Growth and characterization of new oxide and fluoride crystals for optical applications

K SHIMAMURA*, S L BALDOCHI[†], NA MUJILATU, K NAKANO and T FUKUDA

Institute for Materials Research, Tohoku University, Sendai 980-8577, Japan

[†]Instituto de Pesquisas Energeticas e Nucleares, SP, Brazil

Abstract. High quality oxide and fluoride single crystals for optical applications have been grown by the Czochralski technique. Lattice parameter investigation of grown $\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$ suggested that this crystal will be a superior material as substrate for optical isolators with large Faraday effect. Growth conditions of $(\text{La,Sr})(\text{Al,Ta})\text{O}_3$ single crystals are discussed. These crystals have excellent lattice matching with GaN, a promising material for optoelectronic devices. Ce-doped fluoride single crystals— LiCaAlF_6 , LiYF_4 and BaLiF_3 —have been grown for solid state UV laser applications. Growth results and optical characterization are discussed.

Keywords. Crystal growth; Czochralski method; oxides; fluorides; UV lasers.

1. Introduction

The growth of single crystals has been developed over the years to meet needs for basic research and applications in several different areas. Oxide crystals are of highest importance for modern optical applications, especially as laser hosts, and as substrate or insulator materials in several devices. Fluoride single crystals, because of their unique properties such as large band gap, also present many advantages as optical materials. The crystal growth research described in this work involved the growth of high quality oxide and fluoride single crystals for optical applications. We have studied the crystal growth and properties of $\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$, $(\text{La,Sr})(\text{Al,Ta})\text{O}_3$, and Ce-doped fluorides such as LiCaAlF_6 , BaLiF_3 , and LiYF_4 single crystals.

2. $\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$ (GYGG), a new substrate for optical isolators with large Faraday effect

Bi-containing iron garnet epitaxial layers grown by the LPE method, $(\text{RE}_{1-x}\text{Bi}_x)(\text{Fe,Ga})_5\text{O}_{12}$ (RE: rare-earth or Y^{3+} ions), are attractive for magneto-optical applications such as optical isolators and recording media (Lataifeh and Lehlooh 1996). The magnitude of the Faraday rotation depends linearly on the Bi^{3+} concentration in the garnet epilayers. As the concentration of Bi^{3+} ions increases, the lattice parameter of the typical iron garnet increases linearly. Garnet single crystal substrates, with lattice parameter $\sim 12.6 \text{ \AA}$, are needed in order to grow iron garnet epitaxial layers with higher Bi^{3+} ion concentrations. Although GGG garnet single crystals are used as sub-

strates, their small lattice parameter ($12.38 \text{ \AA}$) is a limiting factor for increase of the Bi^{3+} ion concentration. We have designed a new candidate, $\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$ (GYGG), by substituting the A-sites of GGG with Yb^{3+} . The variation in lattice parameter in the GYGG Czochralski grown crystals and site occupation was studied.

The garnet structure can be expressed as $\text{C}_3\text{A}_2\text{D}_3\text{O}_{12}$, where C, A and D represent cations. The unit cell of the cubic system garnet structure contains 8 units of $\text{C}_3\text{A}_2\text{D}_3\text{O}_{12}$. The 24 C, 16 A, and 24 D ions lie in dodecahedral (C-sites), octahedral (A-sites), and tetrahedral (D-sites) sites, respectively. In order to enlarge the lattice parameter of the GGG single crystals, we have designed a new substance, $\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$ (GYGG), by substituting the A-sites of GGG with Yb^{3+} .

$\text{Gd}_3\text{Yb}_x\text{Ga}_{5-x}\text{O}_{12}$ crystals were synthesized by solid state reaction, and single crystals were grown by the Czochralski technique, with several x values. Figure 1 shows the variation of the lattice parameter, a_{ss} (solid state reaction) and a_{CZ} (CZ single crystals) as a function of the x value. The lattice parameter a increased with the concentration of Yb^{3+} , corresponding to x , the variation being almost linear up to $x=1$.

Figure 2 shows GYGG ($x=2$) single crystals pulled at the rate of 0.6 mm/h with a rotation rate of 10 rpm . Lattice parameters measured for single crystals plotted in figure 1, ranged from $12.38 \text{ \AA}$ to $12.68 \text{ \AA}$, i.e. in the vicinity of the desired lattice constant of $12.62 \text{ \AA}$. GYGG single crystals had a lattice parameter over $12.6 \text{ \AA}$, when x was larger than 1.

Chemical composition and structural analysis of the GYGG ($x=2$) single crystal showed that the C-site was occupied by Gd^{3+} and Yb^{3+} , the A-site by Yb^{3+} and Ga^{3+} ions, and the D-site only by Ga^{3+} ions. Based on the

*Author for correspondence

result of the structural analysis, the formula of the GYGG garnet composition can be expressed as $\{\text{Gd,Yb}\}_3[\text{Yb,Ga}]_2(\text{Ga})_3\text{O}_{12}$, where curly brackets, square brackets, and parentheses enclose the C-, A-, and D-sites, respectively.

The ionic radius of Yb^{3+} is smaller than that of Gd^{3+} , and larger than that of Ga^{3+} . In GYGG single crystals grown by the CZ method, Yb^{3+} preferentially substitutes Gd^{3+} , which occupies C-sites in the x range from 0 to 0.5. For larger x , Yb^{3+} starts to substitute Ga^{3+} in A-sites selectively, and then substitutes Gd^{3+} in C-sites again when x exceeds ~ 1.7 . That is, each site of the CZ grown GYGG single crystals is occupied by cations in a different manner depending on the x range. The variation of a_{ss} has a similar tendency.

3. $(\text{La,Sr})(\text{Al,Ta})\text{O}_3$ (LSAT), a new substrate for GaN epitaxial growth

Due to its wide direct bandgap and high thermal stability, GaN is a promising material for optoelectronic devices such as diodes and lasers emitting from blue to ultraviolet

wavelength regions, as well as for electronic devices operating at high temperatures. It has been reported that GaN layers could be grown on sapphire substrates by MOCVD, (Morkoc *et al* 1994; Nakamura *et al* 1994, 1995, 1996; Akasaki *et al* 1995). However, dislocation densities in these layers are very large, because of the large lattice mismatch between the epitaxial layers and the substrates (14–16%) (Lester *et al* 1995; Osinski and Barton 1996). These dislocations have recently been shown to strongly affect the lifetime of continuous wave blue laser diodes (Amano *et al* 1986). We have found a new material: $(\text{La,Sr})(\text{Al,Ta})\text{O}_3$ (LSAT) as a substrate for GaN epitaxial growth. LSAT single crystals have been grown by the Czochralski technique, and lattice parameter and crystal quality were investigated.

Figure 3 shows an as-grown LSAT single crystal pulled along the $\langle 111 \rangle$ direction at a pulling rate of 1 mm/h. Figure 4a shows a cross-polarized light image of the wafer cut parallel to the growth direction at the shoulder part of the crystal shown in figure 3. The growth interface was fairly convex towards the melt. The existence of a core in the central region was clearly shown by the bright area in the image. Figure 4b shows a similar image of the wafer prepared from the crystal grown under the same growth conditions, except that the rotation rate was 40 rpm rather than 10 rpm. Although the growth interface at 40 rpm was still convex towards the melt, the core observed in figure 4a disappeared, and the shape of the growth interface became smooth. This suggests that a higher crystal rotation speed is preferable in order to decrease the stress in the grown crystals.

The lattice parameter of the LSAT single crystal was 7.735 Å. The lattice parameter was almost constant along the growth axis. The (111) plane of the LSAT crystals is a close-packed plane of the anion framework, which forms the primitive cubic lattice, and thus the $\langle 111 \rangle$ direction is the body-diagonal direction of this cubic cell. The surface structure, therefore, has hexagonal symmetry. The lattice parameter a_{hex} of the LSAT (111) plane, considered as a 2-dimensional hexagonal crystal, is given by

Figure 1. Dependence of the lattice parameter on the Yb^{3+} concentration, x , in GYGG crystals.

Figure 2. As-grown GYGG single crystals pulled at rates of 1.0 mm/h and 0.6 mm/h.

Figure 3. As-grown LSAT single crystal pulled along the $\langle 111 \rangle$ orientation.

$$a_{\text{hex}} = \sqrt{6}/3 \cdot a_{\text{cub}} \quad (1)$$

where a_{cub} is the conventional lattice parameter of the cubic crystal. Since the measured a_{cub} of LSAT was 7.735 Å, a_{hex} was calculated to be 6.316 Å, which was close to twice the corresponding lattice parameter of GaN (3.160–3.190 Å). The lattice mismatch was calculated to be 0.06–1.00%.

4. Ce-doped fluorides for UV laser applications

Coherent optical sources in the ultraviolet wavelength region are useful for many practical applications such as medicine, semiconductor processing, optical communications and remote sensing (Sarukura *et al* 1995). Different kinds of dye lasers and tunable colour centre lasers have been investigated as UV sources, but their applications are still restricted because they are limited concerning temperature operation and crystal deterioration. Recently, Ce-doped LiCaAlF₆ (Ce : LiCAF) crystals

have been reported as candidates for UV solid state lasers (Dubinskii *et al* 1993; Marshall *et al* 1994). This matrix can be directly pumped by the fourth harmonic of a Nd : YAG laser and is suitable for tunable all solid state lasers in UV wavelength region. Very few UV laser materials appropriate for direct pumping are known. The search for new materials includes studies on BaLiF₃ (BLF) (Dubinski *et al* 1997), a new candidate material, and known laser host as LiYF₃ (YLF). We describe the growth of LiCAF : Ce, YLF : Ce and BLF : Ce single crystals by the Czochralski technique. Laser performance and tunability were demonstrated for LiCAF : Ce.

We have studied the LiCAF : Ce growth, under inert and fluorinating atmospheres, investigated the deviation of composition and Ce³⁺ incorporation. Commercially available AlF₃, CaF₂ and LiF powders, of high purity (> 99.99%), were used as the starting materials, CeF₃ powder (> 99.99%), was used to dope the crystal in the range 1–2 mol%. A glassy carbon crucible was used for CZ growth. All crystals were grown from *a*-axis oriented seed crystals. The pulling rate was 1 mm/h and the rotation rate was 10 rpm. The Ce³⁺ doping level in the grown crystals was measured, and the distribution coefficient of Ce³⁺ (k_{Ce}) in LiCAF was estimated as 0.01.

Figure 5 shows an Ce : LiCAF single crystal growth from a stoichiometric charge under inert atmosphere. The crystal surface is covered by white foreign substances. Figure 6 shows a wafer, which was cut perpendicular to the growth axis and polished. It is interesting to note that the inside part is transparent and without cracks or macroscopic inclusions.

Figure 7a shows a SEM image taken of the foreign substance stuck to the crystal surface. It is formed mainly of volatile fluorides such as LiAlF₄, and small amounts of AlF₃ and LiF. CaF₂ were also detected in these foreign substances. The presence of oxides and oxyfluorides such as Al₂O₃, Al₄LiO₆F and CeOF was also detected. Figure 7b shows a SEM image of the spherical inclusions appeared just close to the surface. These spherical inclusions were

Figure 4. Cross-polarized light image of wafers from the shoulder part of the crystal in figure 3, cut parallel to the growth direction. The crystal rotation rate was 10 rpm (a) and 40 rpm (b).

Figure 5. As-grown LCAF single crystal pulled along the *a*-axis.

composed of the same compounds as the above foreign substance.

We can conclude that the formation of these inclusions resulted from evaporation of the melt components and oxide or oxyfluoride formation in the melt because of the presence of water and oxygen in the system or in the raw materials. AlF_3 and LiF have high vapourization pressures; the evaporation of these components shifts the melt composition in the CaF_2 -rich direction. The excess CaF_2 in the melt aggregates and sticks to the growing crystal surface, and/or reacts with other contaminants present in the melt, such as oxides and oxifluorides produced from water and/or oxygen not completely eliminated from the growth system or from the raw materials.

Samples of the central part of the grown crystals were used in preliminary tests of laser action. A schematic diagram of the $\text{Ce}:\text{LiCAF}$ laser resonator is shown in figure 8. The $\text{Ce}:\text{LiCAF}$ crystal is located between the two cavity mirrors. There is no coating on the parallel end faces of the crystal that are perpendicular to the optical axis of the resonator. The fourth harmonic of a Q-switched $\text{Nd}:\text{YAG}$ laser is used as the pumping source. Figure 9 shows the obtained output energies at 289 nm as a function of the absorbed pump energy at 266 nm. High-energy pulse generation from a $\text{Ce}:\text{LiCAF}$ laser operating at 289 nm wavelength was obtained, at a repetition rate of 10 Hz, with up to 30.5 mJ output energy and slope efficiency 39%. This laser performance is the highest ever reported for $\text{Ce}:\text{LiCAF}$ crystals, to our knowledge.

Figure 10 shows a $\text{LiCAF}:\text{Ce}:\text{Na}$ crystal grown under CF_4 atmosphere from a nonstoichiometric composition. An AlF_3 and LiF -rich composition was used to compensate the evaporation loss. It is well known, from the works of Pastor and Arita (1975), that CF_4 reacts with water

in the environment, resulting in HF and CO_2 , yielding a double effect of purification by elimination of water-related impurities and generating a slightly fluorinated atmosphere which also acts on oxygen-derived impurities. New laser measurements with samples prepared from the crystal shown in figure 10 are now under investigation.

The discovery of the $\text{LiCAF}:\text{Ce}$ properties for direct UV pumping stimulated several research groups to investigate the potential of other fluoride hosts as laser media in the UV region. $\text{BLF}:\text{Ce}$ and $\text{YLF}:\text{Ce}$ crystals are indicated as promising candidates for a tunable UV laser. BLF is an inverse perovskite material with cubic structure (space group $O_h^1 - \text{Pm}3\text{m}$) where the monovalent ion Li^+ is at the centre of a F_6 octahedron and the Ba^{2+} divalent ions are in the 12-fold environment site, resulting in a different field interaction from classic perovskite structure. YLF crystals are well known laser hosts, usually doped with rare earth ions as Nd^{3+} , Er^{3+} , Ho^{3+}

Figure 6. As-grown wafer cut perpendicular to the growth axis.

Figure 7. SEM images of (a) a cross-section of the grown crystal with foreign substances on the crystal surface, and (b) a spherical inclusion observed close to the grown crystal surface.

or Tm^{3+} . It is a scheelite material with tetragonal structure (space group $C_{4h}^6 O I_{1/a}^4$).

Both BLF and YLF melt incongruently, and a single crystal must be grown from a nonstoichiometric melt to avoid precipitation of other phases. We have grown large and optically high-quality crystals of Ce-doped BLF and Ce-doped YLF by Czochralski pulling under CF_4 reactive atmosphere (figure 11). Commercially available YF_3 , BaF_2 and LiF powders, of high purity (>99.99%), were used as the starting material. CeF_3 powder (>99.99%), was used to dope the crystal in the range 1–2 mol%. A platinum crucible was used for CZ growth. The pulling rate was 1 mm/h and the rotation rate was 15 rpm.

Figure 8. Experimental setup of a high-energy Ce:LiCAF laser.

Figure 9. Laser output energy as a function of absorbed pump energy in a Ce:LiCaAlF₆ sample.

Figure 12 shows an IR transmission spectrum of both crystals. In fluoride crystals, free OH^- radicals present absorption bands in the region $4000\text{--}3500\text{ cm}^{-1}$; from the figure we can confirm that for the BLF:Ce crystal the CF_4 atmosphere eliminates all traces of OH^- .

Figure 10. As-grown LiCAF:Ce:Na crystal grown under CF_4 atmosphere.

K

Figure 11. As-grown YLF:Ce crystal grown under CF_4 atmosphere.

Figure 12. IR transmission spectra of 10 mm polished samples of BLF:Ce and YLF:Ce.

Figure 13. As-grown BLF:Ce:Na crystal grown under CF_4 atmosphere.

impurities. We observed only small bands in the range $3000\text{--}2800\text{ cm}^{-1}$ that can be attributed to HCO^- molecule. In the case of YLF:Ce, there is no trace of OH^- substitutional defects, but we detected the presence of complexes in the range $3200\text{--}2800\text{ cm}^{-1}$. These bands were also observed in YLF:Er: OH^- co-doped crystals (Ranieri *et al* 1996). The OH^- molecules dissociate during the growth of the YLF crystal, producing O_mH_n complexes that absorb in a wide spectral range from 4000 to 2400 cm^{-1} . We also noted in this crystal the presence of HCO^- . The presence of this molecule is attributed to carbon contamination from the commercial fluorides, but also can result from CF_4 reaction products. The presence of $\text{Me}(\text{OH})_2$ complexes, characterized by narrow bands in the range $3600\text{--}3550\text{ cm}^{-1}$, and usually detected on commercially RE-doped YLF, was not observed.

Usually, oxygen-related centres result in scattering, but we have not detected scattering in our samples. It is important to note that the IR spectrum of figure 12 was measured in large samples (10 mm) with high resolution. The concentrations of these impurities are very low. The dependence of YLF:Ce and BLF:Ce crystal optical quality on the concentration of these complexes is not yet established, but it is supposed that their presence should not affect UV laser performance.

The measured Ce^{3+} concentration in the BLF:Ce crystal was equal to 0.011 at%. The low incorporation of Ce^{3+} ions in the BaLiF_3 host is probably influenced by the different in valence state between these ions and available substitutional sites in the crystal. Theoretically, cerium ions can be introduced in the crystal in more than one valence state with formation of compensating defects, such as vacancies, in the host lattice. The simplest measure to stabilize the valence state, and increase the Ce^{3+} concentration, is to co-dope the crystal with a second impurity or a charge-compensating additive. Figure 13 shows a $\text{BaLiF}_3\text{:Ce:Na}$ grown under CF_4 atmosphere. The boule is transparent, without inclusions on the surface or inside. The measurement of the Ce^{3+}

concentration in co-doped crystals showed a lower Ce incorporation when compared to the singly doped one. The addition of Na^+ ions presented an inverse result from the expected one: sodium decreased the cerium incorporation in the BLF host.

5. Summary

Various new oxide and fluoride crystals were developed. Detailed study of the Czochralski crystal growth process and characterization of the grown crystals allowed the design of new materials for optical applications, such as GYGG, a good candidate as substrate for the growth of iron garnet epitaxial layers with high Bi^{3+} ion concentrations, and LSAT, having high potential as a substrate for the growth of high-quality GaN epitaxial layers with low dislocation density. Study of purification and crystal growth process resulted in the growth of large crystals of LiCAF:Ce:Na, BLF:Ce and YLF:Ce with good optical quality. High-energy pulse generation (30.5 mJ) was obtained from a Ce:LiCAF laser. This laser performance proves that Ce:LiCAF is a promising material for high-energy UV pulse generation combined with a high-power, Q-switched Nd:YAG laser.

Acknowledgements

The authors would like to express sincere thanks to Dr S Durbin, Institute for Materials Research, Tohoku University, for his critical reading of the manuscript and fruitful discussions. The authors are also indebted to Mr Y Murakami, Institute for Materials Research, Tohoku University, and Dr N Sarukura, Institute of Molecular Science, for their help in carrying out the chemical composition analysis and laser experiments.

References

- Akasaki I, Amano H, Sato S, Sakai H, Tanaka T and Koike M 1995 *Jpn J. Appl. Phys.* **34** L1517
- Amano H, Sawaki N, Akasaki I and Toyoda Y 1986 *Appl. Phys. Lett.* **48** 353
- Dubinskii M A, Semashko V V, Naumov A K, Abdulsabirov R Y and Korableva S L 1993 *Laser Phys.* **3** 216
- Dubinski M A, Schepler K L, Semashko V V, Abdulsabirov R Y, Galjautdinov B M, Korableva S L and Naumov A K 1997 *OSA TOPS, Advanced Solid State Lasers* Vol. 10, p. 30–34
- Lataifeh M S and Lehlooh A F D 1996 *Solid State Commun.* **97** 805
- Lester S D, Ponce F A, Craford M G and Steigerwald D A 1995 *Appl. Phys. Lett.* **66** 1249
- Marshall C D, Payne S A, Speth J A, Krupke W F, Quarles G J, Castillo V and Chai B H T 1994 *J. Opt. Soc. Am.* **B11** 2054
- Morkoc H, Strite S, Gao G B, Lin M E, Sverdlov B and Burns M 1994 *J. Appl. Phys.* **76** 1363

- Nakamura S, Mukai T and Senoh M 1994 *Appl. Phys. Lett.* **64** 1687
- Nakamura S, Senoh M, Iwasa N and Nagahama S 1995 *Jpn J. Appl. Phys.* **34** L797
- Nakamura S, Senoh M, Nagahama S, Iwasa N, Yamada T, Matsushita T, Kiyoku H and Sugimoto Y 1996 *Jpn J. Appl. Phys.* **35** L74
- Osinski M and Barton D L 1996 *Proceedings of the international symposium on blue laser and light emitting diodes* p. 217
- Pastor A C and Arita K 1975 *Mater. Res. Bull.* **10** 493
- Ranieri I M *et al* 1996 *J. Crystal Growth* **166** 423
- Sarukura N *et al* 1995 *IEEE J. Selected Topics in Quantum Electronics* **1** 792