

Proton NMR in CeNiInH_{0.53} down to liquid helium temperature

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MS received 19 December 1997

Abstract. The NMR probe and the matching network has been designed for the ¹H NMR study in CeNiInH_{0.53} down to liquid helium temperature using Bruker MSL 100 spectrometer. NMR line-shape measurement shows the absence of any signature of proton pairing in CeNiInH_{0.53} down to 3.86 K, as it was observed for high hydrogen concentration. The measurement of the spin-lattice relaxation time in the temperature range 300–20 K reveals that the relaxation rate is mainly governed by the Korringa-type relaxation mechanism.

Keywords. NMR; line shape; spin-lattice relaxation.

PACS Nos 76.60; 76.00

1. Introduction

The evidence of an ordered arrangement of proton pairs with H–H separation ~ 1.5 – 1.6 Å, as found in Fe₂P type intermetallic hydrides CeNiInH_{*x*} and PrNiInH_{*x*} ($x \geq 1$) [1, 2] from NMR, has attracted much interest. Such type of pairing of protons with separation 1.9 Å has recently been observed by Kunitomo *et al* [3] in the sodium insertion compound of hydrogen molybdenum bronze. These results violate Switendick's criterion – the nearest neighbour H–H distance in a metal lattice has never been found to be less than 2.1 Å. However, the discovery of the stable bonding neutral dihydrogen molecules to molybdenum and tungsten complexes by Kubas *et al* [4] has led to the speculation regarding the possibility for two hydrogen atoms in bulk metal hydride to favour either a molecular form of pairing or a pairing of the two hydrogen atoms mediated by a metal atom [5].

In this paper, we report on ¹H NMR studies on CeNiInH_{0.53} in the temperature range 3.8–70 K. Our earlier measurements above 77 K show that in CeNiInH_{*x*} with $x < 1.0$, protons do not exist in pairs and they are highly mobile in the interstices. The possibility of pairing of protons as mentioned above even in CeNiInH_{0.53} at low temperatures, wherein hydrogens are supposed to be in rigid condition in comparison with NMR frequency, has led us to do this measurement. Results, however, indicate that there is no such type of pairing of protons in this compound even at liquid helium temperature. Nevertheless, the variation of spin-lattice relaxation time, T_1 with temperature (20–300 K) throw some light on the behaviour of *f* electron of the cerium ion in the compound.