saturation magnetization ( $M_s$ ) and coercive field ( $H_c$ ) as a function of particle diameter are given in table 1.  $M_s$  were measured at a magnetic field of 2000 Oe. It is seen that the substitution of  $Nb^{5+}$  ion decreases the magnetization and increases the coercive field as the particle size of  $NiFe_2O_4$  is reduced.  $M_s$  decreases from 42 emu/g to 15 emu/g as the particle diameter of  $NiFe_2O_4$  is decreased from 1  $\mu m$  to 30 nm (Mishra *et al* 2006). This behaviour can be explained introducing core-shell model of magnetic nanoparticles (Sato *et al* 1987; Kundu and Chakravorty 1999). In the present case, as the  $Ni^{2+}$  ions in  $NiFe_2O_4$  grains are substituted by  $Nb^{5+}$  ions, a layer of  $Nb^{5+}$  doped  $NiFe_2O_4$  phase is generated surrounding the undoped  $NiFe_2O_4$  region. The doped region of particles is defined as shell and the undoped regions as core. The shell has a smaller magnetization compared to the core due to the closed shell structure of  $Nb^{5+}$ . The shell is visualized to be the zone in which  $Nb^{5+}$  ions have created vacancies and the superexchange interaction has been destroyed. Earlier core-shell morphology in nickel ferrite nanoparticles consisting of a ferrimagnetically aligned core spins and a spin-glass like surface layer was introduced to explain the reduction and non-attainment of saturation magnetization (Nathani *et al* 2005). In those cases the spin-glass like surface originated due to broken exchange bonds at the surface of the nanoparticles.  $H_c$  increases from 68 to 111 Oe as the particle diameter is decreased. According to the conventional theory for ferromagnetic particles, the coercivity,  $H_c$ , can be written as (Kundu and Chakravorty 1999)

$$H_c = (2K/M_s)[1 - (T/T_B)^{0.5}], \quad (1)$$

where  $M_s$  is the saturation magnetization,  $K$  the volume magnetocrystalline anisotropy,  $T_B$  the blocking tempera-



**Figure 1.** Transmission electron micrograph of 20 mole%  $Nb_2O_5$  doped  $NiFe_2O_4$ .

ture, which is proportional to the value of the magnetic anisotropy energy  $KV$ , where  $V$  is the volume of the ferromagnetic particle. Previous reports show that  $T_B$  decreases when the particle size is decreased. The decrease in  $T_B$  may cause a reduction of  $H_c$  according to (1). Similarly a decrease in  $M_s$  due to lowering of particle size may also result in an increase in  $H_c$  according to (1). Here, the experimental results show that  $H_c$  for all the specimens increases with the decrease of particle diameter. It means the saturation magnetization is playing a dominant role in determining the variation of coercive field with particle size. A lowering of the value of  $M_s$  with particle size can thus explain the increase of  $H_c$  with a decrease in particle size of  $NiFe_2O_4$ . The Mössbauer spectrum from specimen 'a' consists of six-finger pattern. However, the spectra from the doped specimens contain a doublet in addition to the six-finger pattern (figure 2). The doublet may arise due to the smaller particles in the specimens and is known as superparamagnetic doublet. Larger sized particles are also present in the doped specimens and therefore, the superparamagnetic doublet appears in the background of six-finger pattern.

### 3. Nanostructured $MnFe_2O_4$

The target compositions of the materials are given in table 2. Weighed amounts of mixed precursor are calcined at 800°C for 2 h in air. The powders are taken in a mold for cold compaction at a pressure of 5 bars and finally sintered in air for 2 h at 1100°C.

X-ray diffraction patterns of the sintered samples show that all the expected lines of the  $MnFe_2O_4$  spinel structure are present (Mishra *et al* 2006). Thus the present set of spinel ferrites are formed with  $Ti^{4+}$  getting into the structure. The grain diameters were calculated from X-ray line broadening. The grain size of undoped sample remains in the order of a few micrometer. With the doping of  $TiO_2$ , the size of  $MnFe_2O_4$  grains is gradually reduced down to 30 nm. Figure 3 shows a typical TEM picture for the specimen doped with 20 mole% of dopant. The room temperature saturation magnetization and coercivity for the samples are given in table 2. This magnetic behaviour is quite similar in nature to that obtained for the nanostructured  $NiFe_2O_4$  and thus a similar explanation is also valid for  $MnFe_2O_4$ . In this case some of the  $Mn^{2+}$  ions in  $MnFe_2O_4$  specimens are substituted by  $Ti^{4+}$  ions during processing. Thus the vacancies should be in  $Mn^{2+}$  sites. The resultant vacancies in the presence of  $Ti^{4+}$  ions are believed to break the coupling of ferrimagnetically active oxygen polyhedra.

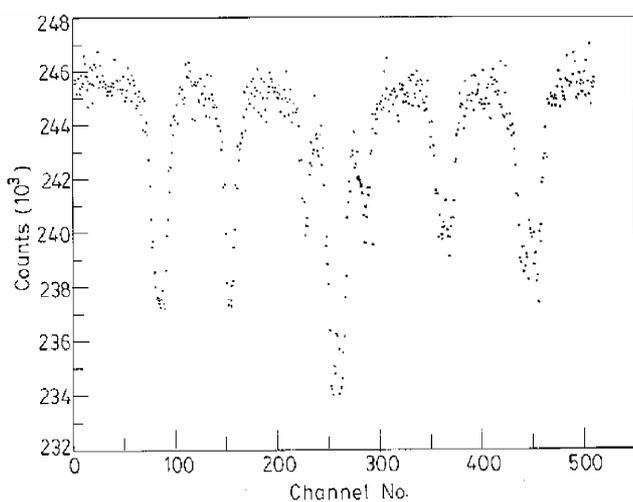
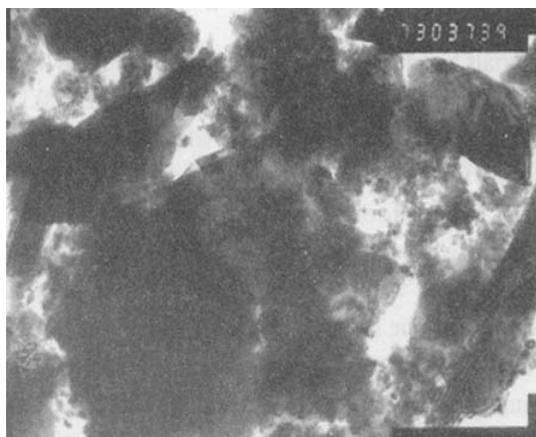
The resistivity decreases significantly as the  $MnFe_2O_4$  samples are doped with  $TiO_2$ . Previous reports show that the electrical resistivity of iron-based samples arises due to a small polaron hopping between  $Fe^{2+}$  and  $Fe^{3+}$  sites (Bramha *et al* 1991). We have fitted the data to Mott's

**Table 1.** Particle diameter, saturation magnetization and coercive field for the NiFe<sub>2</sub>O<sub>4</sub> samples.

Specimen no.	Sample composition (mole%)	Particle diameter from TEM	H <sub>c</sub> (Oe)	M <sub>s</sub> (emu/g)
a	50NiO, 50Fe <sub>2</sub> O <sub>3</sub>	~2 μm	67.98	42.23
b	10Nb <sub>2</sub> O <sub>5</sub> , 40NiO, 50Fe <sub>2</sub> O <sub>3</sub>	67 nm	90.00	29.02
c	20Nb <sub>2</sub> O <sub>5</sub> , 30NiO, 50Fe <sub>2</sub> O <sub>3</sub>	30 nm	111.00	14.96

**Table 2.** Particle diameter, saturation magnetization and coercive field for the MnFe<sub>2</sub>O<sub>4</sub> samples.

Specimen no.	Sample composition (mole%)	Particle diameter from TEM	Coercive field (Oe)	M <sub>s</sub> (emu/g)	ε <sub>p</sub>
1	50MnO, 50Fe <sub>2</sub> O <sub>3</sub>	0.5 μm	44	37.0	1.2
2	10TiO <sub>2</sub> , 40MnO, 50Fe <sub>2</sub> O <sub>3</sub>	45 nm	79	1.6	1.5
3	20TiO <sub>2</sub> , 30MnO, 50Fe <sub>2</sub> O <sub>3</sub>	30 nm	2929	0.45	1.7

**Figure 2.** Mössbauer spectrum of 20 mole% Nb<sub>2</sub>O<sub>5</sub> doped NiFe<sub>2</sub>O<sub>4</sub>.**Figure 3.** Transmission electron micrograph of the 20 mole% TiO<sub>2</sub> doped MnFe<sub>2</sub>O<sub>4</sub>.

equation for small polaron hopping between localized states. The equation describing such a transport is given by (Mott 1969)

$$\rho = [kTR/(\nu_0 e^2 C(1 - C))] \exp(2\alpha R) \exp(W/kT), \quad (2)$$

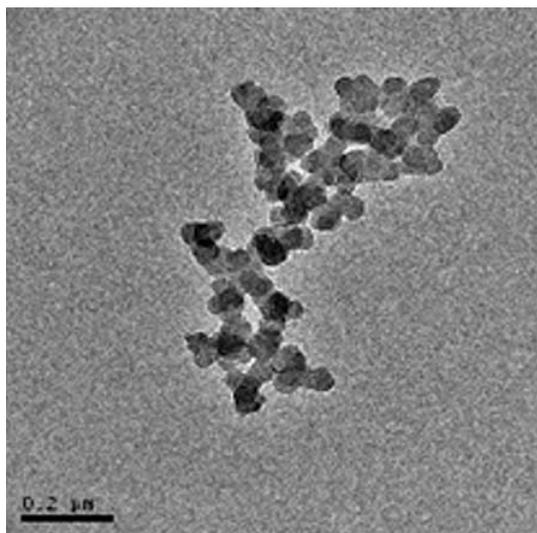
where  $R$  is the average hopping distance,  $\nu_0$  the optical phonon frequency,  $C$  the ratio of Fe<sup>2+</sup> ion concentration to the total Fe ion concentration,  $\alpha^{-1}$  the localization length describing the localized states at each transition metal ion site and  $W$  the activation energy for the hopping conduction. The parameters  $W$ ,  $\alpha$ ,  $R$ ,  $C$  and  $\nu_0$  have reasonable values, and show strong localization. As the value of  $C$  in all the specimens is close to unity, the major fraction of transition metal ions in the specimens remains as Fe<sup>2+</sup>. Effective dielectric constants ( $\epsilon_p$ ) of the interfacial amorphous phase have been estimated from  $W$  and  $R$  using the following relation (Bogomolov *et al* 1968)

$$\epsilon_p = (e^2/2WR)(6/\pi)^{1/3}. \quad (3)$$

The values of effective dielectric constant are given in the last column of table 2. The small values of dielectric constants confirm the presence of an interfacial phase that controls the d.c. electrical properties of the specimens.

#### 4. Nanostructured (NiZn)Fe<sub>2</sub>O<sub>4</sub>

Nanostructured (NiZn)Fe<sub>2</sub>O<sub>4</sub> materials were generated using similar approach as that of NiFe<sub>2</sub>O<sub>4</sub> and MnFe<sub>2</sub>O<sub>4</sub>. The target compositions were  $z$ Nb<sub>2</sub>O<sub>5</sub>,  $(0.5 - z)/2$ MnO,  $(0.5 - z)/2$ ZnO, 0.5Fe<sub>2</sub>O<sub>3</sub> with  $z = 0.0, 0.10, 0.20$  and  $0.25$ , respectively. Final products of nanostructured (NiZn)Fe<sub>2</sub>O<sub>4</sub> magnetic materials in the form of pellet were developed by using the same experimental procedure as discussed in §2. Figure 4 shows a typical TEM picture for the 20 mole% doped (NiZn)Fe<sub>2</sub>O<sub>4</sub>. The lowest diameter of 48 nm is generated with 25 mole% Nb<sub>2</sub>O<sub>5</sub> doping. It is observed that the size of (NiZn)Fe<sub>2</sub>O<sub>4</sub> grain gradually



**Figure 4.** Transmission electron micrograph of the 20 mole% TiO<sub>2</sub> doped NiZnFe<sub>2</sub>O<sub>4</sub>.

decreases from 1 μm to 48 nm with the increase of Nb<sup>5+</sup> content. Magnetic measurement shows that  $M_s$  decreases from 20 emu/g to 4 emu/g and  $H_c$  increases with the decrease of (NiZn)Fe<sub>2</sub>O<sub>4</sub> diameter. The microstructural and magnetic results are very similar to that of nanostructured NiFe<sub>2</sub>O<sub>4</sub> and MnFe<sub>2</sub>O<sub>4</sub> prepared through the same method. Therefore, these experimental results support well the arguments given in §§2 and 3 about the formation of nanostructured materials by aliovalent ion doping.

## 5. Summary

In summary, we have explored the solid-state reaction route for the production of nanocrystalline spinel ferrites. A series of spinel ferrites like NiFe<sub>2</sub>O<sub>4</sub>, MnFe<sub>2</sub>O<sub>4</sub> and (NiZn)Fe<sub>2</sub>O<sub>4</sub> are developed by doping with Nb<sup>5+</sup>/Ti<sup>3+</sup> ions using this conventional ceramic processing route. Nanostructured ferrite particles are found to evolve with increasing concentration of dopant ions. Since the ionic radius of the dopant is quite close to that of Ni<sup>2+</sup> or Mn<sup>2+</sup> or Zn<sup>2+</sup> ions, it substitutes the ions with the creation of vacancy. The presence of dopant ions and the resultant vacancies are believed to break the coupling of ferrimag-

netically active oxygen polyhedra. Saturation magnetization and coercive field show a strong dependence on the size of the ferrite grains. The Mössbauer spectrum of nanostructured NiFe<sub>2</sub>O<sub>4</sub> with 30 nm particle size shows superparamagnetic doublet that confirms the presence of smaller sized particles in the specimens. Zero field cooled and field cooled curves from 30 nm sized MnFe<sub>2</sub>O<sub>4</sub> particles showed a peak at  $T_B$  (~ 125 K), typical of superparamagnetic blocking temperature. These results are explained in terms of core/shell structure of the materials. The d.c. resistivity of the doped specimens decreases by at least five orders of magnitude compared to pure sample. This is ascribed to the presence of an interfacial amorphous phase between the sites.

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