

Phase instabilities during Ti-nitride precipitation in nitrogen implanted Ti–6Al–4V alloy

N RAJAN, T S SAMPATH KUMAR, K G M NAIR* and K KRISHAN†

Department of Nuclear Physics, University of Madras, Guindy Campus, Madras 600 025, India

*Materials Science Division, Indira Gandhi Centre for Atomic Research, Kalpakkam 603 102, India

†National Research and Technology Consortium, Parwanoo 173 220, India

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Abstract. Grazing incidence X-ray diffraction, performed on a 70 keV nitrogen implanted Ti–6Al–4V system, reveals phase instabilities, during the course of nitride formation. With the build up of unbound N atom concentration, for a dose of 1×10^{16} ions/cm², the surface region becomes α -rich, whereas, on precipitation of Ti-nitrides at a high dose of 1×10^{17} ions/cm², the β -Ti phase reappears, at the surface and beyond the implanted zone. The low concentration of V and the strain in the nitrided zone, have led to radiation induced martensitic transformation of the β -Ti phase.

Keywords. Phase instabilities; nitrogen implantation; X-ray diffraction.

1. Introduction

The two-phase Ti–6Al–4V alloy finds wide use in a number of areas. Its use for aviation and space applications are well known. The past two decades have witnessed widespread use of the material for artificial joint prosthesis. An increase in the use of the material has been possible because of its advantages such as high strength to weight ratio, creep rupture properties and excellent biocompatibility (Bennet *et al* 1980; Hubler and Macafferty 1980; Buchanan *et al* 1987; Qiu *et al* 1991; Rieu *et al* 1992). Because of its poor tribological properties, Ti and its alloys are subjected to surface engineering (Qiu *et al* 1991; Rieu *et al* 1992), to improve the service life and performance. Nitrogen implantation in these systems has shown a promising path for the successful increase (Buchanan *et al* 1987) in their service life and some important studies are available on processing details for nitride precipitation (Hohmuth and Rauschenbach 1985; Elder *et al* 1989). A more detailed understanding of the evolution of the surface structure of this two-phase alloy during nitrogen implantation appears desirable. Clearly the surface and sub-surface matrix structure (Rieu *et al* 1992) and composition are expected to dictate the wear and corrosion properties.

Combinations of various implantation parameters, such as beam current, use of elevated temperature and multiple energy implantation, could possibly be controlled to produce a variety of surface properties. It is to be noted

that there is no straightforward recipe for an optimum dose or concentration profile for the best service life and performance. Along with these, considerable difficulties exist in the identification of the nature of the compounds formed and their chemical bonding with the various elements in the alloy. Possible existence of nitrides (Qiu *et al* 1990) as Ti(V)N is reported in the literature. Copious precipitation of β -Ti precipitates during MeV ion irradiation of Ti–6Al–4V are known (Wang *et al* 1982; Plumton *et al* 1987). The β -Ti precipitates, thus formed are believed to owe their origin to radiation induced segregation (RIS) phenomenon. The mechanism of transport of undersized solutes towards sinks (Okamoto and Rehn 1979), operates in the segregation phenomenon. Likewise, complex phase instabilities are expected to arise during low energy room temperature implantation of a reactive species, such as nitrogen, which also stabilizes the α -Ti phase. It is important to understand the evolution of the surface matrix under build up of nitrogen concentration. The present work deals with the preliminary investigation of N⁺ implantation induced phase instabilities in Ti–6Al–4V alloy containing deformed β -Ti, arising out of thermomechanical treatment. The use of grazing incidence X-ray diffraction (GIXD), permits depth resolved structural information (Marra *et al* 1979; Arnaud and Brunel 1988–89), surface and sub-surface phase identification by varying the incident angle. Systematic GIXD studies were performed on samples, irradiated to two different doses of nitrogen, one representing a low peak concentration and the other a sufficiently high concentration leading to precipitation of nitrides.

*Author for correspondence

2. Experimental

Ti-6Al-4V (RT-31, MIDHANI, India; composition given in wt%) sheets of 4 mm thickness, cut into coupons of $1 \times 1 \text{ cm}^2$ were used in the study. Surface preparation involved mechanical polishing using silicon carbide papers of grit 400 through 1000, followed by ultrasonic cleaning and final polishing with diamond paste. An optically flat surface was obtained. The samples were cleaned using distilled water and acetone and were thoroughly degreased using trichloroethylene before implantation.

Nitrogen ion (N^+) implantation was carried out using a low energy accelerator. The energy used was 70 keV and the beam was mass analysed. A low beam current of $1.25 \mu\text{A}$ was maintained to avoid heating of the sample. Implantation was carried out at room temperature under ultra high vacuum conditions. Samples were implanted to two doses, viz. 1×10^{16} , and 1×10^{17} ions/ cm^2 . The projected range and the concentration profile were computed using TRIM (Biersack and Haggmark 1980), Monte Carlo code and is shown in figure 1 for a typical dose of 1×10^{17} ions/ cm^2 .

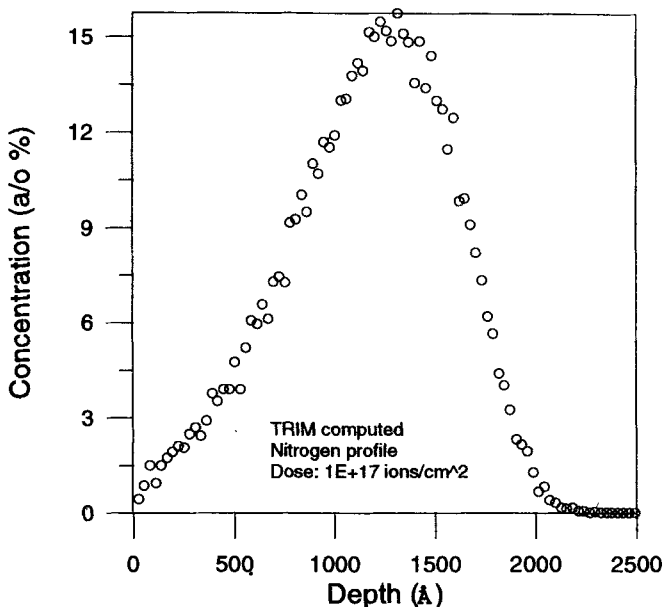


Figure 1. TRIM computed nitrogen concentration profile for 70 keV nitrogen implantation in Ti-6Al-4V alloy.

Table 1. X-ray penetration depth in Ti-6Al-4V alloy.

Incident angle (deg)	Penetration depth (Å)
0.5	796
1.0	1,926
2.0	4,020
3.0	5,686
5.0	8,041

The as-received alloy was characterized by X-ray diffraction (XRD) and optical microscopy. GIXD was carried out using a Siemens D500 X-ray diffractometer with CuK_α radiation. The incident angle θ_i was varied from 0.5° to 10° . The computed penetration depth of X-rays for various θ_i in Ti-6Al-4V alloy is listed in table 1.

3. Results and discussion

Ti-6Al-4V is a two-phase system, comprising hcp α -Ti and the bcc β -Ti phase. The β transus of this system is 995°C . The high temperature β -Ti phase is stabilized to exist at room temperature through alloying with substitutional β stabilizing elements such as V, Mo, Fe, Cr, Mn and Ni.

Figure 2 shows the XRD pattern of the as-received Ti-6Al-4V alloy. The β -Ti(110) line is observed, convoluted with the α -Ti(1011) line. The inset in figure 2 shows the deconvoluted components. The presence of a retained β -Ti phase in a highly deformed and strained state is perceived from the broad hump like β -Ti(110). Unstable β -Ti phase arises out of rapid cooling or quenching from high temperature. This will result in a retained β -Ti with lower V content (Akmoulin *et al* 1994) and remains unstable and susceptible to further transformations. A stable retained β -Ti phase with larger V concentration can be obtained by annealing.

Partial GIXD scans in the range 37° – 43° are shown in figures 3, 4 and 6, for various irradiation doses, as well as for different incident angles θ_i : 0.5° – 10° . The figures represent a series of transformations with dose and depth. Figure 3 represents the partial GIXD scans for the unirradiated sample. The α -Ti(0002) and α -Ti(1011) lines are seen at 38.577° and 40.559° , respectively

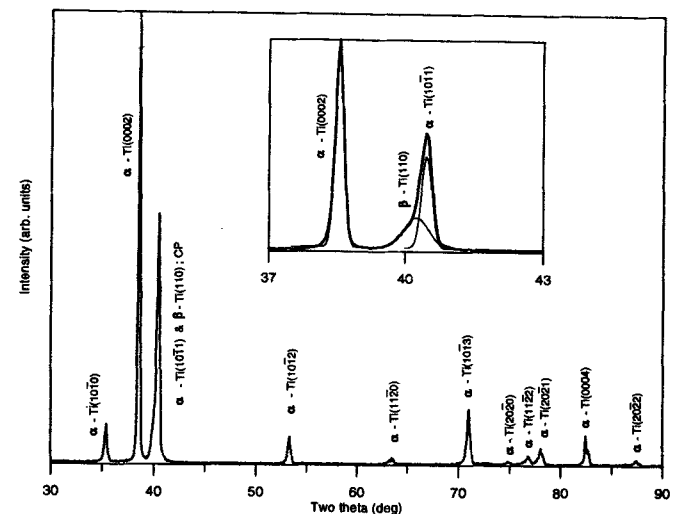


Figure 2. XRD of as-received Ti-6Al-4V alloy. The inset shows the deconvoluted components.

for a θ_i of 10° . The α -Ti(10 $\bar{1}1$) peak is asymmetric owing to the convolution of the β -Ti(110) peak. Deconvolution of this convoluted peak (CP) into gaussians reveals a broad hump for β -Ti(110) with a large full width at half maximum (FWHM) of 0.56. This is essentially an outcome of the microstructure with unstable and non-uniformly strained β -Ti phase. The asymmetry involved in the CP and the ratio (A_1/A_2) of integrated intensities: $I[\alpha\text{-Ti}(10\bar{1}1)]/I[\beta\text{-Ti}(110)]$ shown in figure 7

for samples irradiated to various doses, bring out most of the information regarding the relative abundance of phases and their instabilities under N^+ irradiation. Figure 3 for the unirradiated case shows minor structural variations with depth. The CP shows a greater degree of asymmetry for θ_i : 10° , 5° and 3° . The symmetry, however, is improved for lower θ_i : 0.5° – 2° . Changes in FWHM and intensity of the convoluted components are seen. The intensity of β -Ti(110) is low for higher θ_i . However,

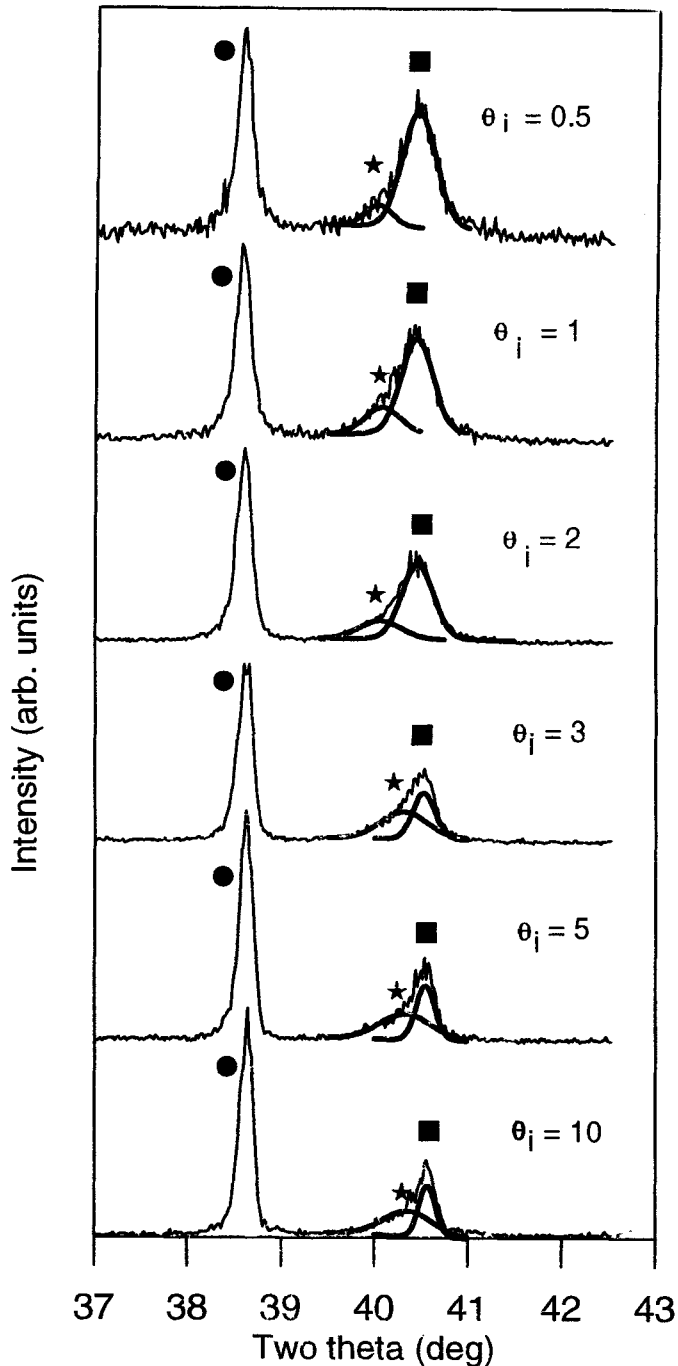


Figure 3. Partial GIXD scans of unirradiated Ti-6Al-4V alloy. (●, α -Ti(0002); ★, β -Ti(110); ■, α -Ti(10 $\bar{1}1$)).

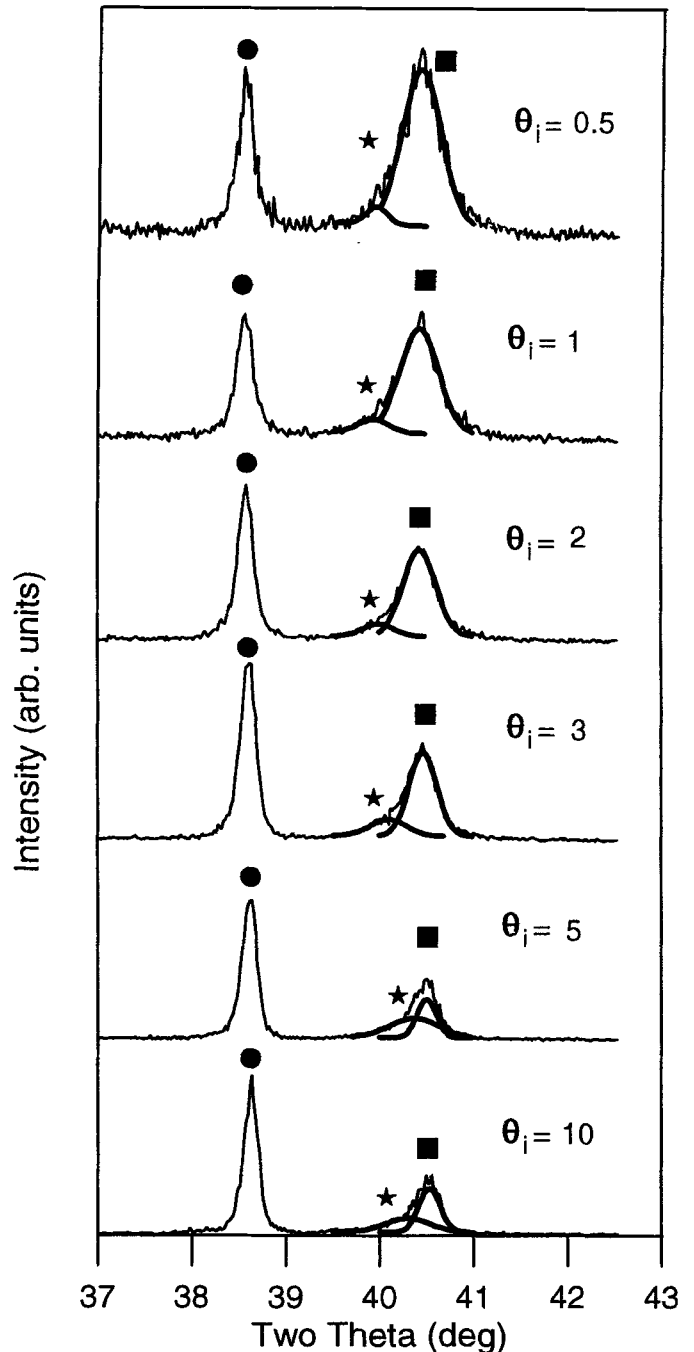


Figure 4. Partial GIXD scans of 1×10^{16} ions/cm 2 N^+ implanted Ti-6Al-4V alloy. (●, α -Ti(0002); ★, β -Ti(110); ■, α -Ti(10 $\bar{1}1$)).

the integrated intensity is much larger in comparison with the near surface cases. The ratio $A1/A2$ for the unirradiated case shown in figure 7, shows an increase in the β -Ti phase volume fraction with depth, which finally stabilizes thereafter to bulk constituents. The surface, thus appears to be α -rich. There is a possibility of carbon and oxygen contamination, which are α stabilizers. These are difficult to be removed completely unless sputter eroded *in situ* before implantation. The results characterize the unirradiated material surface and is useful to compare the changes produced in the surface and sub-surface regions after implantation.

Figure 4 represents the partial GIXD scans for various θ_i of the 1×10^{16} ions/cm² irradiated sample. A peak concentration of 1.76 a/o at a mean depth of 1195 Å (TRIM computed value) is introduced into the system. Best fit and deconvolution of the CP show a rise in the intensity and FWHM of the α -Ti(10 $\bar{1}$ 1) peak and a simultaneous fall in the β -Ti(110) intensity. This is already apparent from the peaks with improved symmetry for θ_i : 0.5°–2°. The ratio $A1/A2$ for the 10^{16} ions/cm² irradiated sample in figure 7 shows a clear and enormous increase, implying an α -phase rich surface zone. There is approximately a linear rapid increase in the α -Ti volume fraction, from bulk to surface. Since N is an α -stabilizing agent, its concentration build up must have stabilized more α volume fraction. The vanadium, stabilizing β -Ti could have possibly undergone a transport away from the nitrated zone, or its concentration is simply not sufficient to stabilize the β -Ti phase, which was previously present. The β -Ti(110) peak is also observed to have shifted towards lower 2θ value, implying

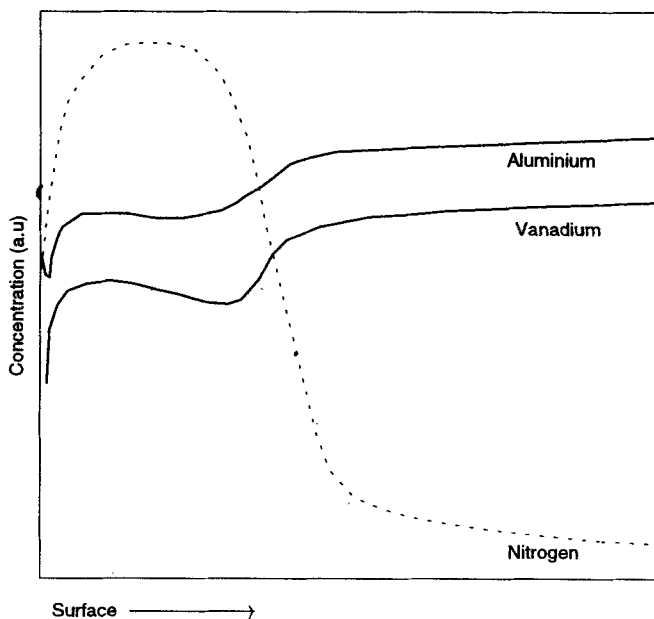


Figure 5. Schematic plot representing Al and V depletion due to high dose nitrogen implantation in Ti-6Al-4V alloy.

a 'de-stabilized' β -Ti (Akmoulin *et al* 1994). In an earlier study by Adami *et al* (1994), depletion of vanadium was seen in N⁺ implanted Ti-6Al-4V alloy for a dose of 5×10^{16} ions/cm² and above. The V concentration at the surface region is therefore, not sufficient to stabilize the previously present β -Ti volume fraction. A schematic depicting vanadium and aluminium depletion due to high

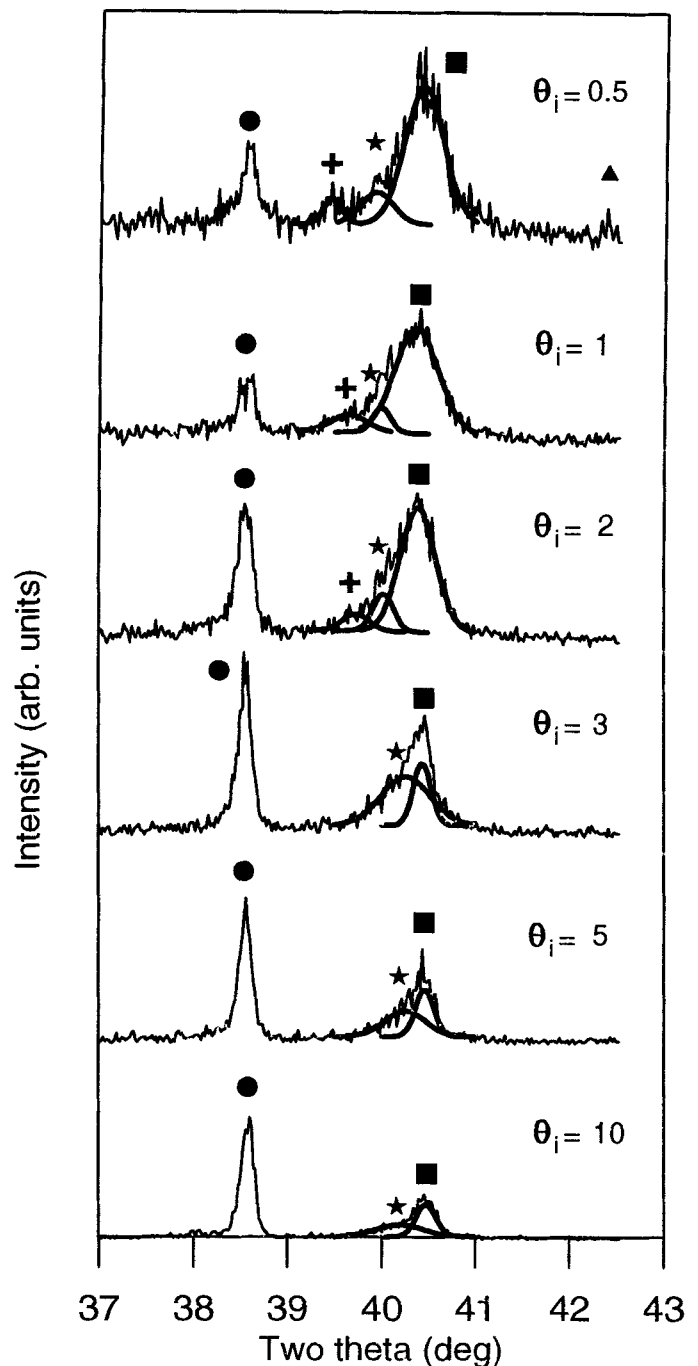


Figure 6. Partial GIXD scans of 1×10^{17} ions/cm² N⁺ implanted Ti-6Al-4V alloy. (•, α -Ti(0002); ★, β -Ti(110); ■, α -Ti(1011); +, ϵ -Ti₂N(111); ▲, δ -TiN(200)).

dose of nitrogen implantation (from literature) is shown in figure 5. The combined effect of the presence of nitrogen atoms and a vanadium depleted zone (Rajan *et al* 1997) is the reason for the observed increase in α -phase.

Figure 6 represents, for various θ_i , the partial GIXD scans of the sample irradiated to a dose of 1×10^{17} ions/cm². The peak nitrogen concentration is estimated to be around 15.22 a/o. The concentration is well above the solubility limit at room temperature, and leads to precipitation. The GIXD pattern at $\theta_i = 0.5^\circ$ shows the presence of the tetragonal ϵ -Ti₂N. The peaks are very broad owing to large amounts of strain and disorder introduced by the high dose N implantation. At this high dose, the intensity of β -Ti(110) line has increased for $\theta_i = 0.5^\circ$. The shift of the β -Ti(110) line towards lower 2θ was intermediate with respect to unirradiated and 1×10^{16} ions/cm² irradiated samples. From figure 7, it can be inferred that there is a clear re-appearance of β -Ti phase at the surface. The ratio A1/A2 falls below the value for the unirradiated case. The increase in the β -Ti volume fraction at surface is expected to have arisen owing to V enrichment implying transport of V towards the surface. Also the presence of β -Ti in the nitrated zone is considerable as observed from figures 4 and 6. This can be attributed to the absence of enough unbound N atoms, which have already participated as nitrides. Likewise, one can observe a highly asymmetric CP for $\theta_i = 3^\circ$, in comparison with the 1×10^{16} ions/cm² and the unirradiated case. The contribution of intensity to the CP is more by β -Ti(110) as can be deduced from

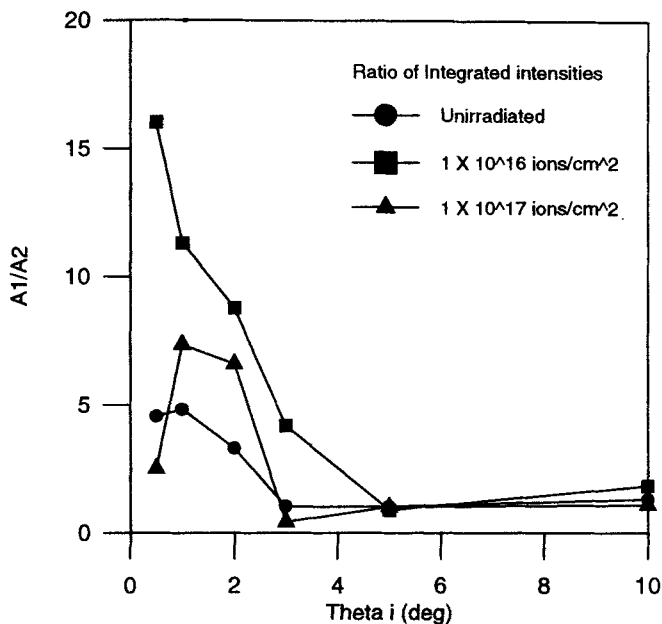


Figure 7. Ratio of integrated intensities A1/A2: $I[\alpha\text{-Ti}(10\bar{1}1)]/I[\beta\text{-Ti}(110)]$.

deconvolution. A rise in the ratio A1/A2 for $\theta_i: 1^\circ$ and 2° and a fall for 0.5° and 3° above and below, respectively, in comparison with the unirradiated case is clearly seen. This suggests depletion of V atoms in the nitrated zone as well as transport of V towards surface and pile-up zone beyond the implanted layer (Rajan *et al* 1997), as it is otherwise not possible to stabilize a larger β -Ti volume fraction which is seen from the relatively intense β -Ti peaks at 0.5° and 3° incident angles. The depletion of V and the other alloying element Al in 30 and 80 keV implantation of N atoms, in the implanted zone has been reported (Adami *et al* 1994) previously. Depletion of V and Al was also observed in an r.f. plasma nitrated (Raveh 1993) Ti-6Al-4V alloy in the nitrated zone. RIS of V atoms towards free surface during MeV irradiation of V (Wang *et al* 1982) or Al (Plumton *et al* 1987) at high temperatures have also been reported in earlier literature. However, RIS has remained difficult to identify in reactive N⁺ implantation in earlier studies (Qiu *et al* 1991), owing to surface and near-surface modification using low energies. The presence of an oxide layer and other contaminations have probably made this observation difficult. In any case, further studies are required in order to understand the phenomenon. The consequence of this result can be important when one considers applications such as in artificial joint prosthesis (Qiu *et al* 1991; Rieu *et al* 1992), as there exists the possible release of toxic V atoms into the body environment (Adami *et al* 1994) with passage of time. The presence of V and Al in the passive oxide layer under corrosion conditions has also been reported. However, the transport of Al or its depletion has not shown any direct manifestation in the X-ray diffraction patterns since Al as well as N are α -phase stabilizers.

In figure 6, a split in the α -Ti(0002) peak is observed for $\theta_i = 1^\circ$. This splitting is possibly due to the formation of α'' martensite. A complete scan over the 2θ range 30° - 90° , with better time step, has revealed splitting of peaks α -Ti(0002), α -Ti(10 $\bar{1}$ 1) and α -Ti(10 $\bar{1}$ 2), characteristic (Akmoulin *et al* 1994) of the α'' martensite. The V depleted region containing unstable β -Ti might easily undergo shearing of the lattice under the existing strain in the nitrated zone, leading to the formation of the martensite. The α'' martensite formation is strongly composition dependent and is known to occur in Ti-Mo as well as in Ti-Al-V systems. The formation of radiation induced martensite in N⁺ implanted zone in Ti-6Al-4V alloy is being reported for the first time, though there are a number of studies on the transformation of β -Ti to martensite α'' phase during thermal treatment and mechanical deformation (Akmoulin *et al* 1994). Radiation induced martensite has also been observed in other systems, such as Fe-10Cr-15Ni (James 1990) as well as in type 304 stainless steel (Chayahara *et al* 1991), previously. Likewise stress or deformation induced

martensites (Niinomi *et al* 1990; Akmoulin *et al* 1994) have been observed in Ti-6Al-4V system as well as in other ($\alpha + \beta$) Ti systems. The low concentration of V in the nitrided zone can be the reason for not sustaining the stabilization of the β -Ti phase under irradiation, which leads to the diffusionless transformation to α'' . The deformed β -Ti in the as-received alloy is expected to favour these transformations. Probably, this might not be the case in a microstructure containing equiaxed α with intergranular β . Further studies are being conducted to understand the complex evolution of the surface and sub-surface phases during N⁺ implantation in Ti-6Al-4V alloy.

4. Conclusion

Ti-6Al-4V alloy is found to experience nitrogen implantation induced phase instabilities. The surface zone turns out to be α -rich at a dose of 1×10^{16} ions/cm². At a high dose of 1×10^{17} ions/cm², the β -Ti phase is found to reappear at the surface and beyond the implanted zone, significantly. Vanadium seems to undergo transport and gets depleted in the implanted layer. The low concentration of vanadium in the strained nitrided zone has led to radiation induced martensitic transformation of β to α'' .

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References

- Adami M *et al* 1994 *X Int. conf. IIT-94, Catania, Italy* (eds) S Coffa, G Ferla, F Priolo and E Rimini, p. 889
- Akmoulin I A, Niinomi M and Kobayashi T 1994 *Met. Metall. Trans.* **A25** 1655
- Arnaud Y and Brunel M 1988-89 *Appl. Surf. Sci.* **35** 345
- Bennet M J, Houlton M R and Dearnaley G 1980 *Corr. Sci.* **20** 69
- Biersack J P and Haggmark L G 1980 *Nucl. Instrum. & Meth.* **174** 257
- Buchanan R A, Rigney E D and Williams J M 1987 *J. Biomed. Mater. Res.* **21** 355
- Chayahara A, Nakashima S, Hashimoto M, Sasaki T, Kurokawa, Kiyama M S and Satou M 1991 *Nucl. Instrum. & Meth.* **B59/60** 893
- Elder J E, Thamburaj R and Patnaik P C 1989 *Surf. Eng.* **5** 55
- Hohmuth K and Rauschenbach B 1985 *Mater. Sci. & Eng.* **69** 489
- Hubler G K and Macafferty E 1980 *Corr. Sci.* **20** 103
- James F 1990 *Met. Metall. Trans.* **A21** 1823
- Marra W C, Eisenberger P and Cho A V 1979 *J. Appl. Phys.* **50** 927
- Niinomi M, Kobayashi T, Inagaki I and Thompson A W 1990 *Met. Metall. Trans.* **A21** 1733
- Okamoto P R and Rehn L E 1979 *J. Nucl. Mater.* **83** 2
- Plumton D L, Kulcinski G L and Dodd R A 1987 *J. Nucl. Mater.* **144** 264
- Qiu X, Conrad J R, Dodd R A and Worzala F J 1990 *Metall. Trans.* **A21** 1663
- Qiu, X, Dodd R A, Conrad J R, Chen A and Worzala F J 1991 *Nucl. Instrum. & Meth.* **B59/60** 951
- Rajan N, Sampath Kumar T S, Tyagi A K, Rajagopalan S, Nair K G M and Krishan K 1997 *Proc. DAE Solid State Phys. Symp. India* **C40** 404
- Raveh A 1993 *Mater. Sci. & Eng.* **A167** 155
- Rieu J, Pichat A, Rabbe L M, Rambert A, Chabrol C and Robelet M 1992 *Mater. Sci. & Tech.* **8** 589
- Wang Z, Ayrault G and Wiedersich H 1982 *J. Nucl. Mater.* **108 & 109** 331